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Advancements in effect-based water quality assessment

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Publication date

2020

Document Version

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Citation for published version (APA):

de Baat, M. L. (2020). *Advancements in effect-based water quality assessment*.

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CHAPTER

SMARTER SEDIMENT SCREENING: EFFECT-BASED QUALITY ASSESSMENT, CHEMICAL PROFILING, AND RISK IDENTIFICATION

5

ABSTRACT

Sediments play an essential role in the functioning of aquatic ecosystems but simultaneously retain harmful compounds. However, sediment quality assessment methods that consider the risks caused by the combined action of all sediment-associated contaminants to benthic biota are still underrepresented in water quality assessment strategies. Significant advancements have been made in the application of effect-based methods, but methodological improvements can still advance sediment risk assessment. The present study aimed to explore such improvements by integrating effect-monitoring and chemical profiling of sediment contamination. To this end, 28 day life cycle bioassays with *Chironomus riparius* using intact whole sediment cores from contaminated sites were performed in tandem with explorative chemical profiling of bioavailable concentrations of groups of legacy and emerging sediment contaminants to investigate ecotoxicological risks to benthic biota. All contaminated sediments caused effects on the resilient midge *C. riparius*, stressing that sediment contamination is ubiquitous and potentially harmful to aquatic ecosystems. However, bioassay responses were not in line with any of the calculated toxicity indices, suggesting that toxicity was caused by unmeasured compounds. Hence, this study underlines the relevance of effect-based sediment quality assessment and provides smarter ways to do so.

INTRODUCTION

Sediments play an indispensable role in the functioning of aquatic ecosystems because benthic organisms drive ecosystem processes supporting biogeochemical cycling and therewith the entire aquatic food web.¹³⁹ Simultaneously, sediments are also the largest chemical repositories on earth, where harmful, often persistent hydrophobic compounds accumulate and are retained long after the pollution of the overlying water has decreased.²⁶ The vast variety of these sediment-associated contaminants may exert harmful effects on the benthic community and can thereby impair ecosystem functioning.^{26,140} Despite this apparent threat to aquatic ecosystems, the European Union Water Framework Directive (EU-WFD)³⁶ mentions water on 373 occasions, but sediment only seven times, and does not require the member states to monitor sediment quality.²⁷ When performed at all, water authorities most often monitor sediment quality by means of chemical target analysis, focusing on only a limited set of specific compounds, potentially overlooking ecotoxicological risks caused by the myriad of (un) known mixtures of sediment-associated compounds.¹⁴¹ Therefore, there is a need for integrated sediment quality assessment methods that consider the risks caused by the combined action of all sediment-associated contaminants to benthic biota.

In recent years, significant advancements have been made in the application of effect-based methods and subsequent risk assessment in environmental quality monitoring.^{11,79,86,88} For sediments, particular attention has been given to the integration of toxicity identification evaluation (TIE) and effect-directed analysis (EDA).²⁸ The integration of TIE and EDA presents promising initial steps in toxicant identification in the quality assessment of contaminated sediments.^{29,142} Nonetheless, further developments of the methods that are currently used offer opportunities for improved understanding of sediment contamination and the accompanying risks to benthic biota.

The importance of bioavailability-based toxicity assessment was recently highlighted, as it allows a much more relevant representation of the exposure of benthic invertebrates to contaminants.^{142,143} Moreover, the concentrations obtained this way allow for subsequent comparison with environmental quality standards or effect concentrations for the water phase, which are much more readily available for water than for sediments.¹⁴⁴ A variety of methods for the determination of bioavailable contaminant concentrations in sediment has been described;^{143,145–148} however, the manipulation of sediments (e.g., sieving and homogenization) that is required for application in toxicity tests and especially for bioavailability-based extractions leads to altered sediment characteristics, affecting layering and pore water concentrations of contaminants,¹⁴⁹ which can lead to over- or underestimation of sediment toxicity.^{150–153} Hence, the use of undisturbed sediments in laboratory toxicity testing mimics the natural situation most closely, increasing the realism of the sediment quality assessment. Similarly, the use of chronic life cycle bioassays mimics the exposure of organisms on relevant time scales, representing ecologically relevant endpoints, and should allow for a more realistic interpretation of sediment toxicity to benthic biota.

The present study aimed to advance sediment quality assessment by combining invertebrate life cycle effect-monitoring and chemical profiling of sediment contamination to gain insight into the drivers of sediment toxicity to benthic biota. To this end, bioassays with intact whole sediment cores from contaminated sites were performed in tandem with explorative bioavailability-based chemical profiling of groups of legacy and emerging sediment contaminants, followed by the calculation of the potential toxicity of the detected contaminants. Based on the results, considerations and recommendations on the integration of effect-monitoring and chemical profiling in future sediment quality assessment are given.

MATERIALS AND METHODS

Outline of the Study

Sampling locations were selected in collaboration with the Dutch Water Authorities and subdivided into four categories according to the predominant surrounding land use or pollution source as follows: urban, waste water treatment plant (WWTP) effluent, agriculture, and reference. Urban sites were located in the city of Amsterdam and were identified as “chemical hotspots” requiring mitigation measures by the local water authority, based on high levels of polycyclic aromatic hydrocarbons (PAHs) and metals in the sediment (Table S1). WWTP sites received effluent from treatment plants from the cities Eindhoven, Hilversum, and Utrecht with a capacity of 120.000–750.000 inhabitant equivalents per day. Agricultural sites were located in areas with predominantly agricultural land use where a wide array of herbicides, insecticides, and fungicides is applied (Table S2). A site on the University of Amsterdam Science Park campus served as the reference location.

To assess the toxicity of the sediments, 28 day life cycle whole sediment bioassays with the nonbiting midge *Chironomus riparius* were performed on the intact whole sediment cores. Survival, emergence after 28 d, and mean emergence time (EmT_{50}) were monitored as endpoints.^{154,155} Standard 48 h *Daphnia magna* immobilization tests indicated no significant effects for simultaneous grab samples of the overlying water for all sampled locations (data not shown).

To elucidate the drivers of effects observed in the sediment bioassays, sediments were investigated for physical characteristics and chemical contamination [see Table S4 for compound properties and limits of quantification (LOQs)]. To this end, a selection of legacy and emerging sediment contaminants was made based on expected or indicative land use specific compounds.

Sediment pollution at the urban locations was expected to arise from metals and PAHs.¹⁵⁶ Therefore, Al, As, Ag, Cd, Cr, Cu, Fe, Mn, Ni, Pb, Se, and Zn were selected representing the metals, while phenanthrene and pyrene were selected as model PAHs because they are representative of the presence and toxicity of complex PAH mixtures and because their pore water dissolved concentrations can be accurately quantified by passive samplers (see below).¹⁵⁷

WWTP effluents typically contain large numbers of contaminants of emerging concern (CECs),¹⁵⁸ including pharmaceuticals, illicit drugs, and personal care products and their metabolites.⁸³ Five WWTP effluent marker compounds were selected, representing CECs

that accumulate in the sediment: the synthetic polycyclic musk fragrance HHCB (galaxolide, 1,3,4,6,7,8-hexahydro-4,6,6,7,8,8-hexamethylcyclopenta-*g*-2-benzopyran), the nonionic surfactant precursor and metabolite nonylphenol, the antimicrobial agent triclosan and its metabolite methyl-triclosan, and the plastic precursor bisphenol A (BPA). These compounds were selected based on their common occurrence in WWTP effluents, high production volumes, persistence, and their tendencies to sorb to the sediment.^{159–162}

Agricultural locations were expected to suffer most from pollution by pesticides used on the surrounding fields.⁷⁰ Therefore, the sediments were subjected to chemical screening for 150 commonly used pesticides (Table S4).

Metal concentrations were determined in total sediment extracts, after which freely dissolved pore water concentrations were calculated. Passive sampling with solid-phase microextraction (SPME) fibers was applied to determine pore water-equilibrated concentrations of the selected organic compounds. The obtained compound concentrations were thus representative of the bioavailable concentrations in the sediment^{143,146} and thereby directly relatable to the toxicity observed in the bioassays.¹⁶³

To identify the compound-based ecotoxicological risks of the contaminated sediments, the potential toxicity of the detected sediment-pore water contaminant concentrations was calculated for each location using three well-established toxicity indices:

1. The exposure-activity ratio (EAR), based on bioactivity data for a wide variety of contaminants generated by the U.S. Environmental Protection Agency (USEPA) ToxCast program.¹⁶⁴
2. The multisubstance potentially affected fraction (msPAF), based on species sensitivity distributions (SSDs) for multiple species and contaminant combinations.¹⁶⁵
3. The toxic unit (TU),¹⁶⁶ based on reported effect concentrations of the detected contaminants for a relevant species and endpoint.

Sampling Locations and Sample Collection

Sediment samples were collected at urban ($n = 5$), WWTP ($n = 3$), agriculture ($n = 3$), and reference ($n = 1$) locations in The Netherlands during March and April 2017 (Supporting Information 2). Each location was sampled on a single occasion, and intact whole sediment cores for the bioassays ($n = 5$ for contaminated sites, $n = 10$ for the reference site) and for the physical and chemical analyses ($n = 5$ for all sites) were collected using a sediment core sampler (UWITEC, Mondsee, Austria) loaded with acrylic tubes (l : 60 cm, d : 6 cm). Cores were collected over a 20 m transect with at least 0.5 m distance between each core, which, given the sediment homogeneity of the heavily modified water bodies that were selected for this study, resulted in a representative sampling of each location. In the laboratory, the top 5.5 cm was transferred to small acrylic tubes (l : 15 cm, d : 6 cm) using a sediment core cutter (UWITEC). The top 5.5 cm was used to ensure sediment stability and a water-sediment ratio that allows sufficient volume for quality measurements in the overlying water. To eliminate indigenous fauna, cores were stored at $-20\text{ }^{\circ}\text{C}$ for at least 48 h before the start of the experiments. Sediment

cores were selected at random for subsequent analyses. For chemical profiling, the top 2 cm of the sediments was analyzed, as it is the zone that is inhabited by chironomids¹⁶⁷ and thus representative of organism exposure to contaminants.

Physical Chemical Characterization of the Sediment

Particle size distribution, C/N ratio, and total organic carbon (TOC) content were determined for all sediments (see Supporting Information 4 for experimental details).

Whole Sediment Bioassays

Test Organism and Culturing Conditions

C. riparius larvae originated from the University of Amsterdam in-house laboratory culture, which was kept at 20 ± 1 °C and a 16:8 h light/dark photoperiod. The culture was maintained in aquaria containing quartz sand overlaid with Dutch standard water (DSW).¹⁶⁸ The culture was fed a mixture of Trouvit (Trouw, Fontaine-les-Vervins, France) and Tetraphyll (Tetrawerke, Melle, Germany) in a ratio 20:1. This mixture was also used as food for the whole sediment bioassays.

Sediment Preparation

Sediment cores were topped off with 125 mL of DSW and thawed for 24 h. In addition to testing a natural reference sediment (SP), a negative laboratory control was performed using artificial sediment according to OECD guideline 218¹⁶⁹ with slight modifications¹⁶⁸ containing 140 mg of food, representing 0.5 mg/larva/day food mixture for the entire duration of the experiment. The artificial sediment was sterilized by autoclaving and homogenized in glass bottles on a roller bank at 20 rpm for >24 h. Per acrylic tube, 260 g of artificial sediment was added, topped off with 125 mL of DSW, and left to settle for 24 h. All sediment cores were aerated throughout the experiment using glass Pasteur pipettes and compressed air. Aeration was turned on 24 h prior to the start of the experiment.

C. riparius 28 Day Life Cycle Whole Sediment Bioassays

The *C. riparius* 28 d life cycle whole sediment bioassays were performed with first instar larvae (<24 h) based on OECD guideline 218.¹⁶⁹ There were five replicates for each contaminated site and 10 replicates for the reference site and the negative control to ensure sufficient quality control of the experiment. One core from the reference site was lost during the transfer from the tube to the small acrylic core. Ten larvae per replicate sediment core were added at the start of the experiment. On day 7 and 14, 17.5 mg of additional food was added, corresponding with 0.25 mg food/larva/day for a period of 7 days. Demineralized water was added to compensate for water loss by evaporation. From day 14 onward, the sediment cores were covered with fine mesh gauze and checked daily for emerging midges, which were sexed and removed. At the end of the 28 d experiment, the sediments were sieved (350 µm) and the surviving larvae were counted. Dissolved oxygen concentration, conductivity, and pH were measured in the overlying water of

each core on day 1, 14, and 28 using a benchtop multimeter (HACH, Tiel, The Netherlands). The ammonium concentration was determined by analyzing 1 mL of filtered (0.2 μm pore size) overlying water of each core on an Autoanalyzer (San++, Skalar, Breda, The Netherlands). The number of surviving adults and larvae, the number of emerged adults, and the emergence time of the adults were monitored as endpoints.

Chemical Profiling

Metal Concentrations in Sediments and Pore Water

To determine the total metal concentrations in the sediments, ground and homogenized sediments were digested in a microwave with HNO_3/HCl .¹⁷⁰ From two replicate cores per location, 250 mg sample from the top 2 cm of the cores was used to extract metals (Supporting Information 3). Metal concentrations, as total concentration per sediment dry weight, were determined using an inductively coupled plasma mass spectrometer (Optima 8300; PerkinElmer). Freely dissolved metal concentrations were subsequently calculated using the SEDIAS tool,¹⁷¹ which considers the TOC content of the sediments to determine sediment-pore water partitioning coefficients for metals. This was possible for Cd, Cr, Cu, Ni, Pb, and Zn.

Passive Sampling with SPME Fibers

SPME fibers (Polymicro Technologies, Phoenix, AZ, USA) consisted of 200 m length glass fibers with an internal diameter of 108 μm and a 34.5 μm polyacrylate coating (coating volume 15.4 $\mu\text{L}/\text{m}$). SPME fibers were wrapped in aluminum foil and cut in 4 cm pieces, sequentially cleaned in three solvents (acetonitrile, methanol, and a 1:1 ultrapure water/methanol mixture; J.T. Baker, Deventer, The Netherlands) for 30 min each and stored in ultrapure water until application. From one frozen sediment core per site, the top 2 cm of each core was removed after 24 h of thawing and homogenized. Three replicate 10 mL vials per core were filled with ~5 g of wet sediment, three SPME fibers, and 5 mL of demineralized water. Three vials were filled with only ultrapure water and three SPME fibers as negative controls. The sediment slurries with SPME fibers were agitated for 28 d (pesticides and WWTP marker compounds) and 56 d (PAHs) on a roller mixer (20 rpm, Stuart SRT9; Cole-Parmer, Stone, UK) at 20 °C to ensure equilibrium partitioning with the sediment pore water.¹⁷² Next, fibers were cleaned using a wet tissue, cut into 1 cm pieces, and placed in 0.2 mL inserts in 1.5 mL high-performance liquid chromatography (HPLC) vials. Organic compounds were subsequently extracted from the fibers by the addition of solvents and agitation on a roller mixer for 1–3 h and stored at –20 °C until analysis. LC-grade acetonitrile (J.T. Baker) (200 μL) was used as the solvent for PAH, pesticide, and BPA analyses, and *n*-hexane (J.T. Baker) (150 μL) was used as the solvent for the other WWTP marker compounds. Chromatographic details for the analyses of organic contaminants are provided in Supporting Information 3 (Tables S5–S10).

Polycyclic Aromatic Hydrocarbons

Phenanthrene and pyrene detection in acetonitrile SPME extracts was performed on a LC system with a fluorescence detector (Shimadzu, Kyoto, Japan).

WWTP Markers

For detection of HHCB, nonylphenol (mixture of isomers), triclosan, and methyl-triclosan, the hexane SPME extracts were analyzed by gas chromatography coupled to mass spectrometry (GC-MS) using a Finnigan Trace MS quadrupole MS (Thermo Fisher Scientific) set to selected ion monitoring mode.

For detection of BPA, acetonitrile SPME extracts were analyzed by LC coupled to tandem MS (LC-MS/MS) using electrospray ionization operating in negative mode coupled to a QTRAP 4000 MS system (AB Sciex, MA, USA).

Pesticides

Acetonitrile SPME extracts were subjected to chemical screening for 150 commonly used pesticides at the laboratory of the water authority of Fryslân using LC-MS/MS and GC-MS, as previously described by de Baat *et al.* (2018).⁷²

Deriving Concentrations of Organic Contaminants in the Sediment Pore Water

Quantified concentrations in the (diluted) SPME extracts were converted to concentrations in the SPME polymer phase (0.616 μL per 4 cm fiber). The fiber-water partition coefficient (K_{fw}) was used to calculate the sediment-pore water concentrations of the target compounds. The $\log K_{fw}$ values for phenanthrene and pyrene were 4.29 and 4.99, respectively.¹⁷³ No K_{fw} values were available for the pesticides and the WWTP markers. Therefore, K_{fw} values for these compounds were estimated using their octanol-water partition coefficients (K_{ow}) and eq 1,¹⁷⁴ which describes the relationship between K_{ow} and K_{fw} for the polyacrylate fibers used in the present study.

$$\text{Eq. 1: } \log K_{fw} = (\log K_{ow} - 0.33) / 0.90$$

Finally, the freely dissolved pore water concentrations were calculated assuming chemical equilibrium between the sediment solids, the slurry water, and the SPME polymer (eq 2).

$$\text{Eq. 2: } C_{\text{aq,free}} = C_{\text{SPME,equilibrium}} / K_{fw}$$

where $C_{\text{aq,free}}$ is the pore water concentration and $C_{\text{SPME,equilibrium}}$ is the concentration in the fiber.

Data Analyses

A detailed description of the performed data analyses is provided in Supporting Information 5. In short, bioassay responses were compared between contaminated sites and the reference site (SP). Midge survival and emergence were tested for statistical differences using a Mann–Whitney U test. The mean emergence time (EmT_{50}), that is, the day at which 50% emergence occurred, was calculated separately for males and females according to a previously described protocol,¹⁷⁵ and significant differences were checked using a one-way ANOVA with Dunnett's multiple comparison post hoc test ($p < 0.05$).

Three previously described toxicity indices were calculated to determine the potential toxicity of the contaminant concentrations in the sediments. A cumulative EAR of the mixture of detected compounds (EAR_{mixture}) was calculated for each location by summing the EAR profiles of each of the compounds using the R package *toxEval*.^{11,176} For metals, no toxicity data were available within the USEPA ToxCast database¹⁷⁷ at the time of writing, and these were hence excluded from EAR_{mixture} calculations. Toxic pressures of individual chemicals, expressed as PAFs, were derived using previously reported chronic NOEC SSDs.¹⁷⁸ Subsequently, mixture toxic pressures, expressed as msPAF-NOEC, were derived assuming mixture toxicity according to the “mixed model” by De Zwart and Posthuma (2005).¹⁶⁵ Cumulative TUs were calculated per location assuming response additivity, in which TU was defined as the ratio of the measured concentration of a given compound to its EC_{50} for *D. magna*.¹⁶⁶ Previously reported threshold values that indicate risks of chemical contamination in surface waters were applied to interpret the calculated compound-based toxicity indices ($EAR_{\text{mixture}} = 1$; msPAF = 5%; TU = 0.1).^{11,179,180} Bioassay responses were summarized in a toxicity index in which each location was attributed a point for the occurrence of lethal and sublethal effects, respectively.

RESULTS

Physical Chemical Sediment Characteristics

Sediment characteristics varied among the different sediments (Table S12), but no land use related pattern in particle size distribution, TOC content, and C/N ratio became apparent (Supporting Information 4).

Sediment Toxicity

Quality Criteria

All quality parameters were in accordance with OECD guideline 218, except for the oxygen concentrations in two cores which were excluded from subsequent analyses (Table S15). The results of the 28 d life cycle whole sediment bioassays with *C. riparius* are depicted in Figure 5.1. Survival and emergence in the negative control were 93 and 92%, respectively. There was no significant difference in survival and emergence between the negative control and the field reference site (SP).

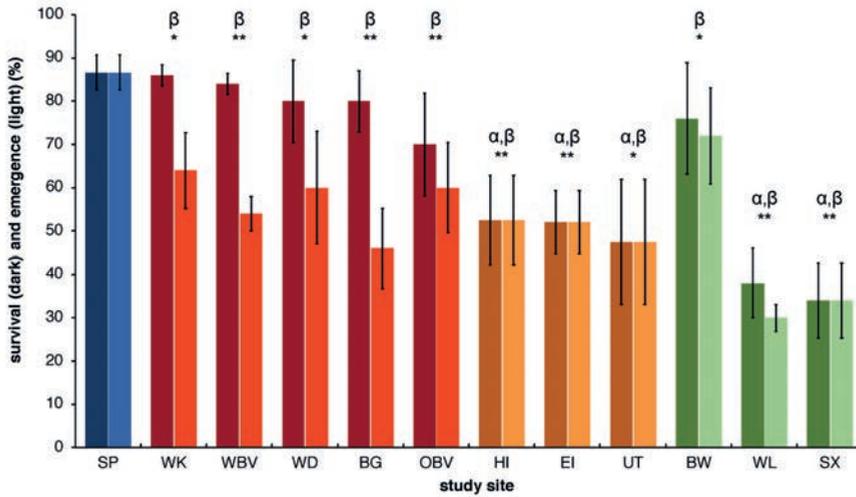


Figure 5.1. Survival (dark bars) and emergence (light bars) (mean \pm SE; % of initial individuals) of *C. riparius* after 28 d exposure to whole sediment cores from reference (blue), urban (red), WWTP (orange), and agricultural (green) sites. Significant differences between mean values of the reference (SP) vs contaminated sites are shown, where α indicates a difference in survival and β indicates a difference in emergence, and significance levels for α and β are indicated as * = $p < 0.05$ and ** = $p < 0.01$.

Survival

For all urban sites, survival was slightly but not significantly ($p > 0.05$) lower (70–86%) than that for the reference site (SP). Survival on the WWTP sediments (47.5–52.5%) was significantly ($p < 0.05$) lower than that on the reference sediment (SP). The agricultural sediment from BW did not significantly ($p > 0.05$) impact midge survival (76%). In contrast, the sediments from agricultural sites WL and SX impacted midge survival significantly ($p < 0.05$) and most strongly, with only 38 and 34% survival, respectively.

Emergence

In contrast to survival, adult emergence for urban sites (46–64%) was significantly ($p < 0.05$) lower than that for the reference site (SP), indicating a reduced larval development rate. Emergence for WWTP sites was significantly ($p < 0.05$) lower (48–53%) than that for the reference site (SP), yet this was attributable to the low survival. Emergence for the agricultural site BW (72%) was significantly ($p < 0.05$) lower than that for the reference site (SP) but higher than on all other contaminated sediments. Almost all surviving midges on the WL and SX agricultural sediments emerged (30 and 34%, respectively).

Emergence Time

As an example of cumulative emergence data, EmT_{50} curves for both genders on the reference sediment are shown in Figure S2. The EmT_{50} values on the artificial and the reference sediment (SP) were 17.2 and 19.1 days, respectively, for female midges and 16.4 and 17.9 days, respectively,

for male midges (Figure 5.2). For both genders, all EmT_{50} values differed significantly from the reference site (SP), except for males for WWTP location HI. The EmT_{50} values for the urban sites were higher than those for the reference site (SP), indicating delayed midge emergence on urban sediments. In contrast, EmT_{50} values on the WWTP sediments were lower than those on the reference sediment (SP), indicating accelerated emergence. The EmT_{50} values for the WWTP location HI were the lowest (17.2 days for females and 16.0 days for males) of all field sediments and were nearly identical to the EmT_{50} values of the artificial sediment. Agricultural sediments affected EmT_{50} values differentially. The sediment from BW and WL caused delayed emergence (22.0 and 20.0 days, respectively, for females and 19.0 and 19.8 days, respectively, for males), and the sediment from location SX caused accelerated emergence (18.3 days for females and 16.7 days for males).

Chemical Profiling

Quality Criteria

The LOQs (analytical and corresponding dissolved field concentration) of the compounds targeted in the chemical profiling are reported in Table S4. SPME measurement reliability and reproducibility were deemed sufficient for chemical profiling purposes (Supporting Information 3). Blank SPME signals of the target analytes were below LOQs, and the artificial sediment showed no signal except for low freely dissolved concentrations of the organic contaminants BPA and nonylphenol and the metals Cr, Cu, Pb, and Zn (Table S11).

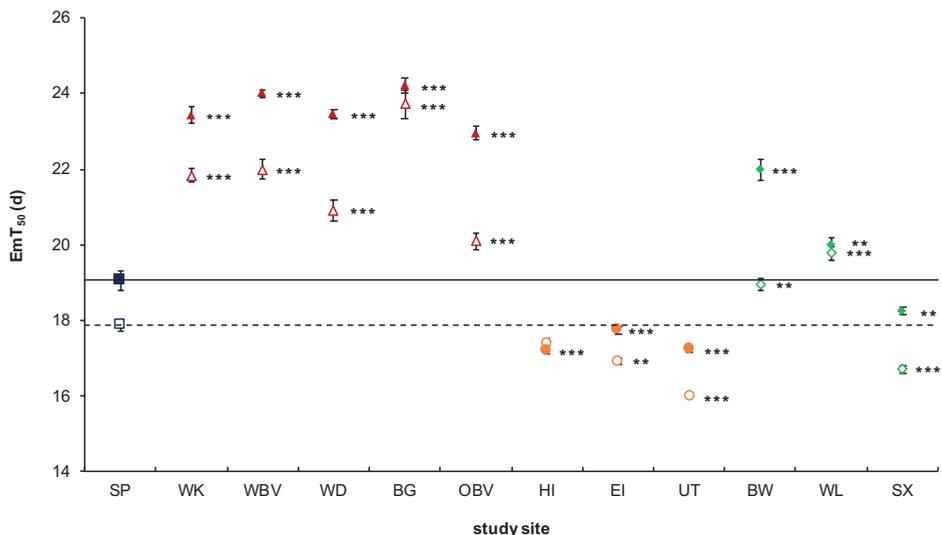


Figure 5.2. Mean (\pm SE) 50% emergence time in days (EmT_{50}) of *C. riparius* females (solid icons) and males (open icons) after 28 d exposure to whole sediment cores from reference (blue squares), urban (red pyramids), WWTP (orange circles), and agricultural (green diamonds) sites. The horizontal line represents the EmT_{50} value of the reference sediment for females (solid) and males (dashed), to which all locations were compared. Significance is indicated as ** $p < 0.01$ and *** $p < 0.001$.

Detected Contaminants

An overview of the detected freely dissolved contaminant concentrations is depicted as a heat map in Figure 5.3. As, Ag, and Se were not detected in any of the sediments. Freely dissolved sediment concentrations of Cd, Cr, Cu, Ni, Pb, and Zn differed between locations. Phenanthrene and pyrene, four of the WWTP markers, and 14 of the target pesticides were detected in the sediment from one or more locations. Metals were detected most frequently and at the highest concentrations for urban locations. PAHs were present in all land use types, but the highest pore water dissolved concentrations were detected for the urban locations WK (phenanthrene 22.2 µg/L, pyrene 2.4 µg/L) and BG (phenanthrene 42.4 µg/L, pyrene 6.3 µg/L). WWTP markers were detected for all sampling sites but more frequently and at higher freely

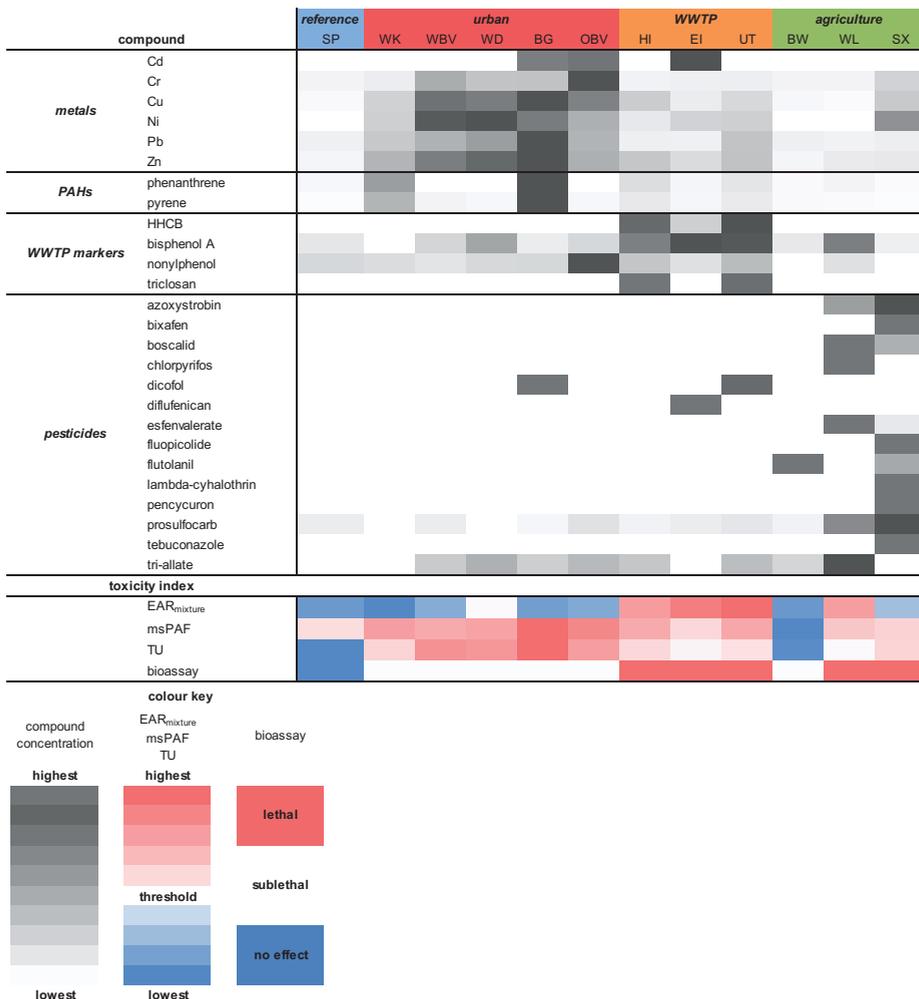


Figure 5.3. Heat map depicting freely dissolved contaminant concentrations and toxicity indices for sediments from sites with different land uses.

dissolved concentrations for WWTP sites. HHCB was detected for all WWTP locations (HI 40 ng/L; EI 10 ng/L; UT 50 ng/L). Triclosan was detected for two of the three WWTP locations (HI 2.9 µg/L; UT 2.4 µg/L). HHCB and triclosan were not detected in sediments from other land uses. BPA and nonylphenol were present at most of the investigated locations but at higher concentrations in WWTP sediments (HI 0.7 and 0.3 µg/L; EI 0.9 and 0.2 µg/L; and UT 0.9 and 0.4 µg/L, respectively), except for nonylphenol for the urban location OBV (1.1 µg/L). Traces of the herbicides prosulfocarb and triallate were detected in nearly all field sediments, but a higher diversity and higher freely dissolved concentrations of pesticides were detected for the agricultural sites. For WL, the fungicide boscalid (2.6 µg/L) and the pesticide esfenvalerate (2.9 ng/L) were detected. For SX, the fungicides azoxystrobin (4.8 µg/L), boscalid (1.1 µg/L), and fluopicolide (1.1 µg/L) and the insecticide λ-cyhalothrin (0.04 ng/L) were detected. For the third agricultural location BW, only the systemic fungicide flutolanil was found at a relatively high concentration of 0.3 µg/L.

Toxicity Indices

A wide range of values was calculated for the EAR (0.06–2.3), msPAF (0.7–74.5%), and TU (0.05–2.7) toxicity indices (Figure 5.3 and Table S11). The top contributing contaminants differed between the toxicity indices yet were very similar for the different locations within an index. The EAR was dominated by toxicity of BPA, msPAF by nonylphenol, and TU by Cu. A detailed overview per location and index is given in Table S14. The highest EAR_{mixture} value was found for the WWTP location UT (2.3) and the lowest was found for the urban location WK (0.06). The EAR_{mixture} threshold value (≥ 1) was met or exceeded by sediments from the urban location WD; WWTP locations HI, EI, and UT; and the agricultural location WL. The highest msPAF value was found for the urban location BG (74.5%) and the lowest was found for the agricultural location BW (0.7%). The msPAF threshold value (5%) was exceeded by all but the agricultural location BW. The highest TU value was found for the urban location BG (2.7) and the lowest was found for the reference location SP (0.05). For TU, the threshold value (0.1) was exceeded by all locations except the agricultural location BW and the reference location SP. The bioassay-based toxicity index, as it followed directly from the (sub)lethal bioassay responses, was the highest for all WWTP locations and the agricultural locations WL and SX and the lowest for the reference location SP. Thus, the toxicity indices produced a divergent outcome for potential toxicity based on the detected compound concentrations, which in turn was different from the outcome of the bioassays.

DISCUSSION

All contaminated sediments caused lethal and/or sublethal effects on the midge *C. riparius*, a species that is relatively resilient to sediment contamination.^{181,182} Contrastingly, acute bioassays with the sensitive invertebrate *D. magna* elucidated no toxicity in the overlying water at the same locations. This illustrates the severity of sediment contamination, present at a variety of locations with different pollution sources, on benthic invertebrates. No relationship

between TOC, C/N ratio, or particle size distribution and bioassay responses was observed, and the addition of food ensured that differences in the nutritional quality of the sediments did not affect bioassay responses. Hence, the clear differences in toxic effects could be attributed to the chemical profiles of the investigated sediments.

Intact Sediment Cores Can Simulate Realistic Contaminant Exposure

The need to improve the accuracy of ecological risk assessment, especially with regard to sediment manipulation before use in toxicity testing, was recently highlighted.¹⁵³ In the present study, the traditionally performed sediment manipulation was minimized because the use of intact whole sediment cores maintained natural layering and thereby contaminant availability in the sediments. As the upper centimeters of freshwater sediments are occupied by benthic organisms that live on top of and in the layered sediment,^{139,183} a realistic exposure of the test species to sediment contamination was achieved in the present study. The presently described methods thus introduce increased realism into toxicity assessment, while they do not present increased infrastructural demands compared to the use of manipulated sediments.

Life Cycle Bioassays Can Accommodate Sensitive Quality Assessment

The use of the life cycle bioassays allowed the measurement of lethal and sublethal endpoints in a chronic exposure scenario and differences in effects between the different land uses became more distinct with increasing endpoint sensitivity (survival < emergence < EmT_{50}). This illustrates the benefit of the inclusion of sensitive sublethal endpoints in effect-based sediment quality assessment, especially when the endpoints are indicative of stress responses that directly relate to the population level.¹⁸⁴ Responses to toxic compounds can, however, vary greatly between benthic species^{185,186} and the risk that contaminated sediments pose to benthic communities can be over- or underestimated if only one test organism is used. For example, chironomids are relatively resilient to chemical contamination,¹⁸⁷ and sediment bioassays with more sensitive species may elucidate toxicity at lower compound concentrations or for other toxic modes of action. Therefore, in line with Tuikka *et al.* (2011),¹⁸⁶ the use of additional test organisms is recommended. In an ideal situation, a suite of life cycle bioassays would be used as a powerful tool in the interpretation of a wide variety of ecologically relevant effects of sediment contamination. However, the high resolution of life cycle bioassays comes at a significant cost of time and labor intensity, making their regular implication in monitoring strategies less likely. Therefore, the development of simplified shorter bioassays could represent a valuable advancement in effect-based sediment quality assessment, provided that they will allow the determination of equally sensitive or more sensitive endpoints and maintain realistic exposure to the full pollution spectrum. Candidate endpoints include biomarkers for specific oxidative, neuronal, and energy metabolism stress that were shown to sensitively elucidate responses in *C. riparius* after 48 h at lower effect concentrations than responses in larval development and emergence after 28 d.¹⁸⁸ Additionally, molecular endpoints such as stress-related gene expression can be used to demonstrate responses of chironomids at the cellular level,¹⁸⁹ which can be observed on time

scales from hours to days. These cellular or molecular responses may provide test setups that can more readily be implemented in sediment quality assessment because of their lower infrastructural demands, but they are more difficult to extrapolate to population-level effects and still come at possibly high operational costs. Hence, given the more realistic interpretation of sediment toxicity to benthic biota that life cycle bioassays offer, their value in regular sediment quality assessment, especially compared to the traditional approaches based only on compound concentrations, should not be underestimated despite their high infrastructural demands.

Bioavailable Contaminant Concentrations Support Realistic Risk Interpretation

The use of total sediment concentrations can lead to misinterpretation of the contaminant exposure that aquatic invertebrates actually experience, as the organic carbon content of the sediment can influence the bioavailability of sediment-associated contaminants.^{142,143,147} Therefore, freely dissolved concentrations for metals were estimated based on the TOC content of the sediments. Although this approach presented an estimation of the bioavailable metal concentrations, ideally these should be determined directly by bioaccessibility-based extraction methods.¹⁴⁸ For metals, this can, for example, be achieved by means of diffusive gradient in thin films,¹⁹⁰ which is hence recommended for future sediment quality assessment. For organic compounds, passive sampling with SPME fibers was applied to measure freely dissolved concentrations of organic contaminants in the investigated sediments. The use of SPME in sediment chemical profiling had several advantages: (i) SPME material availability and cost, (ii) ease of use in terms of method simplicity and scale, (iii) the measurement consistency and reliability, and (iv) the availability of a validated method to calculate freely dissolved concentrations for a broad diversity of chemicals from concentrations in the SPME polymer phase.¹⁷⁴ The most prominent disadvantage of SPME in sediment passive sampling applications is the limited sorption capacity of the polymer phase. This limits the contaminant concentrations that can be obtained in SPME extracts, resulting in analytical detection limits that may exceed (sublethal) effect concentrations of highly toxic compounds (Table S4). Moreover, the small extract volumes that are obtained with SPME limit subsequent application in explorative analytical methods such as EDA,²³ or the recently proposed integration of TIE and EDA, that can greatly aid in diagnosing drivers of toxicity in sediments.¹⁴² Alternatively, polymeric materials with a higher sorption capacity for organic compounds, such as XAD resin¹⁴² or sheets made of silicone rubber, polyoxymethylene or polyethylene,¹⁴³ allow for large-volume bioaccessibility-based extraction of sediment-related contaminants. Additionally, because of the frequently acidic or basic nature of pharmaceuticals and pesticides, the application of ion-exchange polymers alongside polymers that target neutral organic compounds can improve the passive sampling of strongly sorbing but still predominantly charged compounds from sediments. An additional advantage of the use of polymers is their potential application in passive-dosing setups¹⁹¹ that can aid the integration of EDA and TIE for sediment quality assessment by allowing for high throughput determination of bioavailability-based *in vivo* and *in vitro* endpoints.²⁰

The determination of the bioavailable toxicant concentrations in the present study allowed a more accurate representation of the exposure of benthic invertebrates to organic contaminants.^{143,163} Because toxic effect concentrations are much more readily available for water than for sediments,¹⁴⁴ the concentrations obtained this way allowed for subsequent calculation of toxicity indices originally designed for the water phase.

Toxicity Indices Fail to Predict Sediment Toxicity

EAR, msPAF, and TU approaches were used to determine the potential toxicity of the detected contaminant concentrations in the sediments. Low toxicity index scores coincided with low bioassay responses for the reference location SP (except msPAF) and the agricultural location BW, but for all other locations, such convergent outcomes of the compound- and effect-based approaches were not observed. The relatively clean chemical profile and corresponding low bioassay response for the agriculture location BW can be explained by its position in front of a pumping station where the water from the entire agricultural area is collected, which leads to a more diluted pesticide loading to the sediment. The msPAF and TU indices responded most strongly for the urban locations, driven by the detected legacy contaminants, despite low bioassay responses. Contrastingly, the EAR responded most strongly for WWTP locations, strongly driven by WWTP markers, which is partly in line with the observed bioassay responses. It must be noted, however, that the lack of toxicity data for metals in the ToxCast database may have contributed to the relatively low EAR_{mixture} scores for the urban locations. Moreover, the TU calculations were based on toxicity data for *D. magna*, and species-specific sensitivities of *C. riparius* to the detected contaminants may have been over- or underestimated. This is also the case for the other toxicity indices, which are based on responses of a wide variety of either *in vitro* endpoints (EAR_{mixture}) or organisms (msPAF). As such, none of the used toxicity indices take species-specific sensitivity of *C. riparius* and the toxic mode of action of the detected contaminants into consideration, even though these very likely impacted bioassay responses.

Interestingly, all three contaminant concentration-based toxicity indices underestimated the toxicity for the agricultural locations WL and SX that showed the highest toxicity in the bioassays. Apparently, the bioassay responses were caused by contaminants that did not contribute strongly to the toxicity indices, or, more likely, were caused by unmeasured compounds. This illustrates that toxicity indices are strongly dependent on *a priori* selected compound lists, underlining the importance of careful selection of target compounds in chemical profiling.

Target Compound Lists Inevitably Lead to Misinterpretation of Ecotoxicological Risks

The present selection of target compounds was based on the expected pollutants at the sampling locations, originating from their main pollution sources. Land use-specific chemical profiles became apparent, with metals and PAHs predominantly present at urban locations, WWTP markers at WWTP locations, and pesticides at agricultural locations. In turn, land use-specific bioassay responses were observed, suggesting a correlation between the detected compounds

and the toxic effects. However, toxicity indices only partly explained the observed bioassay responses, suggesting that a broader selection of target compounds may have better explained the observed toxicity. This was previously shown to improve the explanatory power of toxicity indices,¹²⁰ and the selection of target compounds for future chemical profiling can be customized to suit any type of sediment, pollution source, or compound (group) of interest. However, toxicity indices will always depend on target compound lists and will consequently overlook the risks of unmeasured or unknown contaminants. Hence, the use of only compound-based toxicity indices can result in misinterpretation of risks in sediment quality assessment.

Sediment: An Environmental Compartment of Concern

In spite of more strict water quality regulations coming into place and generally decreasing dissolved contaminant concentrations, this study underlines the continued and increasing relevance of sediment contamination to aquatic ecosystem health. Midge survival was less impacted for the urban sites, which contained the highest legacy contaminant concentrations, than for WWTP-impacted and agricultural sites, which contained relatively low legacy contaminant concentrations. This illustrates that sediment-associated pesticides and emerging contaminants related to sewage effluent pose an even more severe risk to benthic invertebrates than legacy contaminants. Sediment is not only a reservoir for poorly degradable legacy contaminants but also a sink for other strongly binding, poorly degradable pesticides and emerging contaminants. As long as the prioritization of hazardously contaminated sediments remains based only on legacy contaminants, many sediments that pose an even greater environmental risk will not be identified. As sediments can act as a source of contamination to the relatively clean overlying water,¹⁴⁴ this underlines the importance of sediment as a vital environmental compartment in aquatic ecosystem health assessment.

ACKNOWLEDGEMENTS

The authors thank Frenk Selhorst, Rick Helmus, Samira Absalah, Rutger van Hall, Leo Posthuma, Brett Blackwell, Piet Verdonshot, Pim de Voogt, John Parsons, and three anonymous reviewers for their valuable contribution. This work is a part of the research program “Promotiebeurs voor leraren” with the project number 023.008.029, financed by the Dutch Research Council (NWO).

SUPPORTING INFORMATION

The supporting information to this chapter is available online at:
<https://pubs.acs.org/doi/10.1021/acs.est.9b02732>