



## Supporting Information

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Investigating the Role of Surface Roughness and Defects on EC Breakdown, as a Precursor to SEI Formation in Hard Carbon Sodium-Ion Battery Anodes

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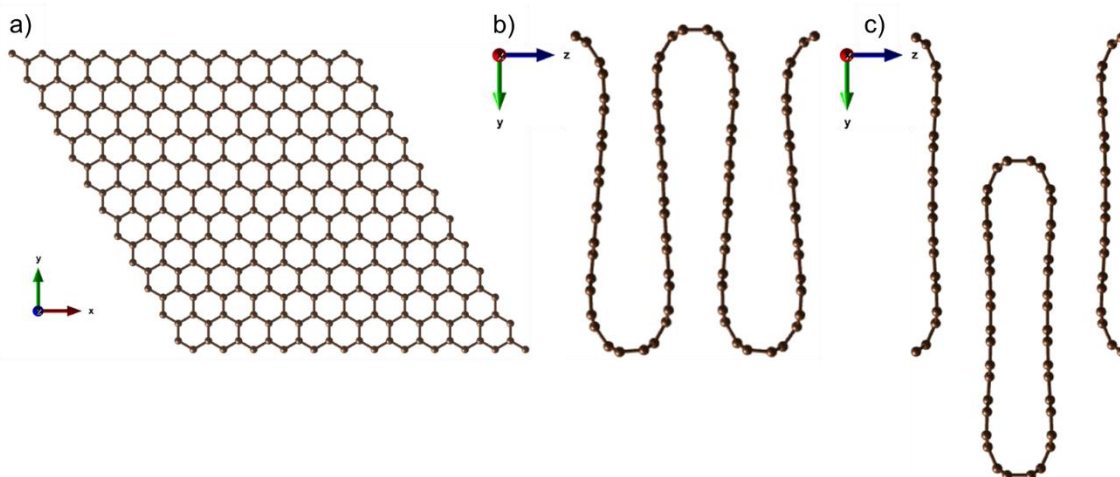


Figure S1. Simulation models used for a) basal plane, b) abrupt surface, and c) rough model. All models are 3D periodic, with a 25 Å vacuum gap in a) the z-direction (blue arrow axis), and b), c) in the y-direction (green arrow axis). Brown spheres are carbon atoms.

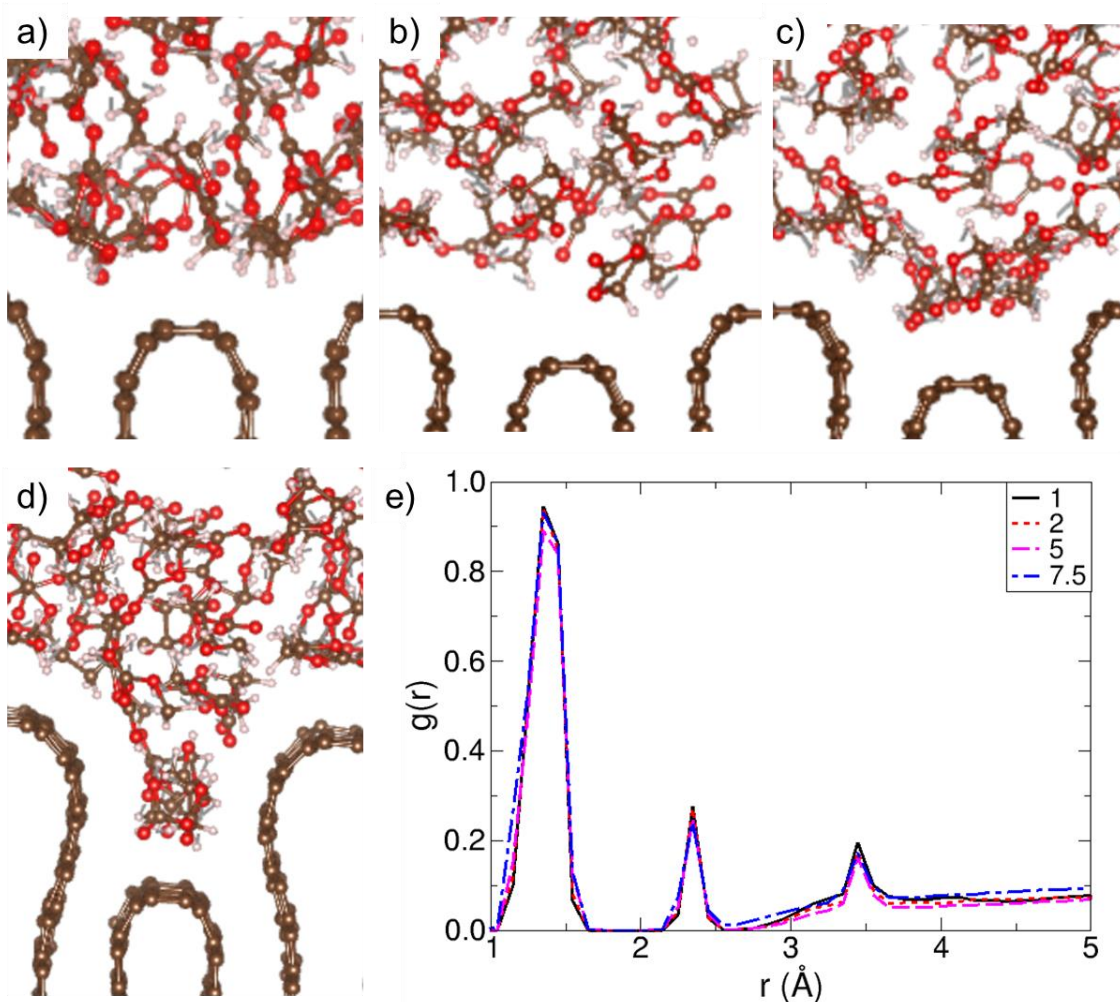


Figure S2. Snapshots of AIMD simulations for models with troughs of a) 1 Å, b) 2 Å, c) 5 Å, and d) 7.5 Å. e) shows the C-O pair correlation function ( $g(r)$ ) for the models in a-d. Brown spheres are carbon atoms, red oxygen, and white hydrogen.

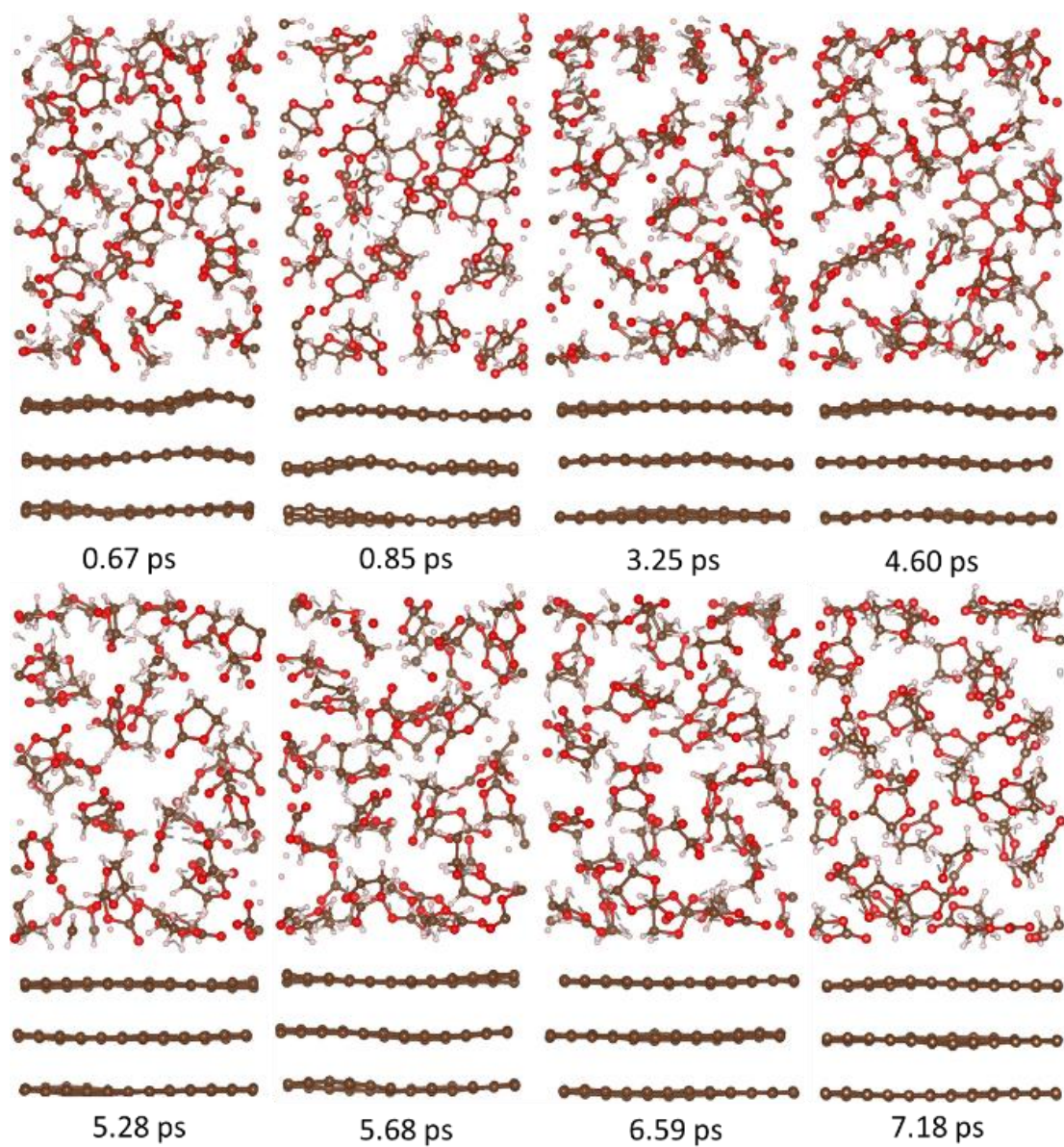


Figure S3. Snapshots of the EC basal plane AIMD simulations showing the rumpling of the graphene sheets. Brown spheres are carbon atoms, red oxygen, and white hydrogen.

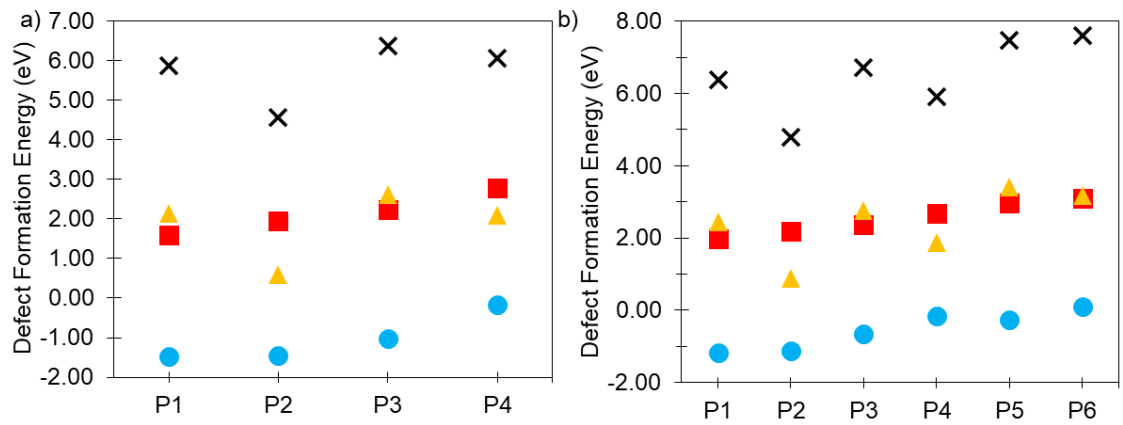


Figure S4. Defect formation energies for 2O<sub>C</sub> (blue circles), O<sub>C</sub> (red squares), O<sub>C</sub>V<sub>C</sub> (yellow triangles), and V<sub>C</sub> (black crosses) in a) 2 Å, and b) 5 Å rough models.

For the figures on the next few pages the figure captions are given on the next consecutive page.



Figure S5. Zoomed in view of optimised structure for the rough models with EC at defect sites (nomenclature for defect sites is given in Figure 1b in the main paper). Row one shows EC at site P1 with a)  $2O_C$  defect, b)  $O_C$  defect, c)  $O_CV_C$  defect, and d)  $V_C$  defect. Row two shows EC at site P2 with e)  $2O_C$  defect, f)  $O_C$  defect, g)  $O_CV_C$  defect, and h)  $V_C$  defect. Row three shows EC at site P3 with i)  $2O_C$  defect, j)  $O_C$  defect, k)  $O_CV_C$  defect, and l)  $V_C$  defect. Row four shows EC at site P4 with m)  $2O_C$  defect, n)  $O_C$  defect, o)  $O_CV_C$  defect, and p)  $V_C$  defect. Row five shows EC at site P5 with q)  $2O_C$  defect, r)  $O_C$  defect, s)  $O_CV_C$  defect, and t)  $V_C$  defect. Row six shows EC at site P6 with u)  $2O_C$  defect, v)  $O_C$  defect, w)  $O_CV_C$  defect, and x)  $V_C$  defect. Row seven shows EC at site P7 with y)  $2O_C$  defect, z)  $O_C$  defect,  $\text{\AA}$ )  $O_CV_C$  defect, and  $\text{\u00c6}$ )  $V_C$  defect. Brown spheres are carbon atoms, red oxygen, and white hydrogen.

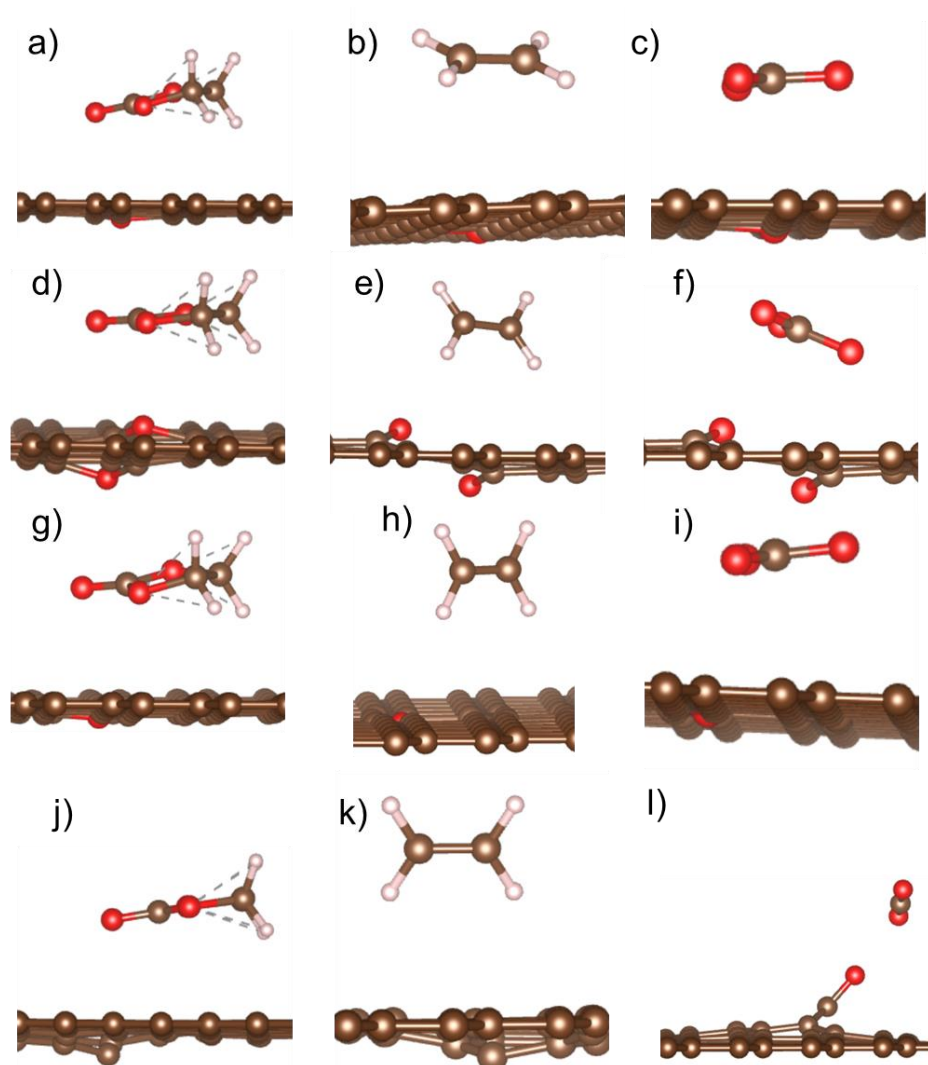


Figure S6. Zoomed in view of optimised structure for the basal plane models with a) EC, b)  $C_2H_4$ , and c)  $CO_3$  on  $O_C$  defect, d) EC, e)  $C_2H_4$ , and f)  $CO_3$  on  $2O_C$  defect, g) EC, h)  $C_2H_4$ , and i)  $CO_3$  on  $O_CV_C$  defect, j) EC, k)  $C_2H_4$ , and l)  $CO_3$  on  $V_C$  defect (with  $CO_3$  broken down forming a  $CO_2$  molecule and a  $=O$  surface functional group). Brown spheres are carbon atoms, red oxygen, and white hydrogen.

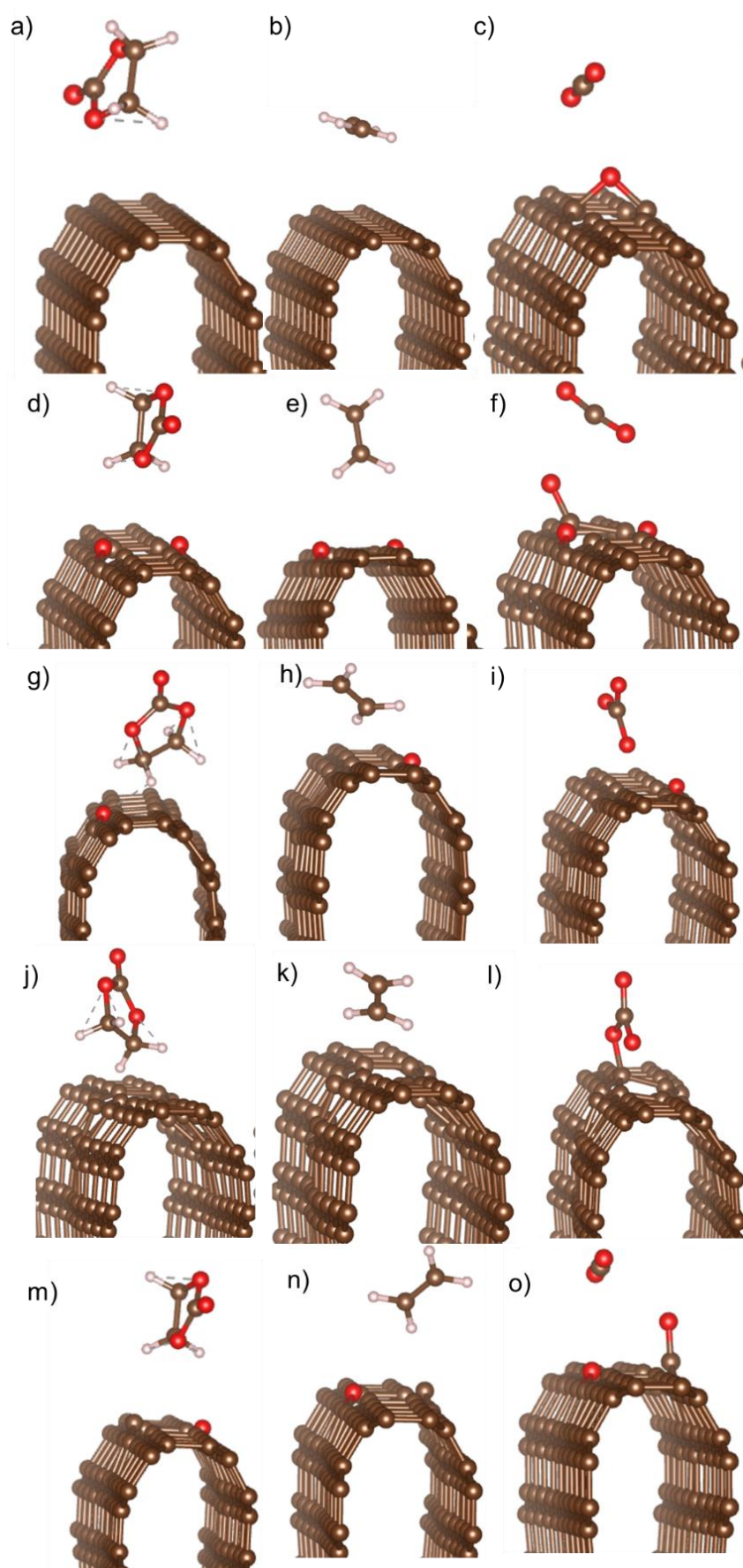




Figure S7. Zoomed in view of optimised structure for the abrupt models (only one carbon surface shown) with a) EC, b) C<sub>2</sub>H<sub>4</sub>, and c) CO<sub>3</sub> on pristine surface (with CO<sub>3</sub> broken down forming a CO<sub>2</sub> molecule and a -O surface functional group), d) EC, e) C<sub>2</sub>H<sub>4</sub>, and f) CO<sub>3</sub> on 2O<sub>C</sub> surface defect, g) EC, h) C<sub>2</sub>H<sub>4</sub>, and i) CO<sub>3</sub> on O<sub>C</sub>V<sub>C</sub> surface defect, j) EC, k) C<sub>2</sub>H<sub>4</sub>, and l) CO<sub>3</sub> on V<sub>C</sub> surface defect, and m) EC, n) C<sub>2</sub>H<sub>4</sub>, and o) CO<sub>3</sub> on O<sub>C</sub> surface defect (with CO<sub>3</sub> broken down forming a CO<sub>2</sub> molecule and a =O surface functional group). Brown spheres are carbon atoms, red oxygen, and white hydrogen.

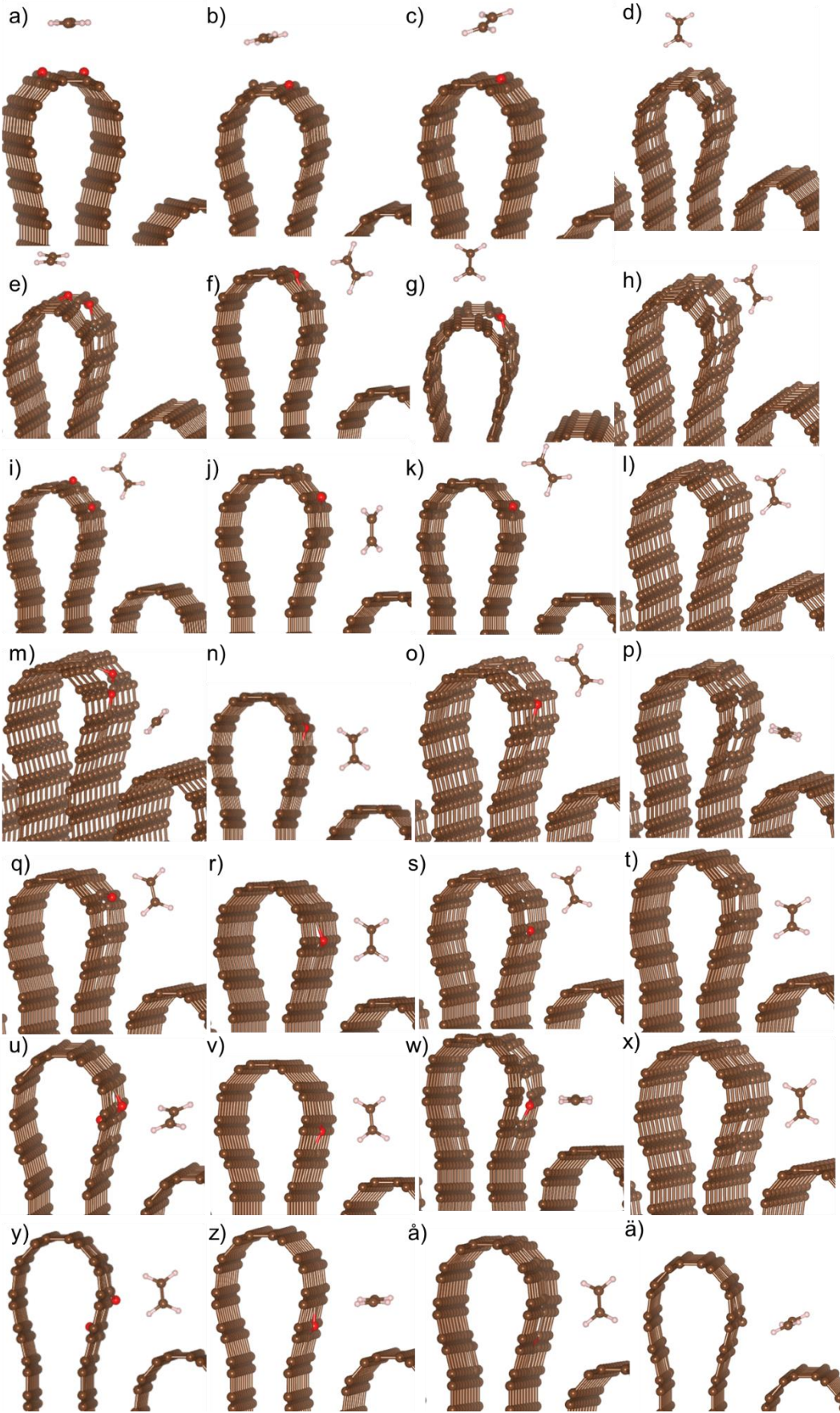


Figure S8. Zoomed in view of optimised structure for the rough models with  $C_2H_4$  at defect sites (nomenclature for defect sites is given in Figure 1b in the main paper). Row one shows  $C_2H_4$  at site P1 with a)  $2O_C$  defect, b)  $O_C$  defect, c)  $O_CV_C$  defect, and d)  $V_C$  defect. Row two shows  $C_2H_4$  at site P2 with e)  $2O_C$  defect, f)  $O_C$  defect, g)  $O_CV_C$  defect, and h)  $V_C$  defect. Row three shows  $C_2H_4$  at site P3 with i)  $2O_C$  defect, j)  $O_C$  defect, k)  $O_CV_C$  defect, and l)  $V_C$  defect. Row four shows  $C_2H_4$  at site P4 with m)  $2O_C$  defect, n)  $O_C$  defect, o)  $O_CV_C$  defect, and p)  $V_C$  defect. Row five shows  $C_2H_4$  at site P5 with q)  $2O_C$  defect, r)  $O_C$  defect, s)  $O_CV_C$  defect, and t)  $V_C$  defect. Row six shows  $C_2H_4$  at site P6 with u)  $2O_C$  defect, v)  $O_C$  defect, w)  $O_CV_C$  defect, and x)  $V_C$  defect. Row seven shows  $C_2H_4$  at site P7 with y)  $2O_C$  defect, z)  $O_C$  defect, å)  $O_CV_C$  defect, and ä)  $V_C$  defect. Brown spheres are carbon atoms, red oxygen, and white hydrogen.

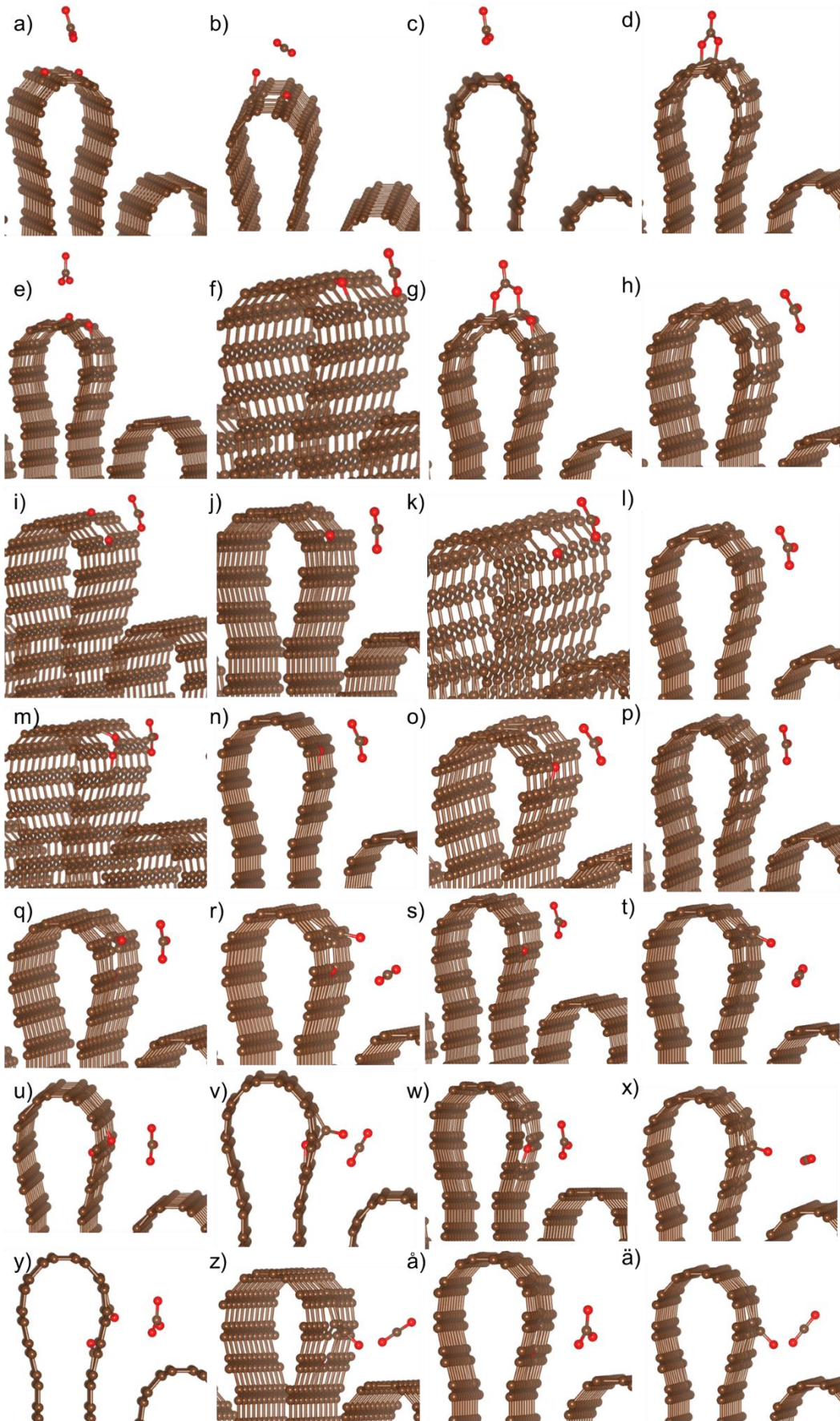


Figure S9. Zoomed in view of optimised structure for the rough models with CO<sub>3</sub> at defect sites (nomenclature for defect sites is given in Figure 1b in the main paper). Row one shows CO<sub>3</sub> at site P1 with a) 2O<sub>C</sub> defect, b) O<sub>C</sub> defect (with CO<sub>3</sub> broken down forming a CO<sub>2</sub> molecule and a =O surface functional group), c) O<sub>C</sub>V<sub>C</sub> defect, and d) V<sub>C</sub> defect. Row two shows CO<sub>3</sub> at site P2 with e) 2O<sub>C</sub> defect, f) O<sub>C</sub> defect, g) O<sub>C</sub>V<sub>C</sub> defect, and h) V<sub>C</sub> defect. Row three shows CO<sub>3</sub> at site P3 with i) 2O<sub>C</sub> defect, j) O<sub>C</sub> defect, k) O<sub>C</sub>V<sub>C</sub> defect, and l) V<sub>C</sub> defect. Row four shows CO<sub>3</sub> at site P4 with m) 2O<sub>C</sub> defect, n) O<sub>C</sub> defect, o) O<sub>C</sub>V<sub>C</sub> defect, and p) V<sub>C</sub> defect. Row five shows CO<sub>3</sub> at site P5 with q) 2O<sub>C</sub> defect, r) O<sub>C</sub> defect (with CO<sub>3</sub> broken down forming a CO<sub>2</sub> molecule and a =O surface functional group), s) O<sub>C</sub>V<sub>C</sub> defect, and t) V<sub>C</sub> defect (with CO<sub>3</sub> broken down forming a CO<sub>2</sub> molecule and a =O surface functional group). Row six shows CO<sub>3</sub> at site P6 with u) 2O<sub>C</sub> defect, v) O<sub>C</sub> defect (with CO<sub>3</sub> broken down forming a CO<sub>2</sub> molecule and a =O surface functional group), w) O<sub>C</sub>V<sub>C</sub> defect, and x) V<sub>C</sub> defect (with CO<sub>3</sub> broken down forming a CO<sub>2</sub> molecule and a =O surface functional group). Row seven shows CO<sub>3</sub> at site P7 with y) 2O<sub>C</sub> defect, z) O<sub>C</sub> defect (with CO<sub>3</sub> broken down forming a CO<sub>2</sub> molecule and a =O surface functional group), ã) O<sub>C</sub>V<sub>C</sub> defect, and ä) V<sub>C</sub> defect (with CO<sub>3</sub> broken down forming a CO<sub>2</sub> molecule and a =O surface functional group). Brown spheres are carbon atoms, red oxygen, and white hydrogen.

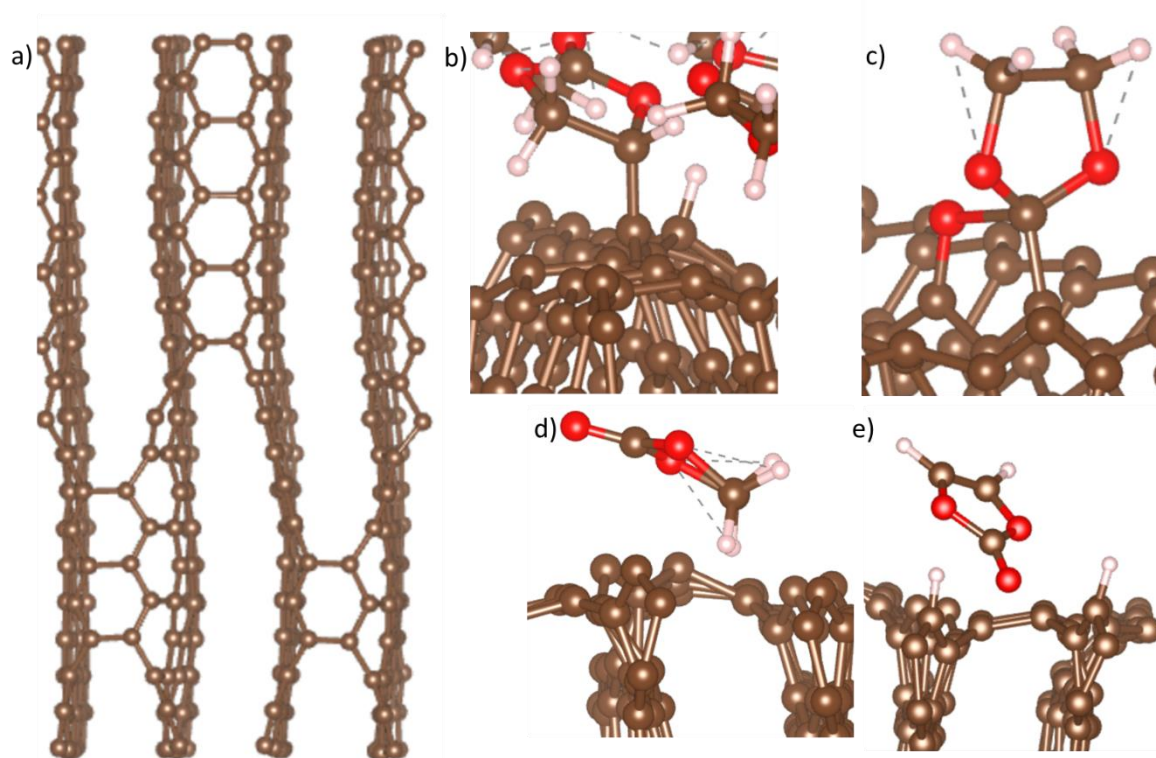


Figure S10. Zoomed in views of the features shown in Figure 7. a) Disordered carbon surface (top view) used for disordered carbon surface EC AIMD simulations. b) shows hydrogen abstraction from an EC molecule at 3.82 ps after equilibration, c) oxygen loss from EC to carbon surface at 6.84 ps after equilibration, d) EC interaction with surface at 25 ps after equilibration, and e) formation of vinylene carbonate through H loss to the disordered surface at 13.54 ps after equilibration.

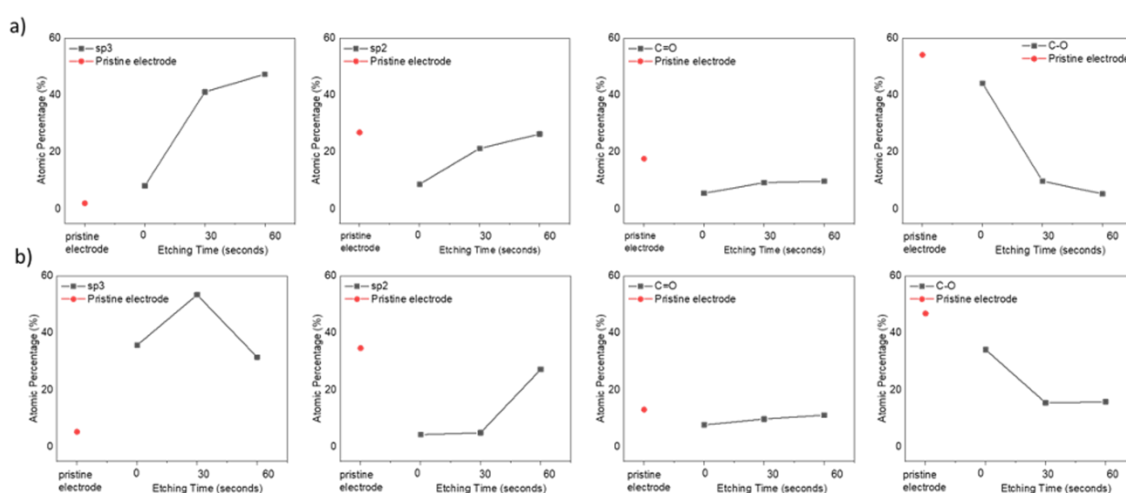


Figure S11. The atomic fractions calculated from the C1s spectra of a) ST1000 and b) ST1500 hard carbon anode cycled in 1M NaPF<sub>6</sub> in EC. ( x=0), after 30 seconds of etching (x=30), after 60 seconds of etching (x=60).

Table S1. The atomic fractions and error margins calculated from the C1s and O1s spectra for ST1000.

ST1000	C 1s					O1s		
	sp <sup>3</sup>	sp <sup>2</sup>	C-O	C=O	NaCO <sub>3</sub>	C-O	C=O	Na kll
Pristine electrode	2 (±0.1)	26.86 (±0.8)	54.15 (±1)	17.64 (±0.8)	-	61.29 (±1)	28.72 (±0.5)	9.99 (±0.3)
0 s	8.15 (±0.5)	8.73 (±1)	44.16 (±0.7)	5.54 (±0.3)	33.42 (±0.8)	26.04 (±0.5)	60.57 (±0.8)	13.39 (±0.5)
30 s	41.13 (±1)	21.26 (±0.8)	9.76 (±0.4)	9.24 (±0.5)	18.61 (±0.3)	14.76 (±0.5)	58.53 (±1)	26.61 (±0.8)
60 s	47.33 (±1)	26.3 (±0.8)	5.3 (±0.3)	9.73 (±0.6)	11.34 (±0.5)	12.25 (±0.3)	58.91 (±1)	28.83 (±0.8)

Table S2. The atomic fractions and error margins calculated from the C1s and O1s spectra for ST1500.

ST1500	C 1s					O1s		
	sp <sup>3</sup>	sp <sup>2</sup>	C-O	C=O	NaCO <sub>3</sub>	C-O	C=O	Na kll
Pristine electrode	5.35 (±0.1)	34.7 (±1)	46.83 (±1)	13.12 (±0.4)	-	60.41 (±1)	29.5 (±1)	10.09 (±0.7)
0 s	35.73 (±1)	4.41 (±0.9)	34.15 (±0.8)	7.81 (±0.3)	17.9 (±0.5)	26.9 (±0.4)	57.65 (±1)	15.45 (±0.7)
30 s	53.38 (±1)	5.04 (±0.1)	15.52 (±0.5)	9.82 (±0.5)	16.25 (±0.5)	16.96 (±0.5)	52.21 (±1)	29.83 (±0.9)
60 s	31.55 (±0.8)	27.23 (±0.5)	15.87 (±0.5)	11.16 (±0.5)	14.08 (±0.4)	15.8 (±0.5)	51.54 (±1)	32.66 (±0.7)