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Comment on “Spontaneous Emission of Organic Molecules Embedded in a Photonic Crystal”

A recent Letter of Petrov et al. [1] addresses the modification of the spontaneous emission of luminescent molecules embedded in a photonic crystal. They report both accelerated and inhibited spontaneous emission, with a factor of ~2 difference evident between the decay components. The central point of their Letter is that the photonic crystal (i.e., its dielectric contrast) is responsible for this effect. We show below why this “photonic band gap effect” cannot be responsible for their observations.

The photonic crystal of Petrov et al. was formed by impregnating an artificial opal of porous silica globules with a polymer (PMMA) containing a fluorescent dye. The ratio of the dielectric constants of the porous silica and the PMMA is 1.3, indicative of a photonic crystal that weakly interacts with light. Indeed, a narrow stop band in the fluorescence emission spectrum of about 6% relative width (FWHM) is observed [their Fig. 3(b)]. The inhibition of fluorescence emission spectrum of about 6% relative width, as estimated by the fraction of solid angle covered by such stop bands—but only for measurements performed at the blue edge of the stop band. The solid angle covered is largest here. This may be seen using a simple Bragg diffraction analogy, that is independent of the network topology: as the wavelength decreases from the red to the blue edge of the stop band, the Bragg condition is satisfied for increasingly larger cones of angles about normal incidence. (A further decrease in wavelength opens up a hole in the middle of the cone because the Bragg condition is no longer satisfied at normal incidence.) Surprisingly, Petrov et al. report measurements at a wavelength of 510 nm, which is at the red edge of the stop band, where they should not see any inhibition.

A more sophisticated estimate with both accelerated (at the red edge) and inhibited (at the blue edge) spontaneous emission follows from photon density of states (DOS) calculations, such as those by Busch and John [2]. Calculations for an opal with a much greater dielectric contrast (2.1 instead of 1.3; see Fig. 4 of Ref. [2]) give a maximum difference between accelerated and inhibited decay components of only 16%, not the near factor of 2 difference evident between the decay kinetics of the two curves in Fig. 1, extending over more than three decades in intensity. We must conclude, both for theoretical and experimental reasons, that the modification of fluorescence decay times reported by Petrov et al. cannot be attributed to photonic effects.

The dye is covalently attached to the silica spheres and covered by a silica layer. The chief advantage of our system is the ability to confidently compare the crystal with a reference system consisting of the same spheres in a random arrangement. This means that we can distinguish photonic effects on the emission rate from other effects, such as chemical interactions between the dye and the environment. In Fig. 1, we show the time-resolved fluorescence of dye molecules within the photonic crystal and the reference, taken at wavelengths near the blue edge of a stop band. Regrettably, similar raw data are not shown in Ref. [1]. Observe the close similarity between the decay kinetics of the two curves in Fig. 1, extending over more than three decades in intensity. We must conclude, both for theoretical and experimental reasons, that the modification of fluorescence decay times reported by Petrov et al. cannot be attributed to photonic effects.

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