

# ChemCatChem

Supporting Information

## **Selective Anthracene Photooxidation over Titania-supported Single Atom Catalysts**

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Here we show other variations of the formula include multiplying by  $(h\nu)^n$  where  $n$  is the corresponding electronic transition (see Figure S1). The parameter  $n$  corresponds to a direct allowed transition for  $n = \frac{1}{2}$ . For  $n = 2$ , this corresponds to an indirect allowed transition.<sup>1</sup>

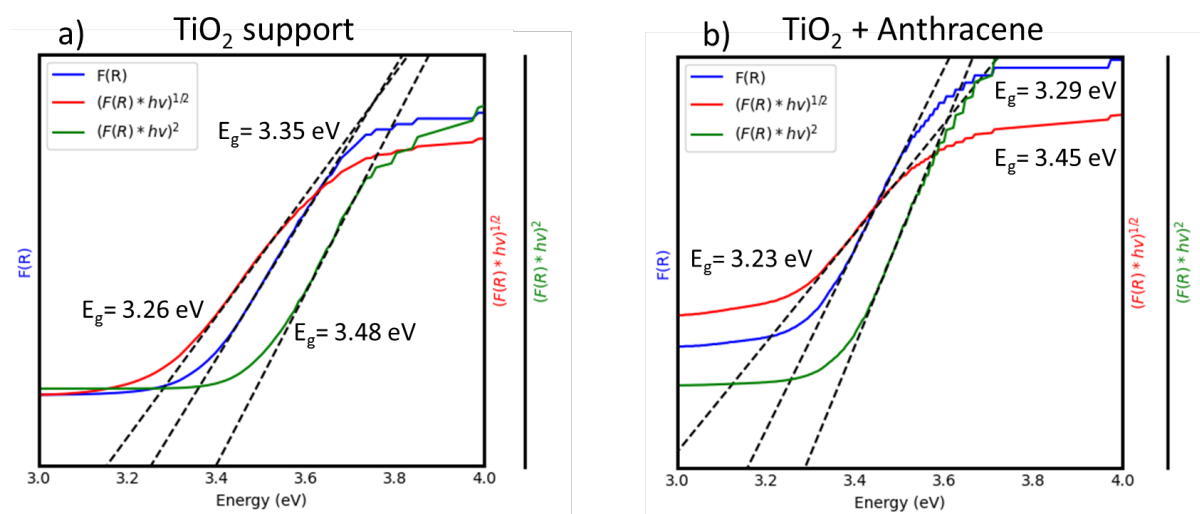


Figure S1 – Other fitting variations of the DRS results. The differences in the bandgap between the  $\text{TiO}_2$  support and anthracene dried over the surface are highlighted.

We also recorded the photochemical conversions for different supports with different bandgaps as reference material. We see a unique behavior of anthracene oxidation possible on  $\text{TiO}_2$ , irrespective of the bandgap.

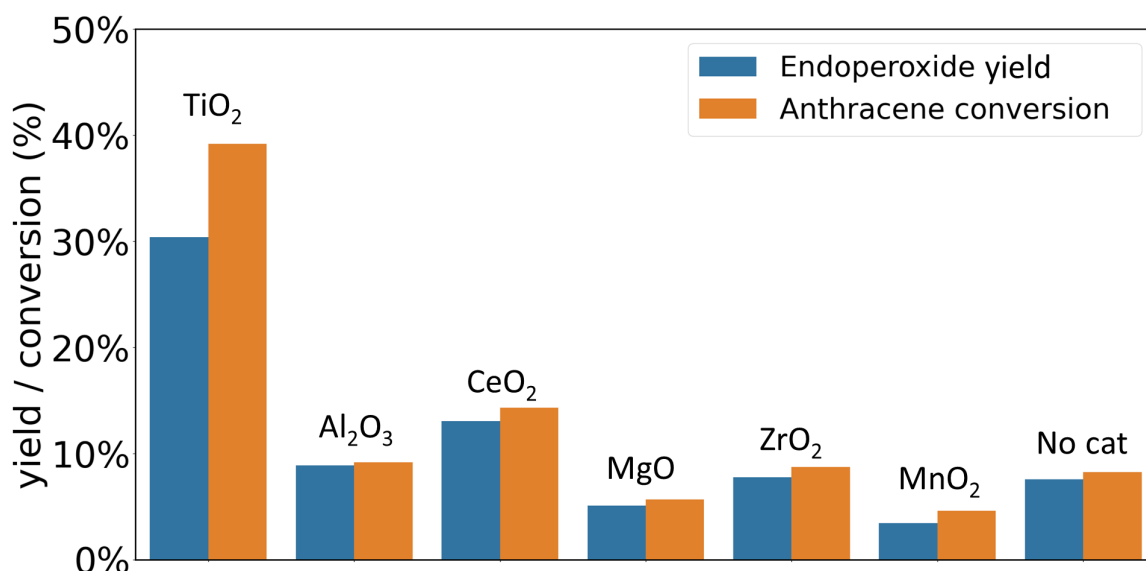


Figure S2 – Endoperoxide selectivity and anthracene conversion for different supports for a 30 min reaction. It shows that anthracene conversion is not limited to the metal oxide. Nevertheless,  $\text{TiO}_2$  showed the highest conversion. Bandgap literature;  $\text{TiO}_2 = 3.0 \text{ eV}^2$ ,  $\text{Al}_2\text{O}_3 = 8.7 \text{ eV}^2$ ,  $\text{CeO}_2 = 3\text{-}4 \text{ eV}^3$ ,  $\text{MgO} =$ ,  $\text{ZrO}_2 = 5.8 \text{ eV}^4$ ,  $\text{MnO} = 2.5^5$ .

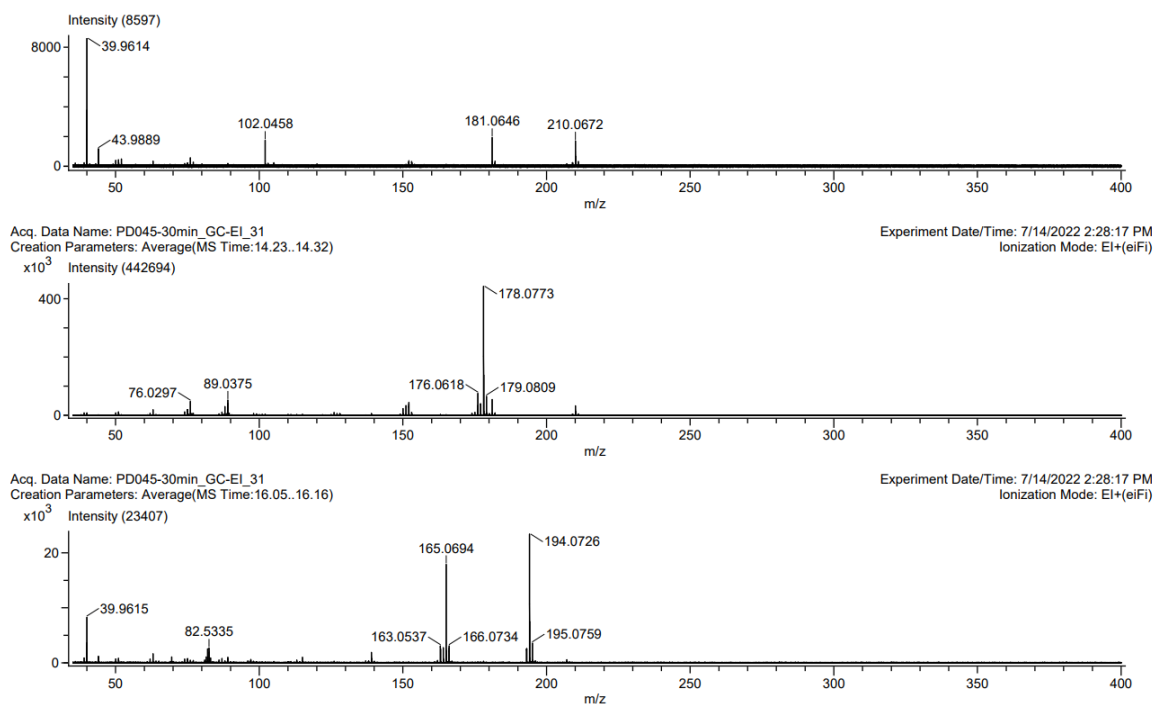


Figure S3 – GC-MS spectrum of the anthracene endoperoxide. 210.0672 matches the exact  $m/z$  ratio for EPO.

We also recorded anthracene conversions to quantify the role of oxygen from the  $\text{TiO}_2$  support. We noticed that there was a small contribution (<10%) of the remaining oxygen from the  $\text{TiO}_2$  support. This could be replenished with gaseous  $\text{O}_2$  and would lead to a Mars van Krevelen type of mechanism for the SAC.

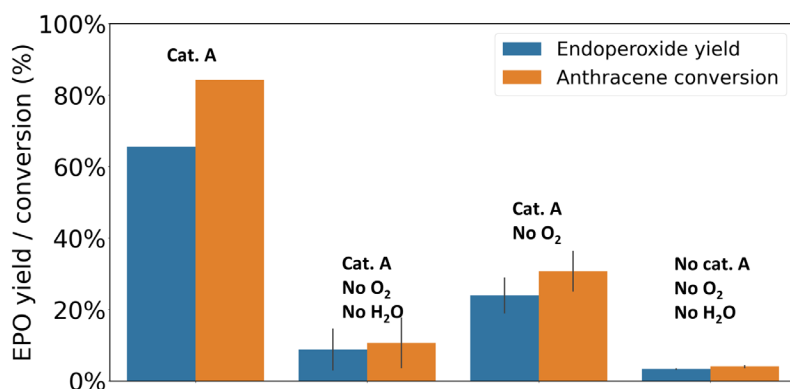


Figure S4 – Conversions when studying the relevance of oxygen. The total conversion decreased completely removing oxygen from the system (duplo measurements). There was a small increase of the conversion when adding a SAC compared to not adding a SAC. These were experiments of 30 minutes.

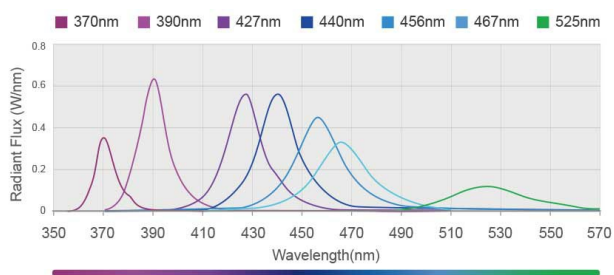


Figure S5 – Emission profile of the lamps available at [https://kessil.com/products/science\\_PRI160L.php](https://kessil.com/products/science_PRI160L.php). Observe the emission profile for the lamp that has a peak maximum of 456 nm.

## References

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