Two-colour picosecond timeresolved \((2 + 1\ C')\) resonance enhanced multiphoton ionization photoelectron spectroscopy on the B \(E_1EC'C'\) and CC' \(E_1AC1E''\) states of ammonia

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Two-Color Picosecond Time-Resolved \((2 + 1')\) Resonance-Enhanced Multiphoton Ionization Photoelectron Spectroscopy on the \(\tilde{B} \, ^1E''\) and \(\tilde{C}' \, ^1A_1'\) States of Ammonia

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The picosecond predissociation dynamics of vibronic levels of the \(\tilde{B}\) and \(\tilde{C}'\) Rydberg states of ammonia have been investigated in real time by \((2 + 1')\) two-color pump–probe ionization in combination with photoelectron spectroscopy. The picosecond real-time results are in reasonable agreement with the results obtained from indirect methods using nanosecond excitation. These indirect methods include investigations of the peak intensities and the natural line widths of the rotational lines in the excitation spectra. The photoelectron spectra obtained for \((2 + 1)\) ionization via the \(\tilde{B}\) state in NH\(_3\) and ND\(_3\) are interpreted and shown to allow for an accurate determination of hitherto unknown vibrational frequencies in the ground state of NH\(_3^+\) (ND\(_3^+\)). For the \(v_1\) symmetric stretch a frequency of \(0.404 \pm 0.007 \text{ eV} (0.304 \pm 0.007 \text{ eV})\) is found, while the frequency of the \(v_4\) asymmetric bend vibration has been established as \(0.197 \pm 0.007 \text{ eV} (0.141 \pm 0.007 \text{ eV})\). The hydrogen atom fragment, which results from the predissociation of the \(\tilde{B}\) and \(\tilde{C}'\) Rydberg states, has been detected in a two-color pump–probe experiment using nanosecond excitation.

I. Introduction

The excited Rydberg states of the ammonia molecule show dynamical decay behavior on a time scale which ranges from femtoseconds to nanoseconds. Numerous studies on the spectroscopic properties as well as on the decay characteristics of the individual rotational levels associated with these excited Rydberg states have made the ammonia molecule a benchmark system for small polyatomic molecules.

Ammonia in its electronic ground state has a \((1e)^4(3a_1^2)\,^1A_1\) configuration in the \(C_3v\) point group and a pyramidal geometry. Due to inversion doubling, the ground state is split into two components separated by 0.793 cm\(^{-1}\) in NH\(_3\) and 0.053 cm\(^{-1}\) in ND\(_3\).\(^1\)\(^2\) These components are usually designated as \(^1A_1'\) and \(^1A_2'\) in the \(D_3h\) symmetry group. The molecule has four vibrational modes: the \(v_1(a_1')\) symmetric stretch, the \(v_2(a_2')\) out-of-plane bend, and two doubly degenerate vibrations, the \(v_3(e')\) asymmetric stretch and the \(v_4(e')\) asymmetric bend. The ground-state configuration in \(D_3h\) is given by \((1e)^4(3a_1^2)\,^1A_1\) (symmetric stretch).

Excitation of an electron from the \(1a_2'^-\) highest occupied lone-pair orbital, centered on the nitrogen atom and responsible for the pyramidal geometry of the ground state, leads to planar geometries for all excited Rydberg states that result from such transitions. The spectroscopic properties of the molecule are consequently in general described in the \(D_3h\) point group. As a result of the large geometry change, the transitions to the Rydberg states show long progressions in the \(v_2'\) out-of-plane bending mode.

The first Rydberg state (\(^1A_1''\)) results from the \(1a_2'' \rightarrow 3sa_1'\) excitation and has been studied extensively. Investigations of the contours and the natural line widths of the rotational lines have shown that the rotational levels associated with the \(v_2'\) vibrational modes are short lived, with typical natural lifetimes of 100 fs in NH\(_3\) and 5 ps in ND\(_3\).\(^3\)\(^4\) These short lifetimes are a consequence of a predissociation process, which yields H and NH\(_2\) fragments. The exact details of this predissociation process in the \(^1A_1''\) state have been studied extensively.\(^5\)\(^6\)\(^7\)\(^17\)\(^24\) Investigations of the spectroscopic properties and the line widths of the individual rotational lines are severely hampered by the lack of rotational resolution for most \(^1A_1''\) vibrational members as a result of the short natural lifetimes. In a resonance Raman study rotational resolution has been obtained, however, and from the line widths the femtosecond lifetimes of individual rotational levels could be established.\(^11\)\(^15\)\(^23\) The \(^1A_1''\) state has attracted interest not only because of its own spectroscopic and dynamic properties but certainly also because of the profound influence it has on the predissociation processes occurring in higher-lying Rydberg states.\(^26\)\(^29\)

It is by now well established that excitation of a \(1a_2'^-\) electron to the \(3p^2\) Rydberg orbital results in the degenerate \(\tilde{B} \, ^1E''\) Rydberg state \((1a_2'^- \rightarrow 3p^2x, y')\) and the \(\tilde{C}' \, ^1A_1'\) Rydberg state \((1a_2'^- \rightarrow 3p^2z')\).\(^4\)\(^28\)\(^30\)\(^33\) Rotationally resolved transitions to the \(\tilde{B}\) \((v_2')\) vibrational members have been examined in vacuum absorption measurements\(^3\)\(^4\)\(^24\) as well as in \((2 + 1)\)\(^29\)\(^35\)\(^40\) and \((3 + 1)\)\(^26\)\(^33\)\(^41\)\(^43\) resonance-enhanced multiphoton ionization (REMPI) studies, whereas the \(\tilde{C}'\) \((v_2')\) levels have, as yet, only been observed by \((2 + 1)\)\(^27\)\(^35\)\(^36\)\(^44\) and \((3 + 1)\)\(^26\)\(^31\)\(^33\)\(^41\)\(^43\) REMPI. These studies have resulted in a vast amount of spectroscopic information on the \(\tilde{B}\) \((v_2')\) and \(\tilde{C}'\) \((v_2')\) Rydberg states. Comparison of the experimentally observed intensities of rotational transitions with theoretically predicted values\(^26\)\(^41\) and accurate sub-Doppler natural line width measurements\(^27\)\(^29\) have enabled an indirect determination of the dynamical predissociation behavior in these Rydberg states.

REMPI in combination with photoelectron spectroscopy (PES) can provide valuable additional information on the ionization process as well as on the spectroscopic and dynamical properties of the excited state under investigation. When REMPI–PES is combined with a picosecond or femtosecond excitation source, it becomes possible to investigate the dynamical processes occurring in these states in real time. In the present study we have used this combination to measure directly the natural lifetimes of rotational levels associated with the \(\tilde{B}\) \((v_2')\) and \(\tilde{C}'\) \((v_2')\) vibrational members of ammonia, which are excited and ionized in a \((2 + 1')\) REMPI process using a picosecond excitation source.

The outline of the present paper is as follows. In section II the experimental details of the picosecond and nanosecond excitation sources will be given. It will be shown that picosecond excitation in combination with photoelectron spectroscopy provides the possibility to perform background-free real-time pump-probe spectroscopy of highly excited molecular states. Subsequently, the $(2+1)$ excitation spectra of the $\tilde{B}$ ($v_2$) and  $\tilde{C}$ ($v_2$) vibrational members in NH$_3$ and ND$_3$, as obtained with nanosecond and picosecond excitation sources, will be discussed (section III.A). The comparison of the intensities of rotational transitions in the experimental excitation spectra with theoretically predicted rotational contour simulations enables an indirect investigation of the rotational-level-dependent predissociation dynamics. The same type of information can be obtained from the individual rotational lines in the excitation spectra (section III.B). The results from these indirect methods will be discussed. In section III.C the photoelectron spectra obtained for $(2+1)$ excitation via the $\tilde{B}$ ($v_2$) and $\tilde{C}$ ($v_2$) vibrational members are described and the $\tilde{B}$ ($v_2$) photoelectron spectra in NH$_3$ and ND$_3$ will be discussed in detail. The latter photoelectron spectra allow for the determination of previously unknown vibrational frequencies in the ground state of the NH$_3$$^+$ and ND$_3$$^+$ ions. In section III.D the results from the real-time lifetime measurements of rotational levels associated with the $\tilde{B}$ ($v_2$) and $\tilde{C}$ ($v_2$) vibrational members in NH$_3$ and ND$_3$ are presented. These results are subsequently compared to the results from the above, indirect, methods in section III.E. Finally, the fragmentation mechanisms of the $\tilde{B}$ ($v_2$) and $\tilde{C}$ ($v_2$) states are discussed in section III.F, and it will be shown that the detection of the hydrogen atom predissociation fragment from the Rydberg states allows for the observation of the predissociation decay channel, competing with ionization channel.

II. Experimental Details

II.A. Experimental Setup. In the present study picosecond and nanosecond laser systems are used in combination with a "magnetic bottle" electron spectrometer. The "magnetic bottle" electron spectrometer setup has been described previously. Briefly, the laser light is focused into the ionization region by means of a plano-convex quartz lens with a focal length of 25 mm. A similar lens is located on the other side of the spectrometer. This lens provides the possibility to focus a second laser beam into the ionization region and thus to perform two-color experiments. A strongly diverging magnetic field parallelizes 50% of the electrons produced in a laser shot, which are subsequently spatially separated according to their kinetic energies in a 50 cm long flight tube. After detection by a pair of microchannel plates the signal is amplified by a home-built preamplifier and stored in a transient digitizer, which is connected to a microcomputer for further analysis. A photoelectron spectrum is built up in the computer by transforming each time only the high-resolution part of the time-of-flight spectrum. By this technique a 15 meV resolution (fwhm) can be obtained at all electron kinetic energies in the present experiments. The spectrometer is calibrated using the lower spin-orbit ionic states of atomic krypton. The laser pulse peak power is kept as low as possible in order to avoid space-charge effects in the focal region.

Apart from kinetic-energy-resolved electron detection the "magnetic bottle" spectrometer also allows for mass-resolved ion detection by application of appropriate voltages to two grids at opposite sides of the ionization region. The detection efficiency is, however, lower than for electron detection.

Figure 1. Schematic diagram of the experimental picosecond setup. The figure contains one dye laser and one dye amplifier, but a total of three dye lasers and two dye amplifiers are available for the experiments: (1) Nd:YLF pump laser; (2) synchronously pumped picosecond or hybridly mode-locked femtosecond dye laser; (3) Nd:YLF regenerative amplifier; (4) 85/15 beamsplitter; (5) dye amplifier; (6) 20/80 beamsplitter; (7) computer-controlled delay line; (8) KDP frequency-doubling crystal; (9) fixed delay line; (10) KDP crystal used for frequency doubling or mixing; (11) "magnetic bottle" electron spectrometer.

NH$_3$ gas (99.96 vol %, Messer Griesheim) and ND$_3$ gas (99.5 atom % D, MSD) were effusively introduced into the spectrometer. Typical pressures were $1 \times 10^{-4}$ mbar in the ionization region and $1 \times 10^{-6}$ mbar in the flight tube with a system background pressure of $2 \times 10^{-7}$ mbar.

Parts of the picosecond laser system have also been described previously. A schematic overview of the picosecond setup used in the present experiments is shown in Figure 1. A CW mode-locked Nd:YLF laser (Coherent Antares 76) produces 18 W average output power at 1053 nm in pulses with a temporal width of 50 ps at a repetition rate of 76 MHz. This beam is frequency doubled in a 12 mm long temperature-tuned LBO crystal. The maximum average output at 526.5 nm is 5 W, but during the experiments the crystal is slightly detuned from the focus to produce 2.5 W output in pulses with a temporal width of 35 ps. The 526.5 nm beam is used to pump synchronously one or two dye lasers (one or two Coherent CR-590 or one Coherent Satori 774). In most of our experiments on ammonia only one dye laser has been used, either the CR-590 picosecond dye laser or the Satori femtosecond dye laser, but in a few cases two CR-590 dye lasers have been employed simultaneously (see section III). The CR-590 dye lasers are operated on rhodamine 6G dissolved in ethylene glycol and have extended cavities, which are matched to the cavity length of the Nd:YLF pump laser. Computer-controlled wavelength selection in the range 560–640 nm on each of these dye lasers is possible by rotating a three-plate birefringent filter using an encoded dc motor. Each CR-590 dye laser is pumped by 1 W of 526.5 nm pump power, producing about 350 mW average output power in near-transform limited pulses with a temporal width of about 3 ps and a spectral width of about 5 cm$^{-1}$. The autocorrelation trace of the CR-590 dye lasers shows a near-Gaussian profile. The Satori femtosecond dye laser is a hybridly mode-locked laser which is operated on rhodamine 6G as the gain dye and DODCI as the saturable absorber. It is pumped by 2 W of 526.5 nm pump power and produces 250 mW average output power in the wavelength range 595–610 nm in near-transform limited pulses with a temporal width of 200 fs and a spectral width of 75 cm$^{-1}$. Wavelength selection is achieved by rotating a one-plate birefringent filter. To maintain the short temporal width of the dye laser pulses over long periods of time, the cavity length is actively stabilized. The femtosecond dye laser autocorrelation trace is also near-Gaussian.

In order to achieve amplification of the dye laser pulses from either the CR-590 or Satori 774 dye laser, 700 mW of the...
remaining infrared beam from the Nd:YLF pump laser is seeded into a Nd:YLF regenerative amplifier (Continuum RGA 47-30) operating at a repetition rate of 30 Hz, which consists of an unstable resonator and a double-pass amplifier. The 1053 nm output with a temporal width of 50 ps is frequency doubled in a 10 mm long angle-tuned KD*P crystal. The output energy of the 526.5 nm pulses (temporal width 35 ps, near-Gaussian autocorrelation trace) can be adjusted by varying the voltage on the flashlamps of the double-pass amplifier stage and is maximally 10 mJ/pulse. The dye laser pulses are amplified using the output of the regenerative amplifier in a three-stage picosecond tunable dye amplifier (Continuum PTA 60). The three stages are all pumped longitudinally for maximum amplification. By splitting the regenerative amplifier output using a 50/50 beamsplitter, the two CR-590 dye lasers can be amplified separately in two different PTAs. In the present study the PTAs are operated on rhodamine 6G, rhodamine 101, or sulforhodamine 640 dissolved in methanol. Due to the solvent, the dye gain curve of the individual dyes is restricted to about 20 nm. In the case of the 3 ps dye laser pulses of the CR-590 dye lasers, the temporal width of the input dye laser pulses is maintained throughout the amplification process, whereas the femtosecond pulses of the Satori 774 dye laser are somewhat broadened to about 250 fs. The maximum output energy per pulse is 0.5 mJ. The short pump pulses from the regenerative amplifier guarantee an amplified spontaneous emission contribution of less than 1%.

For the real-time picosecond experiments described in this study both a pump and a probe pulse are required, preferably of different wavelengths, as will become clear later. The output of one of the PTAs is directed through a retroreflector mounted on a translational stage, which is driven by a computer-controlled stepper motor enabling a maximum delay of 800 ps. This beam is subsequently frequency-doubled in an angle-tuned KDP crystal, which is controlled by a dc motor with encoder. The resulting UV beam is used as the pump beam to achieve excitation. Different types of beams can be used as probe beams. In our experiments on ammonia each possible probe beam enables a different kind of experiment, as will become clear below and in section III. In principle, the same beam might be used to serve as both the pump and the probe beam, but this option does not take full advantage of the possibilities offered by the present setup (vide infra). A second possibility is to frequency-double the output of the second PTA in a second angle-tuned KDP crystal and employ the resulting UV beam as the probe beam. This has been done occasionally in the present study, e.g., to observe the hydrogen atom, which results from predissociation of the B and C Rydberg states of ammonia (see section III). Such a scheme results in a time resolution of about 4 ps. A third possibility is to use only one dye laser and one PTA and to split the output of the regenerative amplifier with a 85/15 beamsplitter. The 85% part is now used to pump the PTA to obtain the pump beam, while the remaining 15% is frequency-doubled in a KDP crystal yielding a 263.5 nm UV pulse which serves as the probe beam. In this case the time resolution is limited by the temporal width of the 263.5 nm probe pulse, i.e., 30 ps. This option has been used to record (2 + 1') picosecond excitation spectra of the B and C Rydberg states of ammonia, for which the probe beam has a fixed time delay with respect to the pump beam (see section III). The last and, in the present experiments, mainly used option is to split in the above scheme, i.e., a 85/15 partitioning of the output of the regenerative amplifier, also the PTA output with a 20/80 beamsplitter. The 20% beam provides the pump beam, whereas the remainder is frequency-mixed in a KDP crystal with the 15% part of the 526.5 nm beam from the regenerative amplifier. The resulting pulse has an energy of \( \hbar \omega_{\text{PTA}} + \hbar \omega_{256.5} \) and a temporal width of about 4 ps. The overall time resolution is found to be about 5 ps. Mixing the 250 fs amplified pulse with the regenerative amplifier output produces pulses with a temporal width of about 2.5 ps, thus showing considerable temporal broadening in the mixing process. The overall time resolution is in this case 2.5 ps. The time delay between the pump and the probe laser pulses can be varied by repositioning the retroreflector in the pump beam path. The timing of both pulses is checked with a fast photodiode. Since this timing is of crucial importance, both the pump and probe beams enter the "magnetic bottle" spectrometer through the same lens on one side of the spectrometer.

The nanosecond laser system has been described previously as well.47-48 A XeCl excimer laser (Lumonics HyperEx-460, 30 Hz repetition rate, temporal pulse width 10 ns, 200 mJ/pulse) pumps a dye laser (Lumonics HyperDye-500), whose output is subsequently frequency-doubled in a unit which is operated with an angle-tuned KD*P crystal (Lumonics HyperTrak-1000). The resulting UV pulses have a spectral width of about 0.15 cm\(^{-1}\). In some experiments in which nanosecond excitation is used two dye lasers have been employed simultaneously. To this purpose, the pump beam is split using a 60/40 beamsplitter. The 40% beam is used to pump the first dye laser as described above. The remaining 60% pumps a second dye laser (Lumonics HyperDye-300) operating on coumarin 480. The dye laser output is frequency-doubled in a second frequency-doubling unit (INRAD II autotracker), which uses an angle-tuned BBO crystal. As the timing between the two UV pulses is not as critical as in the picosecond experiments, the two beams in these experiments enter the spectrometer from opposite sides.

II.B. Real-Time Picosecond Pump–Probe Experiments. In the present study the picosecond laser system is used to perform time-resolved experiments on highly excited molecular Rydberg states of ammonia. In this scheme a pump laser pulse excites the molecule from the ground state to the excited state of interest in a two-photon absorption process. A second laser pulse, the probe pulse, subsequently ionizes the molecule in a one-photon absorption step. By varying the time delay between the pump and probe pulses, it is possible to monitor in real time the excited-state population, which may be depleted by both radiative fluorescence processes and nonradiative processes such as (pre)dissociation. Apart from the ionization by such a (2 + 1') ionization process, a certain number of molecules will also be ionized by only the pump beam, i.e., a (2 + 1) ionization process. The latter process produces a contribution to the signal which is independent of the delay time. The use of the "magnetic bottle" electron spectrometer has the important advantage that these two processes can easily be separated. When ionization occurs by the (2 + 1') process, the photoelectron kinetic energies are given by \( E_{\text{kin}} = 3hv - IE \), while (2 + 1) ionization gives rise to photoelectrons with kinetic energies of \( E_{\text{kin}} = 2hv + \nu h' - IE \), in which \( IE \) is an arbitrary ionization energy. If the difference in the energies of the two colors (\( h\nu - h\nu' \)) is large enough, it is possible to distinguish between the two processes. This is visualized in Figure 2, which shows the electron time-of-flight spectra when only the pump beam is present (Figure 2a) and when both pump and probe beams are present (Figure 2b). The observation that the electrons from both processes arrive at different times at the detector allows us to obtain background-free (2 + 1') wavelength spectra and to study the temporal behavior of the (2 + 1') signal in real time by varying the delay time between pump and probe pulses. Such background-free spectroscopy using the (2 + 1') ionization...
scheme described above would clearly not be possible if (mass-resolved) ion detection would have been used.

A typical decay trace is a convolution of the system response, which is the cross correlation of the UV pump and UV probe laser pulses, with a (multi)exponential decay. The signal increases to a maximum value following the system response and subsequently decreases exponentially to zero. The system response cannot be measured in a simple way and may be expected to vary from day to day, although we have found it to be constant for a time period of several hours. System responses are typically never exactly described by analytical functions. In the analysis of the decay traces in the present study it turned out that excellent fits could be obtained when a Gaussian profile was assumed for the system response. In a complementary experiment to the scheme described above it is also possible to use the probe laser pulse to ionize a (pre)dissociation fragment. The decay trace will then show an increase to a maximum value with comparable temporal behavior as in the scheme described above.

The time resolution in our experiments is limited on the low side by the system response. The approximate widths (FWHM) of system responses for different combinations of pump and probe pulses have been given above. On the high side the time window is limited by the length of the translational stage, which is 12 cm. The temporal window of the present study is thus 2.5–800 ps.

III. Results and Discussion

III.A. Excitation Spectra of Ammonia \( ^3 \)B and \( ^3 \)C States. The \( ^3 \)B and \( ^3 \)C Rydberg states of ammonia, which are investigated in the present study, have been examined previously using one- and two-photon excitation \(^{29,30,33,41–43} \) and with accompanying selection rules \(^{29,30,33,41–43} \) excitation. Due to the geometry change from pyramidal in the ground state to planar in the excited states, the excitation spectra of the \( ^3 \)B and \( ^3 \)C states show long progressions in the \( v_2 \) vibrational out-of-plane bending mode (umbrella motion), peaking in intensity around \( v_2' = 7 \). Although all of these vibronic states have been shown to be affected by predissociation,\(^{3,18,26–29,33,34,41,45,49–51} \) most of them show sufficient rotational structure to allow for detailed investigations of the rotational levels and constants.\(^{4,26–29,32,34–39,41,42,44,45,52} \) The investigations of the rotational structure have resulted in the determination of the excited-state symmetry, as well as in the correct numbering of the vibrational levels of the main progression based on \( v_2' = 2,5,6,9,11,13 \) and 14.\(^{2,5,26,29,31–34,36,39,41,42} \)

Figure 3a shows the \( (2 + 1) \) REMPI excitation spectrum of NH\(_3\) for the \( ^3 \)B and \( ^3 \)C \( v_2' \) vibrational members investigated in the present study using nanosecond excitation in combination with electron detection. The \( v_2' \) numbering of the excited-state vibrational levels indicated at the top of Figure 3 has been adopted from the literature.\(^{5,26,29,31–34,36,39,41,42} \) The excitation spectrum has not been corrected for the dye gain, but the individual connected parts have each been recorded at near-constant laser power. Three sections in Figure 3a have been lifted above the baseline to indicate that they are shown at an enlarged intensity scale (10×). It can be observed that the excitation spectrum is dominated by the \( ^3 \)C\( (v_2' = 1,2) \) states. The \( ^3 \)B\( (v_2' = 6,7) \) states appear with considerably lower intensity, whereas in earlier \( (3 + 1) \) REMPI excitation spectra the \( ^3 \)B and \( ^3 \)C resonances were of comparable intensity.\(^{32,33,43} \) All states except the \( ^3 \)C\( (v_2' = 3) \) state show distinct rotational structure.

Figure 3b shows the equivalent excitation spectrum of NH\(_3\) obtained using picosecond excitation. A similar spectrum obtained with picosecond excitation for the ND\(_3\) states examined in the present study is shown in Figure 4. This figure shows that in ND\(_3\) several \( ^3 \)B and \( ^3 \)C vibronic states overlap. The excitation spectra of Figures 3b and 4 have been recorded with the use of different dyes in the picosecond dye amplifier. Each individual part in these figures represents a different dye, and the picosecond excitation spectra have not been corrected for the dye gain. The rotational structure, which can be observed in the nanosecond excitation spectrum of NH\(_3\) (Figure 3a), is to a large extent lost in the NH\(_3\) picosecond excitation spectrum (Figure 3b) due to the large spectral width of the picosecond UV laser pulses. It is further apparent from Figure 3 that the \( ^3 \)B/\( ^3 \)C intensity ratio is considerably larger with picosecond excitation than with nanosecond excitation, indicating the delicate balance between ionization and decay processes in these states.

A discussion of the excitation spectra of the NH\(_3\) \( ^3 \)C\( (v_2' = 2) \) vibrational band obtained with nanosecond and picosecond excitation sources may serve in many respects as an example for the other states examined in the present study. The out-of-plane bending vibration \( v_2' \) has \( A_2 '' \) symmetry in the \( D_3_h \) point group. It is by now well established that the \( ^3 \)C state has \( A_2 '' \) electronic symmetry in \( D_{3h} \).\(^{31–33,41,42} \) The rovibrionic symmetry of the \( ^3 \)C\( (v_2' = 2) \) vibrational members is therefore \( A_1 ' \) for even \( v_2' \) and \( A_2 '' \) for odd \( v_2' \). Of the two-photon transition tensors only \( T_0 \) and \( T_0 \) with accompanying selection rules \( \Delta J = 0, \Delta K = 0, \Delta l = 1, \pm 1, \pm 2, \Delta K = 0, \) respectively, possess the appropriate symmetry \( (A_1 ^' \) to induce transition intensity in the spectrum. The \( ^3 \)C\( (v_2' = 2) \) excited states with \( A_1 ^' \) \( (A_1 ^{2'} \) rovibrionic symmetry are thus connected with the \( A_1 ' \) \( (A_1 ^{2'} \) levels of the near-degenerate ground-state manifold. The \( T_0 \) tensor dominates the \( ^3 \)C spectra, providing nearly all intensity in the central Q-branch, while the \( T_1 \) tensor contributes only to a small extent, most notably in the wings of each absorption band.

The nuclear spin statistics of the ground state, which result from the fact that the three H atoms \( (I = 1/2) \) or D atoms \( (I = 1) \) in NH\(_3\) or ND\(_3\) are equivalent,\(^{53,54} \) have a profound influence on the rotational contours. For the \( ^3 \)C\( (v_2' = 2) \) band in NH\(_3\) the
nuclear statistics are responsible for a relative enhancement of the transitions originating from rotational levels for which \( K = 3n, n = 0, 1, 2 \ldots \) (ortho levels) compared to transitions beginning in rotational levels with \( K = 3n \pm 1, n = 0, 1, 2 \ldots \) (para levels).

In the present study we have simulated the excitation spectrum of the \( C'(v_2'=2) \) state and the excitation spectra of the other bands with a rotational contour program, which uses the known rotational constants of the ground state\textsuperscript{1,2} and the excited state\textsuperscript{26,29,45} to calculate the energies of the rotational levels associated with these states.\textsuperscript{54} The intensities of the transitions between the rotational levels in the ground state and in the excited state are determined by a line-strength factor,\textsuperscript{55} a Boltzmann factor including the rotational degeneracy as well as the temperature, and the nuclear spin statistics of the ground state. The rotational contour is calculated assuming a certain line shape for the transitions, which is dependent on whether nanosecond or picosecond excitation is employed (vide infra).

Since the simulation using previously determined rotational constants of the \( C'(v_2'=2) \) state in NH\(_3\) exhibited systematic differences with the observed line positions, we have also performed a three-parameter fit on this band in order to obtain values for the \( B' \) and \( C' \) rotational constants as well as the rotationless transition energy \( T_0 \), which could describe the measured excitation spectrum more accurately. A six-parameter fit, in which the effects of the centrifugal parameters \( D_J, D_{JK}, \) and \( D_K \) were included, similar to previous studies on the \( C'(v_2'=0,1) \) yielded unreliable results in the present case. The resulting parameters are given in Table 1, together with previously reported values. Our results compare well with an earlier (2 + 1) study,\textsuperscript{44} but significant differences occur between our (2 + 1) results and the most accurate (3 + 1) study.\textsuperscript{26}

The NH\(_3\) \( C'(v_2'=2) \) excitation spectra obtained with nanosecond and picosecond excitation in combination with electron detection are shown on a larger energy scale in parts b and d of Figure 5, respectively. The resulting simulations using the
rotational levels of the C'(vi=4) vibrational members of ND₃.* In these studies it was shown that transitions to which might compete with the ionization step and depopulate.

For the best nanosecond simulation a Lorentzian line shape with a width of 0.4 cm⁻¹ (fwhm) was used, which indicates that the shape of the rotational lines is determined exclusively by the spectral width of the UV picosecond laser pulses. It can be seen that all rotational resolution is lost in the picosecond excitation spectrum, except for two partly resolved structures corresponding to the largest peaks in the nanosecond spectra. The strong rotational progression designated as (J",K") → (J'=J",K'"=K") where the doubly primed rotational quantum numbers refer to the lower level and singly primed numbers to the upper level, has been marked at the top of Figure 5. This progression is the main contributor to the peaks with the largest intensity in the spectra.

When the heights of the rotational lines in the nanosecond experimental spectrum (Figure 5b) and the simulated spectrum (Figure 5a) are compared, it is clear that considerable differences occur. It is emphasized that these differences do not result from the dye gain or saturation effects. It is well-known that the final step must be wavelength independent, which would be expected in the present case. Secondly, rotational-level-dependent loss mechanisms, which might compete with the ionization step and depopulate the rotational levels associated with the excited state in a J',K'-dependent way, should not be of importance. From previous studies it is known that such loss mechanisms are operative in all Rydberg states of ammonia. Both radiative (fluorescence) and nonradiative (predissociation) processes have been observed. Figure 5a,b shows that the rotational transitions with high values of J'=K'=J"=K" (≥6) have larger intensities in the experimental spectrum than in the simulated spectrum. This suggests that a loss mechanism is operative in the C'(v2'=2) vibrational state in NH₃, which is less effective for the rotational levels which are involved in the transitions whose intensity is too large in our experimental spectrum. In that case the competing ionization step produces more intensity in the electron or NH₃⁺ ion excitation spectrum. From previous studies it is known that this loss mechanism should be ascribed to predissociation.

The influence of predissociation has been observed before in the (3→1) excitation work on the C'(v2'=2) band in NH₃ obtained with nanosecond and picosecond excitation. For the best nanosecond simulation a Lorentzian line shape with a width of 0.4 cm⁻¹ (fwhm) was used, which indicates that the shape of the rotational lines is to a large extent determined by the natural line width. In the picosecond excitation spectrum, however, the best qualitative agreement with the experimental spectrum is obtained when a Gaussian line shape with a width of 12 cm⁻¹ (fwhm) is used, indicating that the shape of the rotational contour is determined exclusively by the spectral width of the UV picosecond laser pulses.

Figure 5a shows the nanosecond (2→1) REMPI excitation spectra of the C'(v2'=2) band in NH₃ obtained with nanosecond (b) and picosecond (d) excitation in combination with electron detection. The main progression, attributable to the (J",K") → (J'=J",K'"=K") transitions, is marked at the top of the figure. In the simulations of the nanosecond (a) and picosecond (c) excitation spectra a temperature of 293 K has been used. In part a Lorentzian line shapes with a width of 0.4 cm⁻¹ (fwhm) have been assumed for the transitions, whereas in part c Gaussian line shapes with a width of 12 cm⁻¹ (fwhm) have been employed.

In the experimental spectrum (Figure 5a) and the simulated spectrum (Figure 5a) it can be observed that most rotational lines are severely broadened, indicating short lifetimes (vide infra), but partly resolved rotational structure is still present, as has been reported previously for the (3→1) REMPI excitation spectrum of this state. We have tried to simulate this spectrum using

| Table 1: Optimal Fit Parameters (cm⁻¹) for the C'(v2'=2) Vibrational Level in NH₃ Obtained from the Present Study and Reported in Previous Studies |
|-----------------|-----------------|-----------------|-----------------|-----------------|
|                 | present study   | ref 26          | ref 32          | ref 41          | ref 44          |
| J"              | 65.601 ± 0.5    | 65.610          |                 |                 |                 |
| B"              | 9.877 ± 0.005   | 9.91 ± 0.01     | 9.93 ± 0.05     | 9.93 ± 0.05     | 9.88 ± 0.05     |
| C"              | 5.392 ± 0.005   | 5.36 ± 0.01     | 5.35 ± 0.1      | 5.38 ± 0.01     | 5.39 ± 0.05     |

| Vibrational Level in NH₃ Obtained from the Present Study and Reported in Previous Studies |
|---------------------------------|-----------------|-----------------|-----------------|-----------------|
| No.  | A    | B     | C     | D                      |
| 1    | 65 601.9 ± 0.5 | 9.877 ± 0.005 | 5.392 ± 0.005 |                  |
| 2    | 65 610   | 9.91 ± 0.01 | 5.36 ± 0.01 | 5.35 ± 0.1     |
| 3    | 65 610   | 9.93 ± 0.05 | 5.35 ± 0.1 | 5.38 ± 0.01     |
| 4    | 65 610   | 9.88 ± 0.05 | 5.39 ± 0.05 | 5.39 ± 0.05     |

*Note: Table 1 includes additional parameters for the vibrational levels in NH₃ obtained from the present study and reported in previous studies.
rotational constants in the simulation slightly from the values hampered by the lack of precise values for the rotational properties of the B(v2') vibrational levels.29,34~39 The two rotational level in the excited state associated with this transition, marked in Figure 6a, whose intensity is considerably underestimated in the simulation. It would seem that the rotational level in the excited state associated with this transition has a substantially longer lifetime than the other levels, as discussed above. Unambiguous identification of this level is not possible. We have therefore changed the rotational constants from the literature32 (B' = 9.49 cm\(^{-1}\), C' = 5.60 cm\(^{-1}\)). However, the peak positions could not be reproduced in this way, leading to an overall blended rotational contour, which did not resemble our experimental result. On the other hand, a detailed rotational analysis with our experimental results is not possible. We have therefore changed the rotational constants in the simulation slightly from the values in the literature, until a good agreement was obtained with the experimental spectrum (B' = 9.39 cm\(^{-1}\), C' = 5.57 cm\(^{-1}\)). It then became immediately clear that there is at least one transition, marked in Figure 6a, whose intensity is considerably underestimated in the simulation. It would seem that the rotational level in the excited state associated with this transition has a substantially longer lifetime than the other levels, as discussed above. Unambiguous identification of this level is hampered by the lack of precise values for the rotational constants in the excited state, but it would seem that it is the (J''K''=6,0) \rightarrow (J',K'=6,0) transition which dominates the experimental excitation spectrum. Figure 6b shows a simulation in which the intensity of this transition has been increased by an additional factor of 7.5. The agreement between the experimental and simulated spectra is now reasonably good.

The B state is electronically degenerate with symmetry 'E'' in D\(_{3h}\). Just as in the case of the C'(v2') states, the v2' vibrational levels in the B state have alternating vibronic symmetry: 'E'' for even v2' and 'E' for odd v2'. The two-photon transition tensors T\(_{2 \pm 1}\), which transform as 'E'' in D\(_{3h}\), carry the transition intensity for the ('A' \rightarrow 'E'' and 'A'' \rightarrow 'E') transitions, resulting in the two-photon selection rules \(\Delta J = 0, \pm 1\) and \(\Delta K = \pm 1\) for the B state. The T\(_{2 \pm 2}\) two-photon transition tensors, which transform as 'E' in D\(_{3h}\), with accompanying selection rules \(\Delta J = 0, \pm 1\), \(\pm 2\) and \(\Delta K = \pm 2\), would also possess the correct symmetry to contribute to the B two-photon excitation strength, but it has been confirmed experimentally that its contributions are negligible.27,39 Earlier studies have led to an accurate description of the spectroscopic properties of the B(v2') vibrational levels.4,29,34,39 The two degenerate components of the 'E'' electronic state are described by the vibronic quantum number \(l\), which can take the values ±1. The \(|J',K',M'|l = -1\) wavefunctions are coupled to the \(|J',K' + 2,M'|l = +1\) wave functions, resulting in additional shifts in the rotational level positions and a splitting between the rotational levels with \('A_{1}' ('A_{2}')\) and \('A_{2}' ('A_{1}')\) symmetry for the \(K' = 1\) levels associated with the vibrational members with 'E'' ('E') vibronic symmetry. Furthermore, the intensities of transitions to the rotational levels in the B(v2') states are influenced by this coupling effect, which is particularly important for rotational levels with high values of \(J'\). The coupling effect is described by an additional parameter \(q\) and is responsible for the I-type doubling in the B state.

All excitation spectra of the B(v2') vibrational levels studied in the literature show well-resolved rotational structure. Due to the perpendicular nature of the transitions to the B(v2') states, they appear in the excitation spectra with large numbers of rotational lines confined to a small energy region. This can be seen, for example, in the excitation spectra of the B(v2'=6,7) states examined in the present study (see Figure 3a). We have simulated these spectra for NH3 using our rotational contour program and the excited-state parameters as determined previously.29,39 The nanosecond excitation spectra are best described by assuming a Lorentzian-shaped line form with a width of about 0.9 cm\(^{-1}\) (fwhm) for the rotational lines. This large line width is the first indication that the lifetimes of the B state rotational levels are shorter than for the C'(v2'=1,2) states. There is qualitative agreement with the fact that the B(v2'=6,7) states in NH3 have less intensity than the C'(v2'=1,2) states in the excitation spectra. The picosecond excitation spectra of the B(v2'=6,7) vibrational members in NH3 (Figure 3b) can now be simulated quite well by changing the line shape to Gaussian with a width (fwhm) of 12 cm\(^{-1}\). The heights of the individual rotational lines in the experimental B(v2'=6,7) spectra can be reproduced satisfactorily in the simulations, and rotational-level-dependent deviations as described above for the C' states are not observed. This suggests that all rotational levels associated with one of the B(v2'=6 or 7) vibrational levels have similar lifetimes; that is, the B(v2') states show considerably less evidence of rotational-level-dependent predissociation than the C'(v2') states.

III.B. Natural Line Width Measurements. In the previous section we have been concerned with the excitation spectra of several vibrational members of the B and C' states of NH3 and ND3. In particular, it has been observed that as the result of predissociation, the intensities of rotational transitions might be reduced. The influence of such predissociation mechanisms will also be apparent in the natural lifetimes of rotational levels associated with the excited states. These lifetimes within different v2' vibrational levels in the B and C' states of NH3 and ND3 have been determined in previous studies using accurate line width measurements.27,39 The natural lifetime \(\tau\) of an excited state is correlated to the natural line width \(\Gamma\) of transitions to this level in the excitation spectrum by \(\Gamma \tau = (2\pi c)^{-1}\). In order to extract the natural line width, and hence the natural lifetime, from the excitation spectrum, several conditions must be satisfied to avoid spectral broadening caused by sources other than the natural lifetime. First and most important, the spectral width of the excitation light must be smaller than the natural line width. Secondly, mechanisms such as pressure, collisional, Doppler, and laser power broadening must be preferably excluded or corrected for. Thirdly, to determine the exact line width, a specific line shape has to be assumed for the spectral line, which introduces uncertainties in the lifetime. Furthermore, it is not always possible to observe...
an individual line as the result of blending of several rotational lines. The above conditions are not always easily satisfied, and it is often extremely difficult to measure the exact lifetimes reliably via the natural line widths. A lower limit for the lifetime, on the other hand, can in most cases be obtained without difficulty.

A typical natural lifetime in the present study on ammonia is 100 ps, corresponding to a natural line width of 0.05 cm⁻¹. To measure line widths of this magnitude, a specialized experimental setup is required. In previous studies of Ashfold et al. sub-Doppler spectroscopy has been applied using a high-power excitation source with a bandwidth of 0.008 cm⁻¹ at the excitation energies required for the B and C states in ammonia.²⁹ These studies have shown that heterogeneous predissociation plays a major role in the C(v' = 2) vibrational level of ND₃.²⁷ The natural line width of the (J',K' = 0,0) → (J,K = 0,0) transition was measured as 0.012 cm⁻¹, corresponding to a lifetime of the (J',K' = 0,0) level of 450 ± 100 ps. The uncertainty in the lifetime resulted from the corrections that needed to be applied for some of the broadening mechanisms described above, as well as from the assumptions concerning the line shape. In a study on the line widths of rotational levels associated with the B(v') states in the ground state of ND₃, the natural line width on the order of or significantly smaller than the bandwidth of our excitation source.²⁷

For the natural line width measurement of transitions to rotational levels in ammonia, the uncertainty in the lifetime resulted from the corrections that needed to be applied for some of the broadening mechanisms described above, as well as from the assumptions concerning the line shape. In a study on the line widths of rotational levels associated with the B(v') states in the ground state of ammonia, the natural line width on the order of or significantly smaller than the bandwidth of our excitation source.²⁷

III.C. Photoelectron Spectra. The REMPI photoelectron spectra (PES) for NH₃ have previously been measured using (2 + 1) and (3 + 1) ionization schemes via several B(v') and C(v') vibrational levels.³⁶,⁴³,⁵⁸,⁵⁹ Both the B and C excited states as well as the X₂A₂'' ionic ground state have planar geometries, and the REMPI photoelectron spectra are consequently expected to be dominated by Aν = 0 photoelectron peaks; that is, the ion is produced with the same vibrational quantum numbers as the excited state from which ionization took place. This has been confirmed experimentally insofar as the total energy of the photons suffices to produce the excited state-selected ions.³⁶,³⁸,⁴³,⁵⁸,⁵⁹ As the v₁ vibration is most important upon excitation from the ground state, the photoelectron spectra via the different v₁' vibrational members of the B and C states consist almost exclusively of intense Aν = 0 transitions. In a previous (2 + 1) REMPI-PES study, it has been shown that small photoelectron peaks corresponding to Aν = ±1 transitions are present as well.³⁶ In the photoelectron spectra obtained by (2 + 1) ionization via the B(v' = 3) states, a number of photoelectron peaks on the low energy side of the Aν = 0,±1 peaks were found to be responsible for a contamination of about 10% to the state-selected ionization process. The strongest of these peaks, which are not observed for (2 + 1) ionization via the C(v') states, was found in a subsequent (2 + 1) study on the B(v' = 10) state in NH₃ to be located at an energy difference of 0.40 ± 0.02 eV from the main Aν = 0 peak. For this photoelectron peak an assignment as X₂A₂''(10v¹'' + v₃'') was suggested.⁵⁹

In the present study we have measured the (2 + 1) REMPI-PES spectra via the B(v' = 2−12) and C(v' = 0−8) vibrational members in NH₃, using the nanosecond excitation source. The photoelectron spectra are, as far as the strongest photoelectron peaks are concerned, similar to the spectra obtained in previous (2 + 1) studies.³⁶,⁵⁹ A typical photoelectron spectrum obtained by excitation via the B(v' = 6) state is shown in Figure 7. The spectrum shows the Aν = 0,±1 peaks as well as the photoelectron peaks on the low energy side of the spectrum, labeled 1 and 2 in Figure 7. In all B(v') photoelectron spectra measured in the present study the latter peaks are located at 0.197 ± 0.007 eV for peak 1 and 0.404 ± 0.007 eV for peak 2 below the Aν = 0 peak. It therefore seems likely that these peaks should be assigned to a progression in the X₂A₂'' ground ionic state, which consists of the main nν₁'' progression plus an extra amount of vibrational excitation, as suggested earlier.⁵⁹ The known vibrational frequencies of the ground states of neutral ammonia and the ammonia cation are listed in Table 2. The frequencies in the ground state of the cation, determined from the present results, are also given in Table 2. The assignment of peak 1 is in the first instance problematic, since none of the known vibrational frequencies of the ground state of the NH₃⁺ ion match the measured interval of 0.197 eV. Inspection of the vibrational frequencies in the ground state of the neutral shows that the asymmetric bending vibration has a frequency of 0.200 eV in this state. In view of the similarity of these two frequencies we therefore assign peak 1 to X₂A₂''(nν₁'' + ν₁'). Evidence of activity of the ν₁'' vibration in the ground state of NH₃⁺ has also been observed recently in a high-resolution ZEKE study.⁴⁰ Due to the near-equivalence of different vibrational excitations in the ion, several assignments are possible for peak 2. The ν₃' vibrational frequency in the ionic ground state is just outside the experimental uncertainty in our results to assign

![Figure 7. Photoelectron spectrum obtained for (2 + 1) ionization via the B(v' = 6) state in NH₃ (hv = 32.499 cm⁻¹). The assignments are given at the top of the figure. The peaks labeled 1 and 2 are discussed in the text.](image-url)

| TABLE 2: Vibrational Frequencies (eV) in the Ground State of Neutral Ammonia and the Ammonia Cation As Determined in the Present and in Previous Studies |
|-----------------|----------------|----------------|----------------|
|                | NH₃⁺⁺           | NH₃⁺           | ND₃⁺           |
| v₁(a₁')        | 0.414           | 0.300          | 0.391 ± 0.012e |
| v₂(a₁')        | 0.116           | 0.093          | 0.111d         |
| v₃(a₁')        | 0.120           | 0.100          | 0.090d         |
| v₁(c₁')        | 0.423           | 0.317          | 0.420d         |
| v₂(c₁')        | 0.202           | 0.148          | 0.197 ± 0.007e |
|                |                 |                | 0.141 ± 0.007e |
| *Reference 53.  |                   |                |                |

+1) study on the B(v' = 10) state in NH₃ to be located at an energy difference of 0.40 ± 0.02 eV from the main Aν = 0 peak. For this photoelectron peak an assignment as X₂A₂''(10v₁'' + v₃'') was suggested.⁵⁹

In the present study we have measured the (2 + 1) REMPI-PES spectra via the B(v' = 2−12) and C(v' = 0−8) vibrational members in NH₃, using the nanosecond excitation source. The photoelectron spectra are, as far as the strongest photoelectron peaks are concerned, similar to the spectra obtained in previous (2 + 1) studies.³⁶,⁵⁹ A typical photoelectron spectrum obtained by excitation via the B(v' = 6) state is shown in Figure 7. The spectrum shows the Aν = 0,±1 peaks as well as the photoelectron peaks on the low energy side of the spectrum, labeled 1 and 2 in Figure 7. In all B(v') photoelectron spectra measured in the present study the latter peaks are located at 0.197 ± 0.007 eV for peak 1 and 0.404 ± 0.007 eV for peak 2 below the Aν = 0 peak. It therefore seems likely that these peaks should be assigned to a progression in the X₂A₂'' ground ionic state, which consists of the main nν₁'' progression plus an extra amount of vibrational excitation, as suggested earlier.⁵⁹ The known vibrational frequencies of the ground states of neutral ammonia and the ammonia cation are listed in Table 2. The frequencies in the ground state of the cation, determined from the present results, are also given in Table 2. The assignment of peak 1 is in the first instance problematic, since none of the known vibrational frequencies of the ground state of the NH₃⁺ ion match the measured interval of 0.197 eV. Inspection of the vibrational frequencies in the ground state of the neutral shows that the asymmetric bending vibration has a frequency of 0.200 eV in this state. In view of the similarity of these two frequencies we therefore assign peak 1 to X₂A₂''(nν₁'' + ν₁'). Evidence of activity of the ν₁'' vibration in the ground state of NH₃⁺ has also been observed recently in a high-resolution ZEKE study.⁴⁰ Due to the near-equivalence of different vibrational excitations in the ion, several assignments are possible for peak 2. The ν₃' vibrational frequency in the ionic ground state is just outside the experimental uncertainty in our results to assign

![Figure 7. Photoelectron spectrum obtained for (2 + 1) ionization via the B(v' = 6) state in NH₃ (hv = 32.499 cm⁻¹). The assignments are given at the top of the figure. The peaks labeled 1 and 2 are discussed in the text.](image-url)
this peak to \( ^2 \tilde{A}_2''(n_3v_2^++v_4^+) \). Several \( \tilde{C} \) \( 1^2 \nu \) bands in the excitation spectra of \( \text{NH}_3 \) and \( \text{ND}_3 \) have been observed by Miller et al. in a \( (2 + 1) \) REMPI-PES study. From the photoelectron spectra, obtained for ionization via these bands, the \( \nu_1^+ \) vibrational frequency in the ground state of \( \text{NH}_3^+ \) was determined as \( 0.391 \pm 0.012 \) eV. In the present study we have remeasured the \( \nu_1^+ \) frequency from the photoelectron spectra obtained at the \( \tilde{C} \) \( 1^2 \nu \) bands and obtained a value of \( 0.401 \pm 0.007 \) eV, in agreement with the value of Miller et al. Peak 2 in the photoelectron spectra obtained for excitation via the \( \tilde{B}(\nu_2) \) states can then be assigned to \( ^2 \tilde{A}_2''(n_3v_2^++v_4^+) \). It is, however, also possible to assign this peak to \( ^2 \tilde{A}_2''(n_2v_2^++2v_4^+) \), because the energy difference between peak 2 and the \( \Delta \nu_2 = 0 \) peak is exactly twice the energy difference between peak 1 and the \( \Delta \nu_2 = 0 \) peak. In view of the assignments of the \( \text{ND}_3 \) photoelectron spectra (vide infra) we favor the former \( ^2 \tilde{A}_2''(n_3v_2^++v_4^+) \) assignment.

To our knowledge there have been no previous reports of photoelectron spectra obtained via \( \tilde{B}(\nu_2) \) or \( \tilde{C}(\nu_2) \) vibrational levels of \( \text{ND}_3 \). In the present study we have measured the \( (2 + 1) \) REMPI photoelectron spectra of the \( \tilde{C}(\nu_2=0-7) \) vibrational members in \( \text{ND}_3 \). As in the case of \( \text{NH}_3 \), the \( \Delta \nu_2 = 0 \) ionization process dominates these spectra, and deviations in the form of \( \Delta \nu_2 = \pm 1 \) photoelectron peaks are small. We have also examined the photoelectron spectra obtained for ionization via several \( \tilde{B}(\nu_2) \) states in \( \text{ND}_3 \) to investigate the behavior of the peaks which are equivalent to the peaks labeled 1 and 2 in the \( \text{NH}_3 \) photoelectron spectra. These photoelectron spectra are very similar to the \( \text{NH}_3 \) spectra, as can be seen from Figure 8a, which shows the photoelectron spectrum obtained for ionization via the \( \tilde{B}(\nu_2=8) \) vibrational level. The peaks labeled 1 and 2, located at photoelectron energies of \( 0.141 \pm 0.007 \) and \( 0.304 \pm 0.007 \) eV below the main \( \Delta \nu_2 = 0 \) photoelectron peak, are equivalent to the peaks labeled 1 and 2 in the \( \text{NH}_3 \) spectra. For higher members of the \( \tilde{B}(\nu_2) \) progression the photoelectron spectra show an increasing \( \nu_2^+ \) activity other than \( \Delta \nu_2 = 0 \). Figure 8b shows the photoelectron spectrum obtained for excitation via \( \tilde{B}(\nu_2=12) \). The dominant \( \Delta \nu_2 = 0 \) photoelectron peak is cut off in Figure 8b in order to emphasize the weaker photoelectron peaks. The positions of the photoelectron peaks in Figure 8b are given in the Table 3, along with their assignments. Apart from the \( \Delta \nu_2 = 0 \pm 1 \) photoelectron peaks, \( \Delta \nu_2 = \pm 2 \) transitions are now also observed, as well as the peaks labeled 1' and 2'. In the spectrum an extra peak is present, which was not observed in the previously discussed photoelectron spectra. This peak, with a photoelectron energy of \( 0.980 \) eV, is located \( 0.390 \) eV from the \( \Delta \nu_2 = 0 \) peak and can therefore be assigned to \( ^2 \tilde{A}_2''(13v_2^++v_4^+) \), i.e., the \( \Delta \nu_2 = +1 \) peak with an additional \( \nu_1^+ \) quantum. The photoelectron peak corresponding to ions in the \( ^2 \tilde{A}_2''(13v_2^++v_4^+) \) state is not observed. In contrast to \( \text{NH}_3 \), the energy difference between peak 2' in \( \text{ND}_3 \) and the main \( \Delta \nu_2 = 0 \) peak is no longer exactly twice the energy difference between peak 1' and the \( \Delta \nu_2 = 0 \) peak. On the basis of the above observations we favor the assignment of \( ^2 \tilde{A}_2''(n_3v_2^++v_4^+) \) for peak 2' (peak 2 in \( \text{NH}_3 \)) over the \( ^2 \tilde{A}_2''(n_2v_2^++2v_4^+) \) assignment. The values reported in the present study for the \( \nu_1^+ \) and \( v_4^+ \) vibrations in the ground state of \( \text{ND}_3^+ \) are listed in Table 2 and are to our knowledge the first determinations of these frequencies in \( \text{ND}_3^+ \).

As a result of the pyramidal to planar geometry change, the He I photoelectron spectrum of the \( ^2 \tilde{A}_2'' \) ionic ground state shows the same \( v_2^+ \) activity which is also observed in the excitation spectra of the \( \tilde{B}(\nu_2) \) and \( \tilde{C}(\nu_2) \) Rydberg states. The \( v_2^+ \) progression peaks at approximately \( v_2^+=7 \), and the \( v_2^+=0 \) photoelectron peak at the position of the adiabatic ionization energy has a very low intensity. The exact numbering of this progression in the ion, and hence the value of the adiabatic ionization energy, has been the subject of recent discussion. In an earlier study the adiabatic ionization energy was placed at \( 10.183 \) eV, and an additional photoelectron peak at a lower energy of \( 10.073 \) eV was interpreted as the \( 2^0 \) hot band. These assignments have been questioned in a recent study in which the adiabatic ionization energy was set at \( 10.073 \) eV, thus changing the \( v_2^+ \) numbering of all peaks in the main progression by one quantum. In the present study we have been able to identify the photoelectron peak at \( 10.183 \) eV as the adiabatic ionization energy on the basis of the \( (2 + 1) \) REMPI-PES results via the \( \tilde{B}(\nu_2) \) and \( \tilde{C}(\nu_2) \) states and the assumptions that the \( \nu_2^+ \) vibrational numbering in these excited Rydberg states is correct and that ionization of these states indeed occurs with a \( \Delta \nu = 0 \) propensity. This \( \nu_2^+ \) numbering is also in agreement with the results from a recent ZEKE study on \( \text{NH}_3 \).

In the photoelectron spectra obtained with \( (2 + 1) \) REMPI via the \( \tilde{B} \) or \( \tilde{C} \) states we have not observed differences between the spectra obtained with nanosecond and picosecond excitation, in agreement with a previous study in which picosecond excitation was used. The photoelectron peaks deriving from ionization via \( \tilde{B}(\nu_2) \) and \( \tilde{C}(\nu_2) \) states, at excitation energies where different vibrational levels of these two states overlap,
are always separated by an amount which enables unambiguous identification of the state under investigation. This energy difference \((n - m)\omega^+\) is about \(5\omega^+\) for NH₃ and \(6\omega^+\) for ND₃. The simple appearance of the photoelectron spectra allows for an immediate identification of the \((2 + 1)\) and \((2 + 1)\) electron signals in the two-color experiments using picosecond excitation, even in cases where different \(B(v')\) and \(C(v')\) vibrational members overlap.

III.D. Picosecond Real-Time Lifetime Measurements. In a previous section it has been described how the natural lifetimes of rotational levels associated with the vibrational components of the \(B\) and \(C\) Rydberg states could be determined indirectly from accurate natural line width measurements. In the present study the lifetimes have been measured in real time in a pump-probe setup, which has been described above. The pump pulse excites the rotational levels associated with the \(B(v')\) or \(C(v')\) Rydberg state under investigation, while the probe pulse subsequently ionizes the molecule in a one-photon absorption step with a strong \(\Delta v = 0\) propensity. The decrease in the \((2 + 1)\) two-color photoelectron signal is monitored as a function of the delay time between the pump and probe pulses and reflects the depopulation of the excited state. In the present study we have examined in real time the lifetimes of the rotational levels in the \(B(v')\) and \(C(v')\) vibrational members, for which the excitation spectra are shown in Figures 3 and 4.

A typical decay trace is shown in Figure 9, where the \((2 + 1)\) decay obtained for ionization via the NH₃ \(B(v' = 6)\) vibrational level is depicted. Due to the large excitation bandwidth of the picosecond laser system, it is expected that several rotational levels within the \(B(v' = 6)\) vibrational level are coherently excited, as shown above in the discussion of the picosecond excitation spectra. The decay trace in Figure 9 is determined by the convolution of the system response and the exponential decay, and its analysis may serve as an example for the other decay traces measured in the present work. The \((2 + 1)\) electron signal first rises to a maximum value around delay time zero, with a temporal behavior determined by both the system response and the natural lifetime of the excited state. In order to extract the lifetime of the excited state from Figure 9, a deconvolution procedure is required. The traces are assumed to be described by a convolution of a Gaussian-shaped system response function and a mono- or biexponential decay and are fitted using a nonlinear least-squares algorithm. The traces are thus described by the following parameters: the position of delay time zero, the width (fwhm) of the Gaussian system response, the preexponential factor(s), and the decay time(s) \(\tau_i\) for \(i\). The best fit for the \(B(v' = 6)\) state, using a monoexponential decay with a time constant of 9.3 ps, is shown in Figure 9 as the smooth solid curve and describes the measured decay quite adequately. The experimental traces are found to be less accurately described when other line shapes, such as Lorentzian or sech², are assumed for the system response. It was typically found that the width of the Gaussian-shaped system response varied somewhat from day to day (±10%), but it was always nearly constant within one measurement session.

Figure 10 shows the decay trace of the NH₃ \(C(v' = 3)\) vibrational state together with the best fit. In this case the femtosecond dye laser has been used to produce the pump pulse, and its output was frequency-mixed with the output of the regenerative amplifier to give the probe pulse. The time resolution of 2.5 ps is considerably better than in Figure 9, and the use of a monoexponential decay results in a decay time of 15 ps.

Figure 11 shows a series of five decay traces (labeled 1–5) obtained at different wavelengths in the excitation spectrum of the ND₃ \(C(v' = 3)\) vibrational level. The one-photon energy difference between two adjacent traces is about 6 cm⁻¹. Trace number 1 has been recorded at the excitation wavelength which gives the maximum signal. From the simulations of the excitation spectra described above we know that this wavelength corresponds to transitions to rotational levels with low values for \(J, K\) in the excited state. It can be observed that these levels have comparatively long lifetimes, and, as a consequence, the ionization signal has not decreased to zero at the maximum delay imposed by the finite length of the translational stage. This fact introduces uncertainties in the exponential decay parameter(s) that result from the fitting procedure. The lifetime uncertainty in decay traces where the natural lifetime is on the same order as the width of the system response is to a major extent determined by assumptions about the shape of the system response. For long decays of the kind shown in Figure 11
When the pump wavelength is tuned toward the region in the excitation spectrum where transitions to higher monoexponential decay. The traces labeled 2-5 in Figure 11 decay times of $t_1$, $t_2$, respectively. The traces labeled 3, 4, and 5 are biexponential with decay times of $t_1 = 100$ ps and $t_2 = 500$ ps for 3, $t_1 = 46$ ps and $t_2 = 360$ ps for 4, $t_1 = 26$ ps and $t_2 = 500$ ps for 5. The decay labeled 1 has been obtained using an excitation wavelength of $32,980$ cm$^{-1}$. In each subsequent trace the one-photon energy has been decreased by $6$ cm$^{-1}$.

Assumptions about the system response are of no importance. When the pump wavelength is tuned toward the region in the excitation spectrum where transitions to higher $J',K'$ rotational levels are located, it can be seen from Figure 11 that the temporal behavior can no longer be described accurately by a monoexponential decay. The traces labeled 2-5 in Figure 11 turn out to require a description by a biexponential decay. This is particularly clear in the traces numbered 4 and 5 in Figure 11.

Biexponential decay has been observed for the $C(v_2'=3,4,5)$ vibrational levels in ND$_3$. This effect can be understood qualitatively if it is assumed that several transitions to different $J',K'$ rotational levels are coherently excited by the picosecond laser pulse and contribute to the ionization signal. In principle, one would expect that if several transitions to rotational levels in the excited state with different lifetimes are coherently excited, the decay traces would be described by a multieponential decay. However, for the traces measured in the present study one or two time constants were sufficient to describe the temporal decay behavior. These observations suggest that rotational levels with high $J',K'$ values lying at lower energies in the excitation spectra of the $C(v_2'=3,4,5)$ vibrational members in ND$_3$ have shorter lifetimes than the rotational levels with low $J',K'$ values.

In the way described above we have measured in real time the natural lifetimes of the rotational levels associated with several vibrational states of the $B$ and $C$ states in NH$_3$ and ND$_3$. The results are listed in Table 4. The rotational levels are in some cases described as $J',K'$ low and $J',K'$ high for reasons mentioned above in the discussion of Figure 11. If the decay behavior is monoexponential, only one lifetime is listed; in case of a biexponential decay the second time constant is given as well.

If the fourth harmonic of Nd:YLF from the regenerative amplifier is used as the probe pulse, $(2 + 1')$ excitation spectra at different fixed delay times between the pump and probe pulses can be obtained. Even though the time resolution is in this case only about 30 ps, these excitation spectra can provide valuable information, as will become clear from the following. Figure 12a shows two $(2 + 1')$ picosecond excitation spectra of the ND$_3$ $C(v_2'=4)$ vibrational band obtained at fixed delay times of 0 and 250 ps, as marked in the figure. The excitation spectrum at delay time 0 ps corresponds to the regular one-color excitation spectrum shown before in Figure 4. In Figure 12a only the photoelectrons originating from ionization of $C(v_2'=4)$ have been monitored, whereas in Figure 4 the total electron signal, which includes electrons deriving from ionization of the overlapping $B(v_2'=10)$ level, was shown. The most striking difference between the excitation spectrum obtained at a delay time of 0 ps and the spectrum obtained at a fixed delay of 250 ps is the difference in width. In the latter excitation spectrum the transitions to the rotational levels with high values for $J',K'$ have a considerably reduced intensity with respect to the intensity of the transitions to the rotational levels with low $J',K'$ values.
values for \(J', K'\). This shows unambiguously that the rotational levels with high \(J', K'\) have shorter lifetimes than the rotational levels with low \(J', K'\). The same behavior is observed for the ND\(_3\) \(C'(v_2'=3,5)\) levels and is in qualitative agreement with the above observation of a biexponential decay for excitation on the low energy side of the maximum. Similar \((2 + 1)\) excitation spectra of the ND\(_3\) \(C'(v_2'=2)\) vibrational member at different fixed delay times between the pump and probe pulses show no dependence on the delay time, again in agreement with the real-time lifetime results from Table 4, which do not show biexponential decay profiles at any excitation wavelength for this state.

Two \((2 + 1)\) excitation spectra of the NH\(_3\) \(C'(v_2'=1)\) state obtained at fixed delay times of 0 and 70 ps are shown in Figure 12b and reveal the opposite behavior as observed for the ND\(_3\) \(C'(v_2'=3,4,5)\) levels. The peak with maximum intensity in the spectrum obtained at delay 0 ps, corresponding to transitions to low \(J', K'\) rotational levels, decreases rapidly in intensity as the delay time is increased, whereas the peak at lower excitation energy, for which the transition to the \(J', K' = 6,6\) rotational level is mainly responsible, maintains its intensity up to a delay time of 70 ps. This behavior is in agreement with the results given in Table 4, which show that the rotational levels around \(J', K' = 6,6\) in the excited state have lifetimes of about 100 ps, whereas the low \(J', K'\) rotational levels have considerably shorter lifetimes of about 25 ps.

### III.E. Comparison between Picosecond and Nanosecond Results

In this section the results from the present, real-time, natural lifetime measurements obtained using the picosecond excitation source will be compared to the results obtained with nanosecond excitation in this study as well as in previous studies. From Table 4 we observe that rotational levels with high \(J', K'\) associated with the \(C'(v_2'=1,2)\) vibrational members of NH\(_3\) have a longer lifetime than rotational levels with low \(J', K'\). The same conclusion has been drawn above from the nanosecond excitation spectra of these states in which the intensities of transitions to rotational levels with high \(J', K'\) were found to be larger than expected on the basis of the simulations (see also Figure 5). The long lifetime determined for the NH\(_3\) \(C'(v_2'=3)\) state (see Table 4 and Figure 10) would seem to be at odds with the weak and diffuse appearance of this band in the nanosecond excitation spectrum. In the previous sections it was shown, however, that at least one rotational level, presumably the \(J', K' = 6,0\) level, appears to have a considerably longer lifetime than the other rotational levels (see Figure 6). If the transition to this level dominates the picosecond excitation spectrum as well, it can be understood that its lifetime of 15 ps dominates the picosecond decay trace, whereas the short lifetimes of the other rotational levels are responsible for the weak and diffuse appearance of the nanosecond excitation spectrum.

The lifetimes of the rotational levels associated with the ND\(_3\) \(C'(v_2'=2)\) vibrational state determined by line width measurements in a previous study were shown to depend on the rotational quantum numbers \(J', K'\). The line widths of the individual rotational levels could be modeled by the equation \(\omega = \omega_0[1 + 0.030(J' + 1) - (K' - 1)^2]\), in which the natural line width of the \((J', K' = 0,0)\) transition \(\omega_0 = 0.012\) cm\(^{-1}\), corresponding to a natural lifetime of 450 ps of the \((J', K') = (0,0)\) rotational level. This equation shows that the largest changes should occur for the rotational levels for which \(K' \ll J'\), such as \(J', K' = 7,1\) and that the differences in lifetime for levels with \(J' \sim K'\) are only minor. The rotational contour simulations of the \(C'(v_2')\) states have shown that transitions to the latter type of rotational levels, i.e., \((J', K') \rightarrow (J = J', K' = K')\), tend to contribute the most intensity to the excitation spectrum. It can thus qualitatively be understood that we observe no rotational level dependence in the picosecond decay measurements of the ND\(_3\) \(C'(v_2'=2)\) vibrational level, whereas such a dependence was concluded from the line width measurements.

Our decay measurements on the higher members of the \(C'\) progression in ND\(_3\), i.e., \(C'(v_2'=3,4,5)\), clearly show that the higher \(J', K'\) rotational levels are shorter lived than the low \(J', K'\) levels (see Table 4, Figures 11 and 12). This effect has been observed previously in the \((3 + 1)\) excitation spectra of ND\(_3\), where it affected the peak heights. The conclusion of earlier studies\(^{26,27,45}\) that a rotationally dependent predissociation mechanism is operative in the \(v_2'\) vibrational members of the \(C'\) state in NH\(_3\) and ND\(_3\) is confirmed in the present results. The effect of a second, homogeneous, predissociation mechanism, which has been concluded to be important for the higher \(C'(v_2' \geq 3)\) vibrational members in NH\(_3\) on the basis of the nanosecond excitation spectra, is not observed in our picosecond real-time measurements on NH\(_3\) due to the reasons discussed above. In ND\(_3\), however, this homogeneous predissociation mechanism is observed to be responsible for a decrease in the natural lifetime as higher \(v_2'\) vibrational levels of the \(C'\) Rydberg state are addressed (see Table 4).

For the \(B(v_2')\) states of NH\(_3\), Table 4 shows lifetimes of the rotational levels which are on the order of 6.8–9.4 ps and only slightly dependent on \(v_2'\), \(J'\), and \(K'\). This is in agreement with the peak heights in the nanosecond excitation spectra as well as with the line width measurements in the present study and a previous study on the \(B(v_2')\) states by Ashfold et al.\(^{29}\) Similar arguments apply to the \(B(v_2')\) states in ND\(_3\), but there is an additional point of interest. In the study of Ashfold et al. the line widths of the rotational lines in the \(B(v_2'=0–6)\) transitions indicated natural lifetimes of 250 ± 20 ps, independent of \(v_2', J'\), and \(K'\). In the line width study evidence was found, however, that the lifetime of the \(B(v_2'=8)\) state in ND\(_3\) is substantially shorter than the lifetime of the \(C'(v_2'=2)\) state and also shorter than that of the lower \(v_2'\) members (\(v_2' \leq 6\)) of the \(B(v_2')\) progression. We find in our picosecond real-time results lifetimes for the \(B(v_2'=8,10)\) vibrational levels of about 75 ps, independent of \(v_2', J'\), and \(K'\). Our results are in agreement with the results of Ashfold et al. In both studies there appears to be a sudden lowering of the natural lifetime in the \(B(v_2')\) progression in ND\(_3\) for \(v_2' \geq 8\), which is unexpected in view of the apparent overall independence of the natural lifetimes on \(v_2'\). The implications of the observed \(B(v_2')\) and \(C'(v_2')\) predissociation dynamics for the identification of the states responsible for the predissociation process have been discussed by Ashfold et al.\(^{26,29}\) Briefly, the \(A'(1A'')\) Rydberg state has been identified as the main cause for the predissociation of the higher-lying Rydberg states: the \(B\) and \(C'\) Rydberg states are coupled to relatively short-lived rovibronic levels associated with this \(A\) state via vibronic operators \(q_i\) or rovibronic operators \(q_iJ_2\), of which the latter are expected to give much smaller matrix elements than the former. The present as well as previous results\(^{29}\) suggest that the predissociation behavior of the rotational levels associated with the \(B(v_2')\) vibrational levels is dominated by a coupling via the \(q_i\) vibronic operator to dissociative rovibronic levels of the \(A'(1A'')\) state. The natural lifetimes of the rotational levels are thus independent of \(v_2'\), since the coupling operator does not contain \(q_i\), and also independent of the excited-state rotational quantum numbers \(J', K'\), because the coupling operator does not include \(J_2\). However, the observation that the lifetime of the rotational levels...
associated with the B(v2') states in ND3 decreases abruptly from about 250 ps for v2' = 0-6 to about 75 ps for v2' = 8.10 finds no explanation within this model. The predissociation dynamics of the rotational levels associated with the C'(v2') states are determined by two coupling mechanisms which might be equally important. The first coupling mechanism couples C'(v2') states via the q2r2 vibronic operator to dissipative vibronic levels of the A' 1A'' state and is responsible for the J, K dependence of the natural lifetimes of the rotational levels in the C' state. The second coupling mechanism couples C'(v2') states via the q2 vibronic operator to relatively long lived vibronic levels in the A' 1A'' state and results in the experimentally observed v2'-dependent homogeneous predissociation in the C'(v2') vibrational members.

In a previous section it was shown that the photoelectron spectra obtained via the B(v2') vibrational members in NH3 and ND3 were dominated by electrons corresponding to ions in the X 2A2'(n~2f) state, while weaker electron signals, corresponding to ions in the X 2A2'(n~2f+1v4') and X 2A2'(n~2f+2v4') states, were identified as well. The appearance of these weak signals is surprising because, within the Born-Oppenheimer approximation, ionization to these vibrational levels in the X 2A2' ionic ground state should be strictly forbidden. The activity of the v4' vibrational could in principle become allowed as a result of the Jahn-Teller effect in the B state.29,33,39 A tentative alternative explanation may be found in the coupling mechanism of the B(v2') states to vibronic levels associated with the A' 1A'' Rydberg state. If the B(v2') states are vibronically coupled to the A' 1A'' states via qa, on which symmetry grounds would be possible, then the photoelectron spectra obtained via the B(v2') states would contain, apart from the b2v = 0 photoelectron peak, also v4' activity. Similarly, vibronic coupling via v2 would induce v2' activity.

IIIF. Fragmentation Mechanisms. The predissociation dynamics of the A' 1A'' Rydberg state of ammonia have been studied extensively in the past.47,101-24 Several of these studies have investigated which fragment species are produced and how the excess energy is redistributed among the translational, vibrational, and rotational degrees of freedom of the fragments. Such studies do not exist for the predissociation dynamics of the B and C' Rydberg states, but as the dynamical behavior of these states is believed to be determined to a major extent by the predissociation behavior of the A' state, their fragmentation is expected to resemble the A' state fragmentation with the important difference that considerably more excess energy is available after fragmentation of the B and C' states. Once a specific rotational level of the B(E'', v2' is even; E', v2' is odd) or C'(A', v1' is even; A', v1' is odd) has been excited in a two-photon absorption step, the following competing processes can occur:

\[ \text{NH(D)} + \text{B}(E'', E') + \text{C}'(A', A') \rightarrow \text{NH(D)} + \text{B}(E', E') + \text{C}'(A', A') + \text{h} \nu \]

\[ \text{NH(D)} + \text{B}(E'', E') \rightarrow \text{NH(D)} + \text{B}(E', E') + \text{h} \nu \]

\[ \text{NH(D)} + \text{B}(E', E') \rightarrow \text{NH(D)} + \text{B}(E', E') + \text{h} \nu \]

Up until now we have almost exclusively discussed the ionization process (1) in our experimental results. The predissociation (2) and fluorescence (3) processes compete with the ionization process. Two-photon excitation spectra of the C'(v2'=0), C'(v2'=2) vibrational levels have been obtained previously by monitoring the strong fluorescence (3) to the A' state. As the predissociation channel (2) becomes progressively more important for the states which lie at higher energies, the parent molecule fluorescence disappears and the fluorescence of the NH(D): A'(2A) fragment appears. Even though the quantum yield for this species is only a few percent, its fluorescence has been observed to be modulated by the B(v2') and C'(v2') vibrational members.18,49-51,64

In the present study only positive ions or electrons can be detected, and in order to observe a neutral predissociation fragment, it needs to be ionized first. The NH(D): X(2B) fragment has been reported to produce REMPI signals in a previous study,23 but in the present experiments we have concentrated on the hydrogen atom. If the laser wavelength of the nanosecond dye laser is tuned to 34 275 cm-1, the rotational levels in the wing of the C'(v2'=5) vibrational level in NH3, about 100 cm-1 lower in energy than the band maximum, are excited in a two-photon absorption process. From this level the molecule can be ionized by process (1) or predissociate by process (2). On predissociation the H(1s 2S) atom can subsequently be ionized in a (3 + 1) REMPI process via the \( ^4P ^2P _{1/2,3/2} \) states using the same wavelength with which NH3 was excited and ionized.55 Under the experimental conditions where the electrons from the H atom ionization can be observed, the electrons deriving from ionization of the parent molecule produce a signal, which is at least 1 order of magnitude stronger. It is nevertheless possible to observe the electrons deriving from the ionization of the hydrogen atoms. From the Doppler broadening of the corresponding peak in the photoelectron spectrum it can be inferred that the neutral hydrogen atoms obtain a kinetic energy of about 0.3 eV upon predissociation. The same experiment using picosecond excitation failed to produce the electrons deriving from the ionization of the hydrogen atoms. Even in a two-color picosecond experiment, in which the first color was tuned to the (3 + 1) resonance in atomic hydrogen as described above and the second color to the maximum of the C'(v2'=5) band, neither electrons in the photoelectron time-of-flight spectrum nor ions in the mass-resolved ion spectrum attributable to ionization of atomic hydrogen could be observed. The fact that atomic hydrogen signals are not observed when picosecond excitation is used possibly arises from a less efficient competition of the predissociation mechanism with the parent ionization process within the picosecond excitation pulse.

In a two-color experiment using nanosecond excitation in combination with mass-resolved ion detection, we have monitored the production of hydrogen atoms from the predissociation of the Rydberg states. In these experiments one of the dye lasers is tuned to a wavelength of 41 129.5 cm-1, corresponding to the strong two-photon 1s 2S1/2 \( \rightarrow \) 2s 2S1/2 transition of the H atom,65-67 while the second dye laser is scanned over various v2' vibrational levels of the B and C' states. In agreement with a previous study we find that the mass-resolved ion spectrum contains only NH3+, ions, when only the laser beam which excites the B(v2') and C'(v2') vibrational levels is present.58 When the laser beam which is set at the H atom resonance is present as well, an additional H+ signal is observed, which disappears completely when either laser beam is blocked. Under optimum experimental conditions for the H+ signal, the NH3+ signal, which results from the (2 + 1) one-color REMPI process, is about 1 order of magnitude larger. The H+ signal is observed to be modulated by the B and C' resonances. Figure 13 shows the excitation spectrum of the C'(v2'=2) vibrational level in NH3 for two competing channels, viz., the ionization channel and...
The use of high-resolution photoelectron spectroscopy has enabled the identification of all photoelectron peaks in the (2 + 1) REMPI–PES spectra obtained for excitation via the \(B(v')\) vibrational members in NH\(_3\) and ND\(_3\). The vibrational frequencies for the symmetric stretch vibration \(v_s\) and for the asymmetric bending vibration \(v_b\) in NH\(_3\)+(ND\(_3\)) as determined from these spectra are 0.404 \(\pm\) 0.007 eV (0.304 \(\pm\) 0.007 eV) and 0.197 \(\pm\) 0.007 eV (0.141 \(\pm\) 0.007 eV), respectively.

In a two-color experiment using nanosecond excitation the hydrogen atom, which results from the predissociation process in the \(B\) and \(C\) Rydberg states, has been observed. The excitation spectra of the competing predissociation and ionization channels in these Rydberg states agree well with the conclusions obtained from the lifetime measurements.

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### References and Notes

$\tilde{B} \ ^1E''$ and $\tilde{C} \ ^1A_1$ States of Ammonia


