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Dry Reforming of Methane under Mild Conditions Using Radio Frequency Plasma

Edwin Devid,* Diyu Zhang, Dongping Wang, Maria Ronda-Lloret, Qiang Huang, Gadi Rothenberg, N. Raveendran Shiju, and Aart W. Kleyn*

Dry reforming of methane (DRM) is a challenging process wherein methane reacts with CO\textsubscript{2} to give syngas. This reaction is strongly endothermic, typically requiring temperatures higher than 500 °C. Catalysts can be used, but the high temperatures (which are a thermodynamic requirement) often lead to catalyst deactivation. Herein, the reaction from another conceptual direction is approached, using low-power radio frequency inductively coupled plasma (RF-ICP). It is demonstrated that this system can give high conversions of methane and CO\textsubscript{2} at near-ambient temperatures. Importantly, the energy costs in this system are considerably lower compared with other plasma-driven DRM processes. Furthermore, it is shown that the yield of hydrogen can be increased by minimizing the C2 compound formation. The factors that govern the DRM process and discuss H\textsubscript{2} emission and its influence on H atom recycling in the process are examined.

1. Introduction

The effects of climate change caused by greenhouse gases are numerous and include deglaciation, marine heatwaves, and threats to global biodiversity.\cite{1–3} Carbon dioxide and methane are the most important anthropogenic greenhouse gases both by volume and by their contribution to the greenhouse effect. Ideally, society should stop emitting these gases completely, and hopefully this will be done in the future (see, e.g., the Sky scenario\cite{4}). Meanwhile, we should focus research on approaches for minimizing the impact of methane and CO\textsubscript{2} and removing them from the atmosphere. This means that we also need chemical reactions that consume methane, without generating CO\textsubscript{2}. One such possibility is the so-called dry reforming of methane (DRM: CH\textsubscript{4} + CO\textsubscript{2} → 2CO + 2H\textsubscript{2}). In this process, carbon dioxide and methane react to give syngas, a mixture of H\textsubscript{2} and CO, which can then be converted via Fischer–Tropsch synthesis into valuable hydrocarbon products.\cite{5,6}

DRM is a strongly endothermic reaction, with ΔH° = 2.56 eV or +247 kJ mol\textsuperscript{-1}. High temperatures (>500 °C) are needed to shift the equilibrium to the products side. However, such high temperatures mean very costly equipment and often cause catalyst deactivation via sintering and/or coking, e.g., see the studies by Ewbank et al. and Stroud et al.\cite{7,8} Running the DRM process at lower temperatures is therefore a worthy scientific and technological challenge.

Plasma-based technologies for DRM have been widely studied.\cite{9–11} DRM can be performed using plasma technology without the use of precious metal catalysts. Activation of carbon dioxide and methane molecules can be achieved in different ways through plasma: by thermal decomposition, using stepwise vibrational excitation, and electron collisions. For thermal plasma, thermal decomposition is the main process. In nonthermal plasma DRM, which is conducted at ambient temperatures, vibrational excitation is a more effective activation route.

The recent advances in DRM with plasma methods are summarized in several excellent reviews.\cite{9–14} These indicate that DRM can be done with many plasma methods, giving conversions of around 40% at an energy cost range between 1.2 and 200 eV per molecule. Energy efficiencies of DRM conducted through plasma technologies are difficult to determine, because during plasma-driven DRM, a variety of species and products are generated, including H\textsubscript{2}, CO, CH\textsubscript{3}, H\textsubscript{2}O, C\textsubscript{2}H\textsubscript{6}, C\textsubscript{2}H\textsubscript{4}, C\textsubscript{2}H\textsubscript{2}, C\textsubscript{3}H\textsubscript{8}, and C\textsubscript{4}H\textsubscript{10}. The wide product distribution affects the definition of energy efficiencies. Thus, plasma-based DRM processes are typically expressed in terms of “energy cost,” i.e., the amount of energy consumed by the DRM process in unit of eV per converted molecule.\cite{11} The energy cost (eV per molecule\textsubscript{conv.}) is therefore
They showed that the main reaction products are $\text{CO}_2$ and $\text{H}_2$ with 50% (at $2020\text{ Pa}$) yield increased steadily and the $\text{C}_2$ yield is unlikely. In our work, we will compare the DRM performance of the improved plasma-enhanced temperature DBD of Ray et al. with our plasma-only RF-ICP reactor.

Another way to produce valuable hydrocarbons from methane without syngas involves the partial oxidation of methane to methanol. Using plasma technology also, this route is available. Our work builds upon the work on methane activation by Mozetic et al. We utilize lower power, flow, and pressure and discover that $\text{C}_2$ compound formation decreases together with increased formation of $\text{H}_2$. This is a relevant finding for DRM. In addition, we show that DRM can be conducted in a RF-ICP setup with very high conversions and can compete on energy cost with other types of plasma-driven DRM processes. Finally, we discuss the observation of water formation and its influence on hydrogen atom recycling in the process.

2. Results
2.1. Quadrupole Mass Spectrometry Analysis of Activation of Pure $\text{CH}_4$

Quadrupole mass spectrometry (QMS) allows us to observe in real time both qualitatively and quantitatively the species formed in plasma at low pressures. Details of the analysis are reported in the Experimental Section.

Figure 1a shows the relative concentrations of product species from pure $\text{CH}_4$ reforming in RF-ICP and the carbon balance at different input powers. For decomposition of pure $\text{CH}_4$, the product’s H/C ratio should be 4 (irrespective of the RF power). In fact, this ratio increases from 4 at 0 W to 4.6 at 150 W. This indicates a loss of carbon by deposition of a carbonaceous film on the reactor wall. From the carbon balance plotted in Figure 1a (calculated as given in the caption and following the study by Ray et al.), it is shown that with increasing power the loss of carbon behaves linearly. With increasing specific energy input (SEI), more $\text{CH}_4$ dissociates into $\text{H}_2$ and $\text{C}$, where carbon gets deposited at the reactor walls.

A direct in situ spectroscopic analysis of this film is not possible. However, we see the formation of a yellow film inside the reactor tube after prolonged operation. In previous studies, we measured the electron temperature. In the present arrangement, this was not possible. However, based on our results, we estimate it to be between 2 and 5 eV. This means that direct electronic excitation of CO$_2$ is unlikely.

Our results show that RF-ICP can effectively activate methane molecules. The main products of $\text{CH}_4$ reforming were hydrogen and $\text{C}_2$ hydrocarbons. Increasing the power, the conversion of methane increased up to 60% (Figure 1a,b). The yields of $\text{H}_2$ and $\text{C}_2$ hydrocarbons sharply increased below 50 W. With a further increase in power, $\text{H}_2$ yield increased steadily and the $\text{C}_2$ yield stabilized. The highest concentration was for $\text{H}_2$ with 50% (at 150 W) of all gaseous products (Figure 1a), and the total $\text{H}_2$ yield over $\text{H}$-containing products was 35% at 150 W. The energy cost for $\text{CH}_4$ reforming follows here a linear trend as a function of the input power, which is lower compared with most plasma-based
Thus, RF-ICP is and CO

Figure 2b shows the relative signiﬁcantly alters the ratio of the products. Increasing the proportion of CO$_2$ from 0% to 70%, the C2 hydrocarbons decrease, whereas the H$_2$ concentration remains almost constant between 40% and 55% at the CO$_2$ fraction. In plasma, pure CO$_2$ is decomposed into CO and O$_2$. For pure CO$_2$, the O$_2$ relative concentration (17%) is almost 50% of that of the CO concentration (40%). With a small amount of CH$_4$ (0.8 CO$_2$/ (CH$_4$ + CO$_2$)) added, the O$_2$ decreased from 17% (at 100% CO$_2$ feed) to 7%. This is attributed to the rapid reaction of CH$_4$ with O to produce H$_2$O or OH, both of which are not detected by QMS. In addition, O$_2$ varies slightly between experiments with low CO$_2$/ (CH$_4$ + CO$_2$) fractions. Most likely, it depends on the amount of H$_2$O adsorption on the system walls.

The yields and selectivity of H$_2$ and CO are shown in Figure 2c. The H$_2$ yield with respect to the incident H atoms (in CH$_4$) increases with CO$_2$ content in the flow. The CO yield is basically constant between 40% and 55% at the CO$_2$/ (CH$_4$ + CO$_2$) feed at the same flow of 50 standard cubic centimeter per minute (herein: sccm) (data not shown). We observe that the conversion of CH$_4$ decreases with increasing pressure. This is a factor of 1.5 over the pressure range mentioned.
fractions 0.4–0.9. The reason for the appearance of a dip in the CO yield around 0.2 and 0.3 feed ratio is not known. The selectivity of H\textsubscript{2} increases from 72.3% to 83.5% when the fraction CO\textsubscript{2}/(CH\textsubscript{4} + CO\textsubscript{2}) reaches unity. The selectivity of CO peaks to 100% at the fraction 0.1. Beyond the fraction 0.1, the selectivity shows small fluctuations around 80%. Higher fractions CO\textsubscript{2}/(CH\textsubscript{4} + CO\textsubscript{2}) show that the presence of more CO\textsubscript{2} in the mix enables the increasing selectivity and yield of H\textsubscript{2}. The decrease in CH\textsubscript{4} in the mix shows that the decomposition of CH\textsubscript{4} affects little to none the selectivity and yield of CO.

Figure 3a shows the power dependence of the relative concentrations derived from QMS signals of the effluent gas from mixed CH\textsubscript{4}/CO\textsubscript{2} reforming in RF-ICP at CH\textsubscript{4}/CO\textsubscript{2} = 1. With increasing SEI, the decrease in carbon balance behaves nonlinearly. At higher input powers, a steady state is reached in carbon loss. The presence of a CO\textsubscript{2} feed at CH\textsubscript{4}/CO\textsubscript{2} = 1 enables to limit the carbon loss. Presumably the carbon from the dissociated CH\textsubscript{4} reacts with the atomic O atoms from the dissociated CO\textsubscript{2}. This observation corresponds with the steady-state concentration of O\textsubscript{2} in Figure 3a that explains why the O\textsubscript{2} concentration does not increase further with increasing power beyond 75 W.

RF-ICP grants here the combination of carbon and atomic oxygen into CO, thereby increasing further the yield of CO compared with the yield of C\textsubscript{2}H\textsubscript{X} molecules.

Measurement conditions are similar to those previously discussed in Figure 2. The H/C and O/C ratios were analyzed to test the consistency of the data. For a 1:1 mixture of
CH₄ and CO₂, about 20% C atoms is “missing” irrespective of power. The O/C ratio can deviate from the ideal value 1 by more than 20%. Formation of H₂O is likely under these conditions.

The yields of H₂ and CO are shown in Figure 3b. Both H₂ and CO yields gradually increase with increasing power. At 200 W, H₂ yield is 65% and CO yield is 74%.

The total conversion of mixed CH₄/CO₂ (1:1) reaches 77% (200 W) by plasma-only RF-ICP (see Figure 3c). This value is better than the absolute conversion of CH₄ and CO₂ of Ray et al., respectively, 68% and 65% obtained in their plasma-assisted thermal DBD reactor with the addition of their best-performing catalyst 15% Ni/Al₂O₃. Without the catalyst, Ray et al. obtain a maximum yield for both CO ad H₂ of 10%. In this case, power and flow are lower so a good comparison cannot be made. DRM driven by RF-ICP plasma only has the advantage that no catalysts are required to obtain high syngas yields, and so there is no need to find suitable catalysts resistant to coke formation.

Figure 3c shows that the energy cost of DRM at a CO₂/CH₄ ratio of 1:1 is here lower than the energy cost of reforming CH₄ only (Figure 1b). RF-ICP on CO₂/CH₄ mixed ratios provides energy costs comparable with DRM studies conducted by atmospheric pressure glow discharge (APGD) (Table 1).

Table 1 shows the total conversion and energy cost for multiple different plasma-driven DRM processes (originating from the study by Snoeckx and Bogaerts). Our RF-ICP-driven DRM process reaches a total conversion range of 0–77% (from 0 W to 200 W) and an energy cost range of 0–79 eV per molecule (from 0 W to 200 W). Comparing our performance ranges with the ones from Table 1, we observe that our RF-ICP-driven DRM outperforms microwave (MW) and DBD-operated DRM processes. The DRM processes operated by DBD reach...
Table 1. Total conversion versus energy cost of the DRM process by different plasma systems.[40,41]

<table>
<thead>
<tr>
<th></th>
<th>Gliding Arc</th>
<th>APGD</th>
<th>Spark</th>
<th>Nanosecond pulse discharge</th>
<th>Corona discharge</th>
<th>MW</th>
<th>DBD</th>
</tr>
</thead>
<tbody>
<tr>
<td>Total conversion [%]</td>
<td>3–44</td>
<td>36–88</td>
<td>7.5–96</td>
<td>0–64</td>
<td>14–90</td>
<td>50–73</td>
<td>0–85</td>
</tr>
<tr>
<td>Energy cost [eV per molecule]</td>
<td>1–50</td>
<td>0.9–60</td>
<td>3.2–10.6</td>
<td>3–100</td>
<td>4.2–30</td>
<td>120–400</td>
<td>12–600</td>
</tr>
</tbody>
</table>

close to the total conversion and energy cost of our RF-ICP-driven DRM performances. Furthermore, our RF-ICP-driven DRM performances reach closely the performances of APGD.[40,41]

Finally, RF-ICP-driven DRM reaches total conversions similar to corona and spark-driven DRM. But the energy cost with our RF-ICP is, respectively, about a factor 2 and factor 5 higher. However, we note that the efficiency of our RF-ICP system is not optimized and that better efficiencies should be obtainable using RF-ICP. By tuning the mixed ratios of the reactants, the energy cost of DRM by RF-ICP can be further changed and optimized.

2.3. Optical Emission Spectra Analysis of Activation of Mixtures of CH₄ and CO₂

We measured the optical emission spectra (OES) for CH₄/CO₂ mixtures as a function of composition (Figure 4) and as a function of input power for an equimolar mixture. In the analysis, we focus on the emission of the CO $b^3Σ^+ \rightarrow a^3Π$ state, OH, O atoms, and H atoms. To compare the emission intensity at different compositions of the mixture, the peaks of H₂ and the 0-0, 0-1 transition of CO $(b^3Σ^+ \rightarrow a^3Π)$ system are integrated (Figure 5).

Figure 4 and 6 show CO vibrational progressions that do not exhibit large vibrational excitations, such as the ones we previously observed for CO₂ plasma diluted by Ar.[20] The CO emission spectrum changes when tuning the CH₄/CO₂ ratios (Figure 4). The emission from CO and O atoms will dominate the OES in a pure CO₂ feed. As the proportion of CH₄ in the feed is increased from 0% to 10%, the emission peak of O atoms decreased drastically, and an OH emission band $(\lambda Σ^+ \rightarrow X^2Π)$ appears in the 300–320 nm region. In addition to the emission from OH, also, a large H₂α emission intensity was observed at 10% CH₄ in the mixed feed CH₄/CO₂ case rather than in pure CH₄ which has highest H/C ratio and H₂ yields.

The integrated intensities of two CO lines, two O atom lines, and H₂ emission are shown in Figure 5, as a function of the relative CO₂ content of the feed. The CO emission shows a linear increase with CO₂ content, very similar to the CO yield observed by QMS (Figure 2b). The linear increase seen in both QMS and OES of the CO signal confirms that the decomposition of the mass 28 signal to C₂H₄ and CO is done correctly. The QMS data show no leveling off of the CO signal, as shown for CO emission in Figure 5. We attribute this to change in plasma parameters at the highest CO₂ fraction, leading to a smaller electronic excitation of CO.

The O atom signal in Figure 5 shows a marked increase at the highest CO₂ content (>0.7). The H₂α emission shows a peak at the CO₂ content of 0.9, where the number of H atoms introduced by CH₄ is already very low.

The power dependence of the OES is shown in Figure 6a,b where the spectral regions are shown as in Figure 4. We see that the emission of H₂ and 0-0, 0-1 transition of CO $(b^3Σ^+ \rightarrow a^3Π)$ system increase as the power increases. For equimolar mixtures, no O atom emission is observed. In addition, the CO₂$^+$ doublet is observed, centered at $λ = 288.3$ nm and $λ = 289.6$ nm. OH emission is also observed (especially at a higher power) for the equimolar mixture. The line integrated intensities are shown in Figure 7. The trends observed are very similar to what is observed by the QMS intensities in Figure 3.

Figure 6b shows that the emission of H₂ during RF-ICP-driven DRM intensifies by increasing the power to 75 W. Above 75 W, the emission of H₂ levels off as a function of the input power. This indicates that RF-ICP-driven DRM enables easy excitation of hydrogen into H₂. RF-ICP offers an easy and accessible way to yield high levels of H₂.

![Figure 4. OES from CO₂ and CH₄ reforming at different CH₄/CO₂ ratios. Reaction conditions: feed: CH₄ and (or) CO₂; input power: 75 W; total flow: 50 sccm; and pressure: 50 Pa.](image-url)
3. Discussion

3.1. Activation of Methane

For a comparison to our DRM experiments, we conducted the RF-ICP-induced activation of pure methane. With increasing SEI, we observed an increasing dissociation. At the highest power (150 W, SEI = 45.71 eV per molecule), 60% of CH$_4$ was converted into H$_2$, C$_2$H$_x$, and C deposit. Conversion efficiencies of CH$_4$ are very significant at these power levels. There are very few similar studies using RF-ICP. The most comparable work is the study by Mozetic et al. [22] These authors also studied pure CH$_4$ discharges. However, their setup is different, as it contains a special after-glow chamber connected to the plasma chamber via a small orifice. Moreover, the RF power used in their study (1250 W) is six times higher than what we use here. Our system gives a higher conversion, which can be attributed to the lower operating pressure and possibly the absence of a buffer chamber. Comparing our results with those of Mozetic et al., we conclude that methane conversion of more than 50% can be obtained at lower power levels and pressures as well. This indicates that efficient DRM can be expected with RF-ICP, as we discuss in the following sections.

3.2. Activation of Mixtures of CH$_4$ and CO$_2$

The product concentration of a CH$_4$/CO$_2$ mixture is shown in Figure 2b as a function of the CO$_2$/CH$_4$ feed composition. As expected, the H$_2$ signal linearly decreases with the CO$_2$/CH$_4 + CO_2$ fraction, whereas the CO$_2$ relative concentration linearly increases. Also, the CO concentration linearly increases.

The CO yield is constant at 48% at 75 W. Figure 2c shows that the CO yield is nearly constant as a function of feed composition. This implies that the conversion of CO$_2$ into CO is independent of feed composition. Figure 3b shows that the CO yield increases with power, where even values of 75% CO yield can be reached at 200 W. The CO yield in DRM at 200 W is higher than what we measured for pure CO$_2$. The limited yield in pure CO$_2$ plasma is attributed to recombination of the reaction products CO + O back to the reactant CO$_2$, e.g., see the studies by Yin and coworkers [45–47]. RF-ICP-driven DRM enables the suppression of the reverse reaction. The product H$_2$ in Figure 2b is not proportional to the fraction of CH$_4$ in the feed. It is almost constant at a level 27–33% and decreases at CO$_2$/CH$_4 + CO_2$ fractions above 0.7. As a consequence, the H$_2$ yield (Figure 2c) increases as a function of CO$_2$/CH$_4 + CO_2$ fraction. Increasing power, as shown in Figure 3b, increased the H$_2$ yield to almost 75%. This shows that the formation of syngas has a high probability. Thus, RF-ICP-driven DRM is an efficient and clean way to produce syngas. The energy cost between CH$_4$ reforming (Figure 1b) versus mixed CO$_2$/CH$_4$ reforming (Figure 3c) shows significant differences, especially when compared with the total conversion. RF-ICP reforming of mixed CO$_2$/CH$_4$ ratios achieves lower energy costs while...
and CO yields at low powers, emission in the feed, where relatively few O, as the CO yields in DRM keep exceeding the yield is very high. In addition, the H driven disso- ratios demonstrates low energy.

Figure 7. Integrated emission intensity of 0-0, 0-1 transition of CO (b’Σg° – a’Πg) system and Hα as a function of input power. Reaction conditions: feed: 25 sccm CH4 and 25 sccm CO2 and pressure: 39 Pa.

significantly increasing both H2 and CO yields at low powers (Figure 3b). Optimizing the coupling of power into the plasma will be a next step in this work to further enhance the energy efficiency.

The CO optical emission intensity shows a similar trend with QMS regarding the mass 28 peak. This indicates that most of the product under CO2 feed is CO. In addition, the H2 yield is very high. This indicates that syngas can be produced by DRM through RF-ICP at a low specific energy.

3.3. Hα Emission

A most remarkable observation in OES is the strong Hα emission at high CO2/(CH4 + CO2) fractions in the feed, as shown in Figure 4 and 5. The highest optical emission intensity of Hα was achieved at 90% CO2 in the feed, where relatively few H atoms are introduced into the discharge. Nevertheless, the strong Hα emission indicates that a high number of H atoms is present in the plasma. We attribute this signal to the presence of H2O in the discharge, preferably for a high CO2 content. H2O cannot be easily detected in the discharge by QMS, but its presence is here deduced from the optical emission of OH (Figure 6). We believe that this is due to the electron impact–driven dissociation of H2O. The H and OH emissions are not equally strong due to different spectral properties. The process leading to strong Hα emission is similar to that observed before by Mucha et al. for diamond forming CH4 discharges.[48]

4. Conclusions

Our work demonstrates that RF-ICP discharges can form syngas by DRM. Syngas yields of more than 70% were observed above 150 W at an equal feed CH4/CO2 ratio. Moreover, RF-ICP-driven reforming of mixed CO2:CH4 ratios demonstrates low energy costs comparable with APGD-driven DRM and much lower than DBD-driven DRM. By controlling the mixing ratios of CO2:CH4, we can tune up the yield of H2 while minimizing the formation of C2 compounds. At high CO2/(CH4 + CO2) fractions in the feed, we observe significant amounts of Hα. The presence of Hα and OH in a RF-ICP-driven DRM process suppresses the recombination of CO + O, as the CO yields in DRM keep exceeding the CO yields found during the RF-ICP-driven discharge of pure CO2. The formation of water plays a key role in the hydrogen atom recycling in an RF-ICP-driven DRM process. All in all, we show that DRM using RF-ICP is a promising alternative for transforming the greenhouse gases methane and CO2 to valuable syngas under moderate conditions.

5. Experimental Section

All experiments were conducted in a designated RF-ICP reactor constructed in house (Figure 8). The plasma reactor consisted of a quartz tube, with a diameter of 40 mm and length of 600 mm. It was supported by two stainless steel flanges and sealed by O rings. The reactor tube was surrounded by a water-cooled copper coil. To establish efficient coupling of RF energy into the plasma, a matching box was placed between the RF power supply (13.56 MHz, 2 kW) and the copper coil. In contrast to our previous experiments, we could not mount a Langmuir probe to measure plasma parameters. A detailed technical description of the setup was published in previous studies.[20,49,50]

The maximum power used was 350 W, and the reflected power was maintained below 1 W by the matching box. However, this does not imply that all power from the supply was coupled into the plasma. Ohmic losses and RF losses by radiation will decrease the power actually coupled into the plasma. We had not optimized the coupling of power into plasma and only note that earlier work shows that very high coupling could be achieved.

The gases used in the reaction were directly obtained from gas cylinders and mixed before going into the reactor. Each gas cylinder was equipped with calibrated mass flow controllers (MFC, Sevenstar D07-19B) to set the flow. The plasma ignited inside the reactor tube, after supplying RF power. Prior to feeding the reaction gases, the reactor was evacuated to 1 Pa by a rotary pump with the nominal pumping speed around 18 L s−1.

The light emitted by the plasma was collected by an optical fiber located 1.5 cm downstream from the coil, viewing the center of the reactor tube. The data were transmitted to our ultraviolet (UV)–vis–near-infrared (NIR)
spectrometers (spectrometer, StellarNet LSR-NIR3b, LSR-UV2, LSR-VIS4b and LSR-VIS4a). Data were analyzed as described earlier. To study the effect of the specific RF-ICP power supply and pressure on DRM, the experiments were carried out in different modes. First, at various power levels (from 0 to 150 W), the CH₄ or CO₂ flow was fixed at 50 sccm. Second, the power was maintained constant at 75 W, and the pressure was changed from 50 to 300 Pa at a total flow of 50 sccm. Third, we studied the effect of the feed gas composition. For that we modified the fraction CO₂/(CH₄ + CO₂) while keeping the total feed gas rate and the power constant.

The composition of the gaseous products from the plasma reactor was determined by QMS. This was a powerful tool, because the gas composition could be measured in real time. In this way, the stability of the plasma can be checked continuously. Analyzing the mass spectra can be tricky, as different components led to ions with the same mass such as 28, and the sensitivity for each species could differ. Moreover, the transmission of the instrument may be mass dependent. To solve this, we used a simple approach to obtain the composition of the effluent gas. We determined the relative yields of equimolar mixtures of H₂, CH₄, CO, O₂ or CO₂, and Ar by QMS. The ratio of the yield of the parent molecular ions concerned and Ar⁺ was used to determine the relative efficiency for each gas. This analysis yielded the relative composition of the product gas as a function of one of the experimental variables.

To deconvolute the contributions to for instance the mass 28 peak, we conducted an interpolation between extreme cases, where the composition was known, such as experiments with pure CO₂ or CH₄. In this interpolation we took the height of secondary peaks into consideration. The interpolation was done by hand on the basis of individual mass spectra. Intensities of CₓHₓ species were hard to determine individually, also because of overlapping peaks in the mass spectrum. We obtained the C₂H₄ signal by adding the intensities of the dominant peaks. QMS had internal consistency checks, namely, the H/C and O/C ratio of the measured intensities. The ratio was set by the reactant flow and could be changed by plasma action. For pure CH₄: H/C = 4 and for pure CO₂: O/C = 2. We checked that the analysis was done in a consistent way. Previously, we reported the analysis data for the decomposition of pure CH₄ and pure CO₂ in this RF-ICP reactor setup. These studies form the basis for this work, which focused on pure methane and mixed CH₄/CO₂ reforming. Our calibration method was similar to the method described by Nguyen et al., where a fixed flow of argon flowed into the plasma reactor with a varying flow of the targeted gas species. From the ratios of the partial pressure of argon and the targeted gas species together with the flow ratios of each gas, a calibration factor could be determined. With the calibration factor (i.e., the slope of the line), the ratio of partial pressures could be plotted now with the ratio of the flow rates of the gases for H₂, CO, CH₄, CO₂, and O₂. This linear extrapolation together with the extrapolation from data between steady-state gases that were activated by plasma (input power > 0) versus not activated by plasma (input power was 0 W) allowed us to determine the contribution to the mass 16 and mass 28 signals. From limiting cases, the composition of both mass 16 and mass 28 was then known, and we could make the distinction between the ions O⁺ versus CH₄⁺ and C₂H₄⁺ versus CO₂⁺.

Further the fragmentation pattern and fragmentation ratio of gas species like CH₄ and CO₂ are considered by a preliminary background scan. By taking a QMS background scan of CO₂, we identified and checked the fragmentation pattern and ratio of CO₂. This enabled us to check and correct the mass spectra on the amount of CO that originated from the fragmentation of the parent molecule CO₂ in the ionizer of the QMS. Conducted an experiment was runs better than 20%. Within a single run, the signals for individual mass peaks reproduced to within 10% after stabilization of the system for a few minutes. The relative intensities of molecular CₓHₓ⁺, CH₄⁺, and C₂H₄⁺ could not be derived easily from mass spectrometry. As this decomposition was not relevant for this study, we added the various contributions and gave a total signal of CₓHₓ. The various molecules had different ionization cross sections, so we expected that there could be systematic errors up to 20% for the CₓHₓ signal.

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**Conflict of Interest**

The authors declare no conflict of interest.

**Keywords**

dry reforming of methane, optical emission spectroscopy, quadrupole mass spectrometry, radio frequency inductively coupled plasma, syngas

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