



UvA-DARE (Digital Academic Repository)

Photoreactions of (E) and (Z)-2,5-dimethylhexatriene. Formation of a cyclopropene derivative : a novel reaction in triene photochemistry

Brouwer, A.M.; Jacobs, H.J.C.

Published in:
Tetrahedron Letters

DOI:
[10.1016/S0040-4039\(00\)84268-9](https://doi.org/10.1016/S0040-4039(00)84268-9)

[Link to publication](#)

Citation for published version (APA):
Brouwer, A. M., & Jacobs, H. J. C. (1986). Photoreactions of (E) and (Z)-2,5-dimethylhexatriene. Formation of a cyclopropene derivative : a novel reaction in triene photochemistry. *Tetrahedron Letters*, 27, 1395-1398. DOI: 10.1016/S0040-4039(00)84268-9

General rights

It is not permitted to download or to forward/distribute the text or part of it without the consent of the author(s) and/or copyright holder(s), other than for strictly personal, individual use, unless the work is under an open content license (like Creative Commons).

Disclaimer/Complaints regulations

If you believe that digital publication of certain material infringes any of your rights or (privacy) interests, please let the Library know, stating your reasons. In case of a legitimate complaint, the Library will make the material inaccessible and/or remove it from the website. Please Ask the Library: <http://uba.uva.nl/en/contact>, or a letter to: Library of the University of Amsterdam, Secretariat, Singel 425, 1012 WP Amsterdam, The Netherlands. You will be contacted as soon as possible.

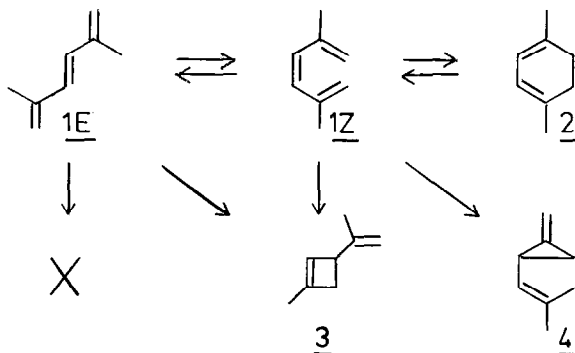
PHOTOREACTIONS OF (*E*)- and (*Z*)-2,5-DIMETHYLHEXATRIENE
 FORMATION OF A CYCLOPROPENE DERIVATIVE: A NOVEL REACTION IN TRIENE PHOTOCHEMISTRY

A.M. Brouwer and H.J.C. Jacobs*
 State University at Leiden, Department of Chemistry
 P.O. Box 9502, NL-2300 RA Leiden, The Netherlands

The products obtained upon direct irradiation of 2,5-dimethylhexatriene at 254 nm have been reinvestigated. The previously observed short-wavelength irradiation product has been identified as 1-methyl-2-(2-methyl-2-propenyl)cyclopropene 6. The proposed intermediacy of a vinylcarbene in the formation of this novel type of product is supported by independent synthesis.

The solution photochemistry of the isomeric 2,5-dimethyl-1,3,5-hexatrienes, 1E and 1Z, was first described by Vroegop *et al.*¹ After prolonged irradiation ($\lambda > 225$ nm) the vinylcyclobutene derivative 3 (major product) and 1-methylene-2-(2-methyl-1-propenyl)cyclopropane 4 were isolated, while evidence was obtained for the presence of a small amount of 1,4-dimethylcyclohexadiene 2 in the irradiation mixture. The triplet-sensitized photochemistry of 1E/1Z was shown¹ to involve only E \rightarrow Z isomerization.

A few years later Gielen *et al.*² demonstrated the strong influence of the irradiation wavelength on the primary photoproduct composition of 1E and 1Z, thus providing strong evidence for the validity of the NEER principle.³ During this study a novel product X was detected by GC analysis. It was produced specifically in short-wavelength irradiation ($\lambda \leq 254$ nm) and appeared to originate from 1E. Scheme 1 summarizes the known photoreactions of the 2,5-dimethylhexatrienes.

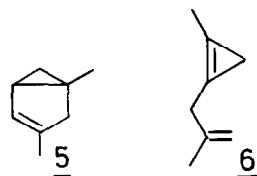


Scheme 1. Photoreactions of (*E*)- and (*Z*)-2,5-dimethyl-1,3,5-hexatriene.^{1,2}

As part of a more general investigation⁴ into the wavelength and conformation effects on the photoreactivity of substituted trienes, we decided to reinvestigate the product mixture resulting from irradiation of 1E in n-pentane at 254 nm, aiming in particular at the isolation and identification of product X. The known products 3 and 4 were isolated and characterized by NMR and IR.⁴ In corroboration of its structure assignment compound 4 was synthesized *via* an independent route and compared to some variously substituted methylene cyclopropanes.⁴ This established the structure of 4 beyond doubt.

Products 3 and 4 account for 75 and 9%, respectively, of the product mixture at 90% conversion of the E-triene.⁵ The proportion of 1Z present in the pseudo-equilibrium 1E \rightleftharpoons 1Z never exceeds 3%; at 90% conversion its amount is negligible. The cyclic isomer 2 is not observed at all under the present conditions of irradiation. According to Gielen *et al.*² its concentration should be less than 0.2% throughout.

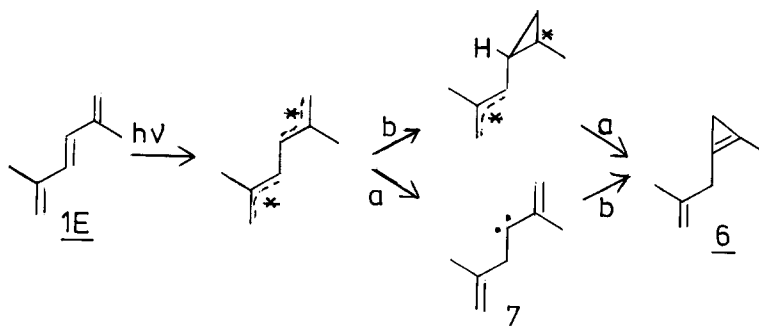
GC shows the presence of two more products, 5 and 6, in the irradiation mixture, accounting for 5 and 12%, respectively. The former compound has escaped attention in earlier studies; it can only be observed separate from 3 by using well-optimized GC conditions.⁶ Samples of 5 obtained by preparative GC invariably contained considerable amounts of 3. In spite of this the signals due to 5 can easily be recognized in the ¹H NMR spectrum of the mixture, and allow the identification of 5 as 1,3-dimethyl(3.1.0)bicyclohex-3-ene.⁷ It must have been formed from 1Z by intramolecular (4 + 2) cycloaddition, a well-known type of reaction in triene photochemistry.



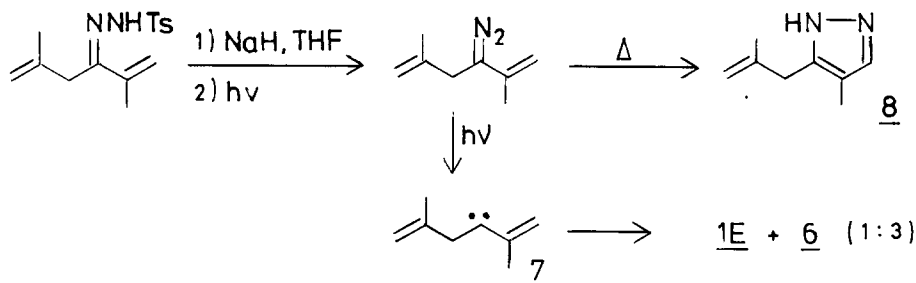
The second novel photoproduct 6 probably is to be identified with product X in the earlier studies.² It was initially characterized by its ¹H NMR spectrum.⁸ A conspicuous feature is the absence of large H-H couplings indicating five separate groups of (nearly) equivalent protons. The chemical shifts and coupling patterns are in favour of the assignment as 1-methyl-2-(2-methyl-2-propenyl)cyclopropene.

The mechanism of the transformation 1E \rightarrow 6 must account for (1.3) bond formation and (1.2) hydrogen shift along the central double bond. Such a rearrangement has not been observed before in triene photochemistry.⁹ The formation of cyclopropenes from butadienes is thought to occur via (1.3) ring closure followed by a (1.3)H shift.¹⁰ That mechanism clearly does not apply to the present rearrangement. Two conceivable routes to the cyclopropene, both involving the twisted bisallyl species but differing in the order of ring closure and hydrogen shift, are depicted in scheme 2.

The sequence a,b implies the intermediacy of carbene species 7. The occurrence of carbenes in the photochemistry of alkenes is well-documented.¹¹ Vinylcarbenes have been proposed as intermediates in allene and cyclopropene photoreactions.¹² To investigate the possible intermediacy of the vinylcarbene 7 it was generated independently by photolysis of the appropriate tosylhydrazone sodium salt¹³ as outlined in scheme 3.



Scheme 2. Possible pathways of the transformation 1E \rightarrow 6
(a: (1.2)H-shift; b: (1.3)bonding).



Scheme 3. Generation and rearrangement of vinylcarbene 7.

The products 1E and 6 obtained proved to be identical (GC, ^1H NMR) to those isolated from the triene photoproduct mixture.

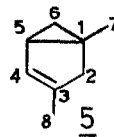
A stepwise mechanism whereby hydrogen migrates along the central double bond followed by cyclization of the vinylcarbene intermediate (scheme 2, sequence a,b) thus adequately accounts for the overall photorearrangement to cyclopropene 6. The feasibility of this route is of interest in view of a recent theoretical treatment¹⁴ of polyene photoisomerization processes, where hydrogen (pseudo)migration has been invoked to explain the fast diabatic transitions between excited and ground states.

The preparation of 6 according to scheme 3 afforded sufficient quantities to record ^{13}C NMR¹⁵ and Raman¹⁶ spectra which gave further support for the structure assignment. Unfortunately we were unable to cool the irradiation mixture sufficiently to prevent the (thermal) ring closure of the diazo intermediate leading to pyrazole 8.¹⁷

At present we are extending these studies to a range of 2,5-dialkylhexatrienes.⁴

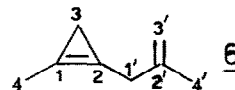
Notes and References

1. P.J. Vroegop, J. Lugtenburg and E. Havinga, *Tetrahedron* **29**, 1393 (1973); P.J. Vroegop, Ph.D. Thesis, Leiden 1972.
2. J.W.J. Gielen, H.J.C. Jacobs and E. Havinga, *Tetrahedron Lett.* 3751 (1976); J.W.J. Gielen, Ph.D. Thesis, Leiden 1981.
3. H.J.C. Jacobs and E. Havinga, *Adv. Photochem.* **11**, 305 (1979).
4. A.M. Brouwer, Ph.D. Thesis, Leiden, to be published.
5. Substance 4 is photochemically labile. When the conversion of 1E exceeds about 90% 4 ($\epsilon_{254} 100$) is no longer protected by the internal filter effect of the strongly absorbing triene and rapidly disappears from the reaction mixture. Its fate is as yet unknown.
6. 30 m OV101 capillary column, 25°C; hydrogen proved to be irreplaceable as a carrier gas.



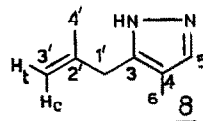
7. ^1H NMR (300 MHz, Bruker WM300, CDCl_3 , chemical shifts in ppm relative to internal TMS): CH_2 -2 2.27 (m) (partly obscured by signals of 3); H_4 5.49 (quin, $J_{4-8} = J_{4-5} = 1.7$ Hz); H_5 1.39 (m, $J_{5-6\text{exo}} = 7$ Hz, $J_{5-6\text{endo}} = 3.4$ Hz); H_6exo 0.66, H_6endo 0.06 ($^2J = 3.4$ Hz); CH_3 -7 1.27 (s); CH_3 -8 1.63 (q).

8. ^1H NMR: CH_2 -3 0.83 (s); CH_3 -4 2.07 (t, $J_{4-1'} = 1.6$ Hz); CH_2 -1' 3.11 (bs); CH_2 -3' 4.77, 4.79 (m); CH_3 -4' 1.78 (t, $^4J = 1.2$ Hz).



9. It now appears to be general in 2,5-dialkylhexatrienes (ref. 4).

10. R. Srinivasan and S. Boué, *Tetrahedron Lett.* 203 (1970); S. Boué and R. Srinivasan, *J. Am. Chem. Soc.* **92**, 3226 (1970).
11. P.J. Kropp, *Org. Photochem.* **4**, 1 (1979); O. Kikuchi, *Bull. Chem. Soc. Jpn.* **57**, 2007 (1984); and references cited therein.
12. M.W. Klett and R.P. Johnson, *Tetrahedron Lett.* **24**, 2523 (1983); M.G. Steinmetz and R.T. Mayes, *J. Am. Chem. Soc.* **107**, 2111 (1985); and references cited therein.
13. H. Dürr, *Chem. Ber.* **103**, 369 (1970).
14. I. Ohmine, *J. Chem. Phys.* **83**, 2348 (1985).
15. 6: ^{13}C NMR (75 MHz, Bruker WM300, solvent CDCl_3 , chemical shifts relative to TMS, multiplicity with respect to $^1\text{J}_{\text{CH}}$): C-1 and C-2 107.9 (s), 106.8 (s); C-3 8.4 (t, $J = 167$ Hz); C-4 22.6 (q, $J = 125$ Hz); C-1' 34.7 (t, $J = 127$ Hz); C-2' 142.6 (s); C-3' 111.2 (t, $J = 154$ Hz); C4' 11.5 (q, $J = 123$ Hz).
16. 1650 cm^{-1} (s): methylene C=C stretch; 1875 cm^{-1} (vs): cyclopropene C=C stretch. We thank Drs. D.J. Stufkens and Th.L. Snoeck (University of Amsterdam) for recording the Raman spectrum.
17. 3-(2-Methyl-2-propenyl)-4-methylpyrazole 8:
 ^1H NMR: H_5 7.32 (bs); CH_3 -6 2.00 (d, $J_{6-5} = 0.6$ Hz); CH_2 -1' 3.32 (bs); $\text{H}_t 3'$ = 4.82 ($J_{t-1'} = 0.8$ Hz, $J_{t-4'} = 1.1$, $J_{c-t} = 2.0$ Hz); $\text{H}_c 3'$ 4.70 ($J_{c-1'} = 1.5$ Hz, $J_{c-4'} = 0.8$); CH_3 -4' 1.71 (dd).
 ^{13}C NMR: C-3 and C-4 113.1 and 116.4 (s); C-5 134.0 (d, $J = 182$ Hz); C-6 8.1 (q, $J = 127$ Hz); C-1' 33.9 (t, $J = 127$ Hz); C-2' 142.6 (s); C-3' 111.7 (t, $J = 156$ Hz); C-4' 22.1 (q, $J = 126$ Hz).



(Received in UK 2 January 1986)