Influence of phosphorus on copper sensitivity of fluvial periphyton: the role of chemical, physiological and community-related factors
Serra, A.; Guasch, H.; Admiraal, W.; van der Geest, H.G.; van Beusekom, S.A.M.

Published in:
Ecotoxicology

DOI:
10.1007/s10646-009-0454-7

Citation for published version (APA):
Influence of phosphorus on copper sensitivity of fluvial periphyton: the role of chemical, physiological and community-related factors

Alexandra Serra · H. Guasch · W. Admiraal · H. G. Van der Geest · S. A. M. Van Beusekom

Accepted: 8 December 2009 / Published online: 19 December 2009
© The Author(s) 2009. This article is published with open access at Springerlink.com

Abstract The influence of eutrophication of fluvial ecosystems (caused by increased phosphorus concentrations) on periphyton Cu sensitivity is explored from a multi-scale perspective, going from the field to the laboratory. The study design included three tiers: a field study including the characterization of land use and the ecological state of the corresponding river sections in the Fluvia River watershed, an experimental investigation performed with natural periphyton from the previously studied stream sites in indoor channels, and finally a culture study in the laboratory. Results showed that differences in copper sensitivity of natural periphyton communities followed the gradient of nutrient concentration found in the field. Results from the culture experiments demonstrated that both, P-conditions during growth and P-content in the media are important factors modulating the toxicological response of algae to Cu. The observations from this study indicate that the ecological effects of metal pollution in rivers might be obscured by eutrophication.

Keywords Eutrophication · Cu toxicity · Periphyton · Phosphate · Rivers

Introduction

Among the many ecosystem stressors, eutrophication and metal pollution are the two major environmental problems in many developed and developing countries (Wang and Dei 2006). Agricultural activity is one of the major sources of phosphorus and nitrogen to aquatic ecosystems (Carpenter et al. 1998; Eckholm et al. 2000). At watershed scale, excessive inputs of phosphorus derived from agricultural practices are closely linked to eutrophication of surface waters (Johnson et al. 1997). In addition to fertilizers, human activities have also contributed to a progressive increase in other substances, especially heavy metals in aquatic environments (Nriagu 1979; Ma et al. 2003; Andrade et al. 2004) resulting in potentially toxic levels for aquatic organisms. As metal pollution is often associated with eutrophic conditions in aquatic ecosystems (López-Flores et al. 2003), it is considered of great interest to elucidate the interaction between nutrients and metal toxicity, and this investigation specifically focuses on the interaction between copper and phosphorus which are commonly found together in fluvial systems draining industrial and urban watersheds (Twiss and Nalewajko 1992). Metal bioavailability, and thus the toxic effects on the biota, can be modulated by the inorganic and organic chemistry of the water body (Genter 1996). Water pH, conductivity, temperature, nutrient availability and inorganic and organic ligands have been described as factors strongly influencing metal toxicity (Stumm and Morgan 1981; Luoma 1983; Meador 1991; Campbell 1995; Sunda and Huntsman 1998; Meylan et al. 2004).

Several investigations have focused on the interrelationships between trace metals and nutrients in phytoplankton and algal biofilms (Wang and Dei 2001; Interlandi 2002; Ivorra et al. 2002; Riedel and Sanders 2003; Guasch
Many studies support the hypothesis that metal toxicity is reduced in response to increases in P concentrations (Harding and Whitton 1977; Say and Whitton 1977; Chen 1994). On the other hand, several studies indicate that metals can induce nutrient limitation which can result in reduced algal growth (e.g. Nalewajko and Olaveson 1994; Paulsson et al. 2002). In these cases nutrient enrichment has been shown to compensate this effect of the metal as, for example, P-addition may allow the formation of cellular polyphosphate bodies which can bind intracellularly heavy metals in a detoxified form (Jensen et al. 1982; Peterson et al. 1984; Twiss and Nalewajko 1992). For instance, a direct influence of phosphorus (P) supply on copper toxicity to algae has also been demonstrated (Hall et al. 1989; Twiss and Nalewajko 1992; Hashemi et al. 1994; Nalewajko and Olaveson 1994; Guasch et al. 2004). Although much information about the interaction between nutrients and heavy metal toxicity is already available, especially in phytoplankton or single species studies, field testing and studies at community level remain scarce (e.g. Guasch et al. 2004).

Periphytic communities are ubiquitous and ecologically important components of many rivers and streams, as they are in the basis of the fluvial food web they have been used for assessing the quality of these ecosystems (Boston et al. 1991) including metal pollution (Johnson et al. 1978; Ramelow et al. 1992; Ivorra et al. 1999). Based on the opportunity to evaluate the response of numerous species simultaneously, community ecotoxicology can provide a much broader context for the assessment of environmental contamination than the study of individual species (Clements and Newman 2002).

In the present investigation, we hypothesize that eutrophication will lead to a reduction of copper sensitivity in natural periphyton communities due to variations in metal bioavailability as well as the phosphate regime during growth. Although both phosphorus and nitrogen supplies contribute to freshwater eutrophication (OECD (Organization for Economic Cooperation, Development) 1982), we have focused on the relationship between phosphorus (P) and copper (Cu) toxicity because phosphorus can often be the main limiting nutrient in freshwater environments (Horne and Goldman 1994). This is also the case of our study area. In addition, P–Cu interaction was also of interest due to the contribution of phosphorus in metal detoxification mechanisms (e.g. Twiss and Nalewajko 1992; Paulsson et al. 2002).

In order to test our hypothesis, the study design included three tiers: a field study including the characterization of land use and the ecological state of the corresponding river sections in the Fluvia River watershed, an experimental investigation performed with natural periphyton from the previously studied stream sites in indoor channels, and finally a toxicological study using algal cultures in the laboratory. Our multi-scale experimental design aimed to elucidate the relative role of the different processes contributing to the interaction effects of P and Cu on fluvial periphyton communities. Specifically the contribution of phosphorus conditions during growth and phosphorus/copper ratio in media on Cu toxicity was evaluated in algal cultures as well as in natural periphyton communities. Whereas the use of algal cultures allows control of biological and environmental conditions, tests using natural communities incorporate the ecological variability of the natural system (Cairns and Niederlehner 1987; Navarro et al. 2002; Sabater et al. 2007).

Materials and methods

Study design

This study was structured in three general sections: a field study, a microcosm study using experimental channels and natural periphyton communities and finally, a laboratory experiment with algal cultures. The field study was conducted in order to characterize the Fluvia watershed and was also used to choose the most suitable sites for the community ecotoxicology study. Natural periphyton communities collected from the chosen river sites were thereafter used to investigate the contribution of trophic conditions during growth on Cu toxicity. Finally, a benthic diatom was cultured under different P concentrations and it was also exposed to Cu in the short term in order to elucidate the effect of cell nutrient status on Cu toxicity. Moreover, the influence of P-supply in the media on Cu toxicity was also evaluated on both, periphyton and algal cultures (Fig. 1).

Field study

The Fluvia River, located in northeast Spain, is a calcareous Mediterranean river which drains a volcanic zone, declared a Natural Park in 1982. The sampling included six sites located in the headwater of the River Fluvia and its main tributaries corresponding to first- and second-order streams (Fig. 2).

The physicochemical parameters of each site were taken on five sampling dates from October 2002 to November 2003. On each sampling, water pH, temperature, conductivity, and dissolved oxygen were measured in the field. One litre of water samples per site were collected for their analysis in the laboratory. Samples were immediately filtered by GF/C Whatman glass microfibre filters, and refrigerated (4°C). They were analysed between 24 h for ammonium and alkalinity measurements and frozen for...
Soluble Reactive Phosphorus (SRP), nitrate and nitrite analyses following American Public Health Association (APHA) (1989).

Periphyton samples for algal biomass measurements, were obtained from 1 cm² of biofilm scraped off three stones from each sampling point in triplicate. Chlorophyll-a concentration was obtained after extraction in 90% acetone, and sonication for 5 min and determined spectrophotometrically, following Jeffrey and Humphrey (1975).

Land uses and land covers of each sub-catchment draining the six sampling sites were determined by using Geographic Information System (GIS). The entire watershed was subdivided into six sub-catchments corresponding to the six sites sampled. Land use and land cover maps were based on a 1:5000 orthophotomap provided by the Institut Cartogràfic de Catalunya (ICC), and digitized contours of sub-catchments from a 1:5000 topographic map using Arc-View GIS 3.2 software (California USA) in order to obtain the percentages of the different land uses.

Experimental channels

Once the gradient of eutrophication in the field was identified, three stream sites were selected for the toxicological study (B1, B2 and T1). Natural periphyton communities from these sites were used to assess Cu toxicity in indoor experimental channels. P-concentration in the media was experimentally manipulated in the Cu toxicity tests in order to evaluate the influence of both P-limitation and P–Cu chemical interaction on periphyton Cu toxicity. In these sites, metal pollution, which could interfere with results from toxicity tests, was not expected. In fact, results from a previous study carried out in the same watershed showed that the three sites selected for the present study (B1, B2 and T1) have dissolved Cu concentrations below 2 μg/l and the most polluted sites also had higher metal concentration (Guasch et al. 2009).

Periphyton colonization and sampling

Mature periphyton communities were obtained after allowing the colonization of artificial sand-blasted glass substrata (1.4 cm²) during 2 months (spring 2003) in each stream site. Colonized substrata were collected in the field and transported to the laboratory in cool boxes, filled with stream water (Guasch et al. 2003). Periphyton from three glass substrata from each sampling site were scraped and fixed with 4% of formaldehyde for taxonomical identification of the major algal groups under the light microscope. Three extra glass substrata from each sampling site were used for measuring algal biomass, estimated as the chlorophyll-a concentration (chl-a) as described above.

The physicochemical parameters of the stream water were measured in the field in each sampling site and date and 1 l of stream water was collected and transported to the laboratory for nutrient analysis as described above.

Cu toxicity in periphyton

For each sampling site, the colonized substrata were incubated in twelve Perspex (methyl methacrylate resin) channels using five replicates per channel (five artificial substrata). Ten litres of stream water (collected in the site of origin of each community) were recirculated in each
channel from containers by centrifuge pumps. All the containers were placed in a water bath to maintain the field temperature. The pH was maintained within the range measured in the field by regular addition of diluted sulphuric acid during the incubation. The incubation was conducted inside a greenhouse allowing the penetration of natural light.

Periphyton communities from each site were exposed to two treatments, six channels were exposed to a gradient of Cu concentrations (referred to as controls in the text) and six channels to the same gradient of Cu and with a supply of 20 μM phosphorus (referred to as +P in the text). Copper was added as copper chloride (copper titrisol, Merck, Darmstadt, Germany) to achieve nominal concentrations of 0, 0.31, 1, 3.16, 10, 31.6 μM. Phosphorus was added as KH₂PO₄ (Fig. 1).

Analyses of Cu concentration of the incubation water were performed periodically using ICP-MS (Agilent 7500c, Japan).

**PAM fluorometry measurements**

The effects of copper and phosphorus on periphyton were measured after 18 h of exposure using a mini-pulse modulation chlorophyll fluorometer (PAM) (Walz Mess, und Regeltechnik, Effeltrich, Germany) as described in Guasch et al. (2002). Before the PAM measurements, periphyton was previously dark adapted for 15 min. The effect of Cu on periphyton was evaluated directly on the colonized glass substrata by measuring the chlorophyll fluorescence yield of dark-adapted cells, also called chlorophyll fluorescence (Fo). This parameter was used as an indirect indicator of biomass (Serôdio et al. 1997; Rysgaard et al. 2001).

**Algal cultures**

Two independent sets of experiments were performed in order to obtain two replicates of the same experimental design (referred to as EXP 1 and EXP 2 in the text). Each set of experiments started after 15 days of culture growth and included six short-term toxicity tests of Cu toxicity on the diatom *Nitzschia perminuta*.

Before starting the toxicity tests, measurements of Alkaline Phosphatase Activity (APA) were performed in order to assess the degree of P limitation of the cultured diatom at P-repleted and depleted conditions. The procedure followed for the APA measurements is detailed below.

In each set of experiments, short-term toxicity tests were performed in duplicate using algal cultures differing in P regime during growth (P-depleted and P-repleted) and/or P in the media during the toxicity test (–P and +P) (Fig. 1).

**Culture preparation**

Non-axenic precultures of the diatom *Nitzschia perminuta* were kept in 1,000 ml Erlenmeyer flasks with 300–400 ml sterile WC medium (Guillard and Lorenzen 1972) modified as follows: the concentration of 1.0 mg/l H₃BO₃ was lowered to 0.006 mg/l, and molybdenum was added as sodium salt (the same molybdenum concentration as in the original WC medium). HEPES (2-[4-(2-Hydroxyethyl)-1-piperazinyl]-ethanesulphonic acid buffer was used to stabilise pH at 7.0. The medium was prepared following Van der Grinten et al. (2004). From the original pre-culture 1 ml was transferred to 250 ml Erlenmeyer flasks containing 5 ml of glass beads (Ø 490–700 μm) that were used as substrate for the algal development, filled with 100 ml of new sterile WC medium as described above and containing phosphate at different concentrations. The high phosphate regime used (100%) was that of the original WC medium, 50 μM representing a saturating concentration, corresponding to the P-repleted conditions and the low phosphate regime was 5 μM (10% of the original medium), corresponding to the P-depleted conditions.

The cultures were placed in an incubator, illuminated from above with fluorescent cool-white tubes at 100 μmol/m² s PAR (photosynthetically active radiation) and the temperature was kept at 20°C using a cooling bath. The flasks were closed with cellulose plugs to allow CO₂ exchange with the air. Microalgae were grown in these conditions for 15 days and growth was monitored by using PAM fluorometry measurements daily, using the chlorophyll fluorescence (Fo) as an indicator of algal biomass. Periodically and simultaneously to the PAM measurements, the number of algal cells was counted using a Burker-turk chamber, and these measurements were correlated with measurements of chlorophyll fluorescence (Fo) obtaining a good correlation ($r^2 = 0.83; p < 0.001; n = 80$) between these two variables.

**Cu toxicity to Nitzschia perminuta**

For each short-term toxicity test, 10–15 ml of algae were centrifuged at 13,000 rpm for 10 min and the pellet was resuspended in sterile media without EDTA, in order to avoid chelation with Cu, to provide cell densities of approximately $1 \times 10^5$ cells/ml. Thereafter, 3 ml of this algal suspension were exposed during 3 h to a growing gradient of Cu concentrations (0, 0.03, 0.1, 0.3, 1, 10, 30 μM Cu, nominal concentrations) added as copper chloride.
PAM fluorometry measurements

PAM fluorometry was used for monitoring algal growth, using the Fo as the end point, as well as for measuring the toxic effects of Cu on the photosynthetic activity of the algal cells by measuring the photochemical yield (Y) (Drábková et al. 2007). All the PAM measurements in the algal cultures were performed in vials of 3 ml of algal suspension. The cells were dark adapted for 15 min before measurements. The parameters measured were: the chlorophyll fluorescence yield of dark-adapted cells (Fo) and the maximal signal of dark-adapted cells obtained with a saturating radiation pulse (Fm). These parameters allow the calculation of the maximal yield of PSII as (Fm/Fo)/Fm, also called the chlorophyll fluorescence yield of dark-adapted cells (Fo) and the maximal signal of dark-adapted cells obtained with a saturating radiation pulse (Fm). These parameters allow the calculation of the maximal yield of PSII as (Fm — Fo)/Fm, also called photochemical yield (Y) which correspond to the capacity of dark-adapted cells to convert photon energy into chemical energy. Y is independent from the algal biomass. This nomenclature is according to Van Kooten and Snel (1990).

Alkaline phosphatase activity measurements

Alkaline phosphatase activity (APA) was determined spectrofluorometrically using 4-MUF-P (methylumbelliferyl phosphate) substrate from Sigma–Aldrich, following the methodology described in Romaní and Sabater (1999). From the P-repleted and P-depleted cultures of N. perminuta, 4 ml of algal suspension were incubated with MUF-P at final concentration of 0.3 mM (saturation conditions). Incubations were performed in the dark with continuous gentle shaking for 1 h at ambient stream temperature. Two blanks of sterilized medium were also incubated. After addition of 0.05 M glycine buffer pH 10.4 (1/1 v/v ratio buffer/sample), fluorescence was measured at 455 nm under 365 nm excitation (Kontron, SFM25).

Data treatment

The ordination of sampling points and sampling dates based on the water chemistry variables was carried out with Principal Components Analysis (PCA). The variables Ammonium, Phosphate and Nitrite were previously log-transformed to obtain the normality. Differences in phosphate concentration between sampling sites and sampling dates were evaluated by means of two-way ANOVA.

In the cultures experiments, Cu toxicity was estimated as Effective Concentration 50% (EC50) by fitting the photochemical yield (Y) to the four-parameter logistic curve model as described below.

\[
Y = \frac{(\text{max} - \text{min})}{1 + (x/\text{EC50})^{-\text{hillslope}}}
\]

where the variable x corresponds to the Cu concentration of the dose–response test and the y is the value of the measured end-point (i.e. yield or Fo). The parameter min equals the baseline, and max is the plateau of the curve. The parameter EC50 gives the transition center corresponding to the concentration causing 50% of inhibition of the measured end-point. The hillslope is the slope of the curve at the transition center.

In the channels experiments, Cu toxicity was estimated as Effective Concentration 50% (EC50) by fitting the chlorophyll fluorescence (Fo) to the four-parameter logistic curve model as described above.

Results

Field study

The Fluvia watershed is mainly covered by forested land (68% of the total surface), it is also influenced by agricultural activities (25% of the total surface) with some areas covered by shrub lands (4%), urban (2%), wetland vegetation (<1%) and denuded space (<1%). Land uses for each sub-catchment showed the following gradient: Bianya 1 has the highest percentage of forest and denuded space, Bianya 2, Turonell 1, Ridaura, Turonell 2 and Fluvia Olot have greater surface occupied by agricultural and urban space, and shrub lands (Fig. 3).

The PCA analysis of water chemistry shows the ordination of sampling points with respect to two principal axes (Table 1, Fig. 4). The first axis explained the 41% variance and separates two sites (Turonell 2 and Ridaura) with high phosphate concentration (higher than 20 μM), high conductivity, ammonium and nitrite concentrations, from the less polluted ones (Bianya 1, Bianya 2 and Turonell 1), where phosphate concentration does not exceed 2 μM. The second axis explained the 22% variance, placing the Fluvia Olot site on the positive side, with high pH, nitrate and dissolved oxygen concentrations. Algal biomass (estimated as chlorophyll-a concentration) of the periphyton colonizing the stream bed was lower in sites with low nutrient
concentration and higher in those with intermediate and high nutrient concentration (Table 1).

The sites Bianya 1 (B1), Bianya 2 (B2) and Turonell 1 (T1) were selected for the toxicity study in the channels. These sites have been chosen because they have similar physical and chemical characteristics but they differ in their background nutrient concentrations (following a gradient in eutrophy derived from farming and agricultural activities developed in their sub-catchments) and this was persistent over the study period (Table 1). The selected sites showed significant differences in their phosphate background concentrations ($\text{F}_{2, 8} = 25.74; p < 0.001$) but no significant differences were found between the 5 sampling dates ($\text{F}_{4, 8} = 1.24; p = 0.368$). In addition, the three

### Table 1: Physical, chemical and biological parameters measured at the studied sites in five dates from October 2002 to November 2003, and percentage of the land uses and land covers of each sub-catchment

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Bianya 1</th>
<th>Bianya 2</th>
<th>Turonell 1</th>
<th>Turonell 2</th>
<th>Ridaura</th>
<th>Fluvia O.</th>
</tr>
</thead>
<tbody>
<tr>
<td>pH</td>
<td>8.07/8.16* (0.165)</td>
<td>7.99/7.96* (0.144)</td>
<td>8.18/8.14* (0.079)</td>
<td>8.19 (0.210)</td>
<td>8.09 (0.165)</td>
<td>8.35 (0.072)</td>
</tr>
<tr>
<td>Temp (°C)</td>
<td>9.15 (3.61)</td>
<td>10.7 (2.51)</td>
<td>10.0 (3.46)</td>
<td>11.3 (3.09)</td>
<td>14.0 (2.65)</td>
<td>10.8 (3.97)</td>
</tr>
<tr>
<td>Cond (µS/cm)</td>
<td>535/520* (98.4)</td>
<td>630/588* (118)</td>
<td>603/590* (103)</td>
<td>1914 (708)</td>
<td>1590 (307)</td>
<td>639 (106)</td>
</tr>
<tr>
<td>Alk (meq/l)</td>
<td>6.6/4.56* (1.26)</td>
<td>7.48/5.04* (1.49)</td>
<td>7.81/5.64* (1.25)</td>
<td>7.84 (1.66)</td>
<td>7.59 (1.46)</td>
<td>6.50 (1.59)</td>
</tr>
<tr>
<td>Oxy (% sat)</td>
<td>93.5 (4.88)</td>
<td>93.0 (4.93)</td>
<td>95.3 (8.86)</td>
<td>93.3 (7.13)</td>
<td>80.2 (12.8)</td>
<td>113 (22.4)</td>
</tr>
<tr>
<td>NO₃ (µM)</td>
<td>31.9/34.3* (18.6)</td>
<td>115/109* (68.4)</td>
<td>155/172* (89.9)</td>
<td>174 (74.2)</td>
<td>113 (75.1)</td>
<td>261 (151)</td>
</tr>
<tr>
<td>NO₂ (µM)</td>
<td>0.016/0.06* (0.019)</td>
<td>0.305/0.32* (0.107)</td>
<td>0.128/0.22* (0.068)</td>
<td>16.3 (8.98)</td>
<td>8.38 (8.46)</td>
<td>1.36 (0.490)</td>
</tr>
<tr>
<td>NH₄ (µM)</td>
<td>1.71/0.46* (1.90)</td>
<td>2.03/1.76* (2.87)</td>
<td>1.01/1.18* (0.688)</td>
<td>67.7 (77.3)</td>
<td>72.28 (88.3)</td>
<td>3.27 (2.61)</td>
</tr>
<tr>
<td>PO₄ (µM)</td>
<td>0.085/0.095* (0.059)</td>
<td>0.155/0.195* (0.053)</td>
<td>1.05/1.228* (0.349)</td>
<td>87.4 (52.8)</td>
<td>23.2 (10.4)</td>
<td>1.02 (0.529)</td>
</tr>
<tr>
<td>N/P</td>
<td>424 (171)</td>
<td>1089 (511)</td>
<td>224 (115)</td>
<td>9.30 (15.5)</td>
<td>11.4 (6.45)</td>
<td>636 (764)</td>
</tr>
<tr>
<td>Chl-a (µg/cm²)</td>
<td>7.06/20.41* (4.65)</td>
<td>22.6/26.53* (5.11)</td>
<td>21.9/33.74* (5.48)</td>
<td>6.70 (2.89)</td>
<td>26.2 (9.95)</td>
<td>10.2 (1.49)</td>
</tr>
</tbody>
</table>

The values are the average and the standard error (in parenthesis) for each site of water temperature (temp), conductivity (cond), alkalinity (alk), dissolved oxygen (oxy), chlorophyll-a content (chl-a)

* Corresponds to the physicochemical parameters of stream water and algal biomass used in the channel experiments (June 2003)

---

**Fig. 4** Principal Components Analysis based on the physicochemical parameters of stream water measured in each sampling date summarized in Table 1. Factor loadings correlated with a significance $p < 0.05$ for each factor are indicated under an arrow for clarity. Significance of each axis is $p < 0.05$
selected sites are the least polluted sites among the six sampled and their sub-catchments contain the highest percentage of forested surface (>75% of the total watershed surface).

Experimental channels

The physicochemical parameters of the stream water, and algal biomass of the periphyton used in the channel experiments, are summarized in Table 1.

Periphyton communities were dominated by green algae in B1 and B2 (76 and 78% respectively), and were also abundant (56%) in T1. B2 and T1 communities had a high percentage of diatoms (19 and 44% respectively) compared to B1 (1.8%). Cyanobacteria were more abundant in B1 (22%) than in B2 (2.5%) and no cyanobacteria were found in T1. In addition, B1 presented filaments of *Rivularia* spp., a colony-forming cyanobacterium which is characteristic of phosphorus-limited ecosystems (Guasch and Sabater 1995).

The three studied communities showed differences in their sensitivity to Cu. In the control treatments (without P-supply), differences in Cu tolerance followed the gradient of nutrient concentrations found in the field (Fig. 5, Table 2). B1 was the most sensitive community, followed by B2 which (1.7 times more tolerant than B1) and T1 (3.6 times more tolerant than B1). Comparing the ranges of the EC50 found between treatments (with and without P-supply), it was observed that Cu sensitivity was lower with P-supply in B1 and B2 communities while no differences in the Cu sensitivity were observed in the T1 community (Table 2). The increase of EC50 values caused by the P-supply was 3.4 and 3.6 fold in B1 and B2 communities, respectively.

Algal cultures

Phosphate concentration of the culture media was $4.75 \pm 0.13 \mu M$ SRP and $46.33 \pm 1.54 \mu M$ in P-depleted and P-repleted treatments SRP respectively. Clear differences in Alkaline Phosphatase Activity (APA) were found between P-depleted and P-repleted cultures in both sets of experiments (EXP 1 and EXP 2) as a result of differences in P availability in the media during growth.

APA activities measured in P-repleted cultures were $1.67 \times 10^{-7} \pm 5.90 \times 10^{-9}$ and $7.92 \times 10^{-7} \pm 1.21 \times 10^{-9} \mu M$ MUF/cell h (in EXP 1 and EXP 2, respectively). APA in the P-depleted cultures were $4.13 \times 10^{-6} \pm 4.70 \times 10^{-8}$ and $2.70 \times 10^{-6} \pm 5.04 \times 10^{-8} \mu M$ MUF/cell h (in EXP 1 and EXP 2, respectively); corresponding to 25 and 3.4 times higher than in P-repleted cultures.

EC50 values for Cu inhibited photosynthesis were highly similar comparing both sets of experiments (Fig. 6). The results showed differences in the toxicological responses of
In this investigation we explored the influence of eutrophication, and more specifically P enrichment, on Cu toxicity to periphyton from a multi-scale perspective which allowed us to integrate different degrees of complexity. Observations from natural communities exposed to different nutrient conditions in the experimental channels supported our hypothesis that eutrophication may lead to a reduction of copper sensitivity in natural periphyton. In addition, observations from algal cultures allowed us to clarify the relative contribution of phosphorus during algal growth and phosphorus in the media to copper toxicity.

Results from culture experiments, which allowed us to isolate the effect of phosphorus on Cu toxicity, confirmed that both the presence of phosphorus in the media and during algal growth leads to an increase in Cu tolerance of algae. Differences in Cu sensitivity between P-repleted and P-depleted cultures, both measured without phosphorus in the media showed that the nutritional state of the algal cells can itself explain differences in the Cu sensitivity of the algae. In agreement with our observations, Hall et al. (1989) also found greater copper toxicity in P-limited cultures of *Chlorella vulgaris*. They concluded that P-limited cells responded much more dramatically to the increasing metal concentration than P-enriched cells. Metal toxicity in the P-deficient cells was found that P-deficient green algae responded much more dramatically to Cu-induced phosphate starvation and that the exogenous addition of phosphate could antagonize the Cu-effect in *Nostoc calcicola*. It has been reported that Cu induces the deficiency of P directly by the inhibition of phosphate uptake, indirectly by reducing the permeability of cell membranes (Nalewajko and Olaveson 1994). Other recent studies also supported this hypothesis; Wang and Dei (2006) found that P-deficient green algae responded much more dramatically to the increasing metal concentration than P-enriched cells. Metal toxicity in the P-deficient cells was observed within a very narrow range of ambient metal concentrations or cellular concentrations. One explanation

### Table 2

<table>
<thead>
<tr>
<th></th>
<th></th>
<th></th>
<th></th>
<th></th>
<th></th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>Min</td>
<td>21.3</td>
<td>1.97</td>
<td>14.8</td>
<td>14.9</td>
<td>19.9</td>
<td>26.2</td>
</tr>
<tr>
<td></td>
<td>(18.4, 24.2)</td>
<td>(−42.7, 46.7)</td>
<td>(−1.79, 31.3)</td>
<td>(−14.1, 43.9)</td>
<td>(7.68, 32.0)</td>
<td>(14.8, 37.6)</td>
</tr>
<tr>
<td>Max</td>
<td>100.7</td>
<td>110.5</td>
<td>117</td>
<td>108</td>
<td>99.7</td>
<td>106.8</td>
</tr>
<tr>
<td></td>
<td>(95.8, 105.6)</td>
<td>(73.6, 147)</td>
<td>(90.2, 144)</td>
<td>(94.8, 122)</td>
<td>(90.4, 109.0)</td>
<td>(95.8, 117.9)</td>
</tr>
<tr>
<td>EC&lt;sub&gt;50&lt;/sub&gt;</td>
<td>0.88</td>
<td>3.01</td>
<td>1.47</td>
<td>5.27</td>
<td>3.15</td>
<td>2.41</td>
</tr>
<tr>
<td></td>
<td>(0.79, 1.00)</td>
<td>(1.12, 8.10)</td>
<td>(0.75, 2.88)</td>
<td>(2.61, 10.7)</td>
<td>(2.11, 4.72)</td>
<td>(1.57, 3.68)</td>
</tr>
<tr>
<td>Hillslope</td>
<td>−34.3</td>
<td>−8.41</td>
<td>−11.5</td>
<td>−14.3</td>
<td>−18.2</td>
<td>−18.0</td>
</tr>
<tr>
<td></td>
<td>(−46.5, −22.0)</td>
<td>(−17.1, 0.27)</td>
<td>(−19.1, −3.84)</td>
<td>(−25.7, −3.01)</td>
<td>(−28.7, −7.60)</td>
<td>(−28.5, 7.40)</td>
</tr>
</tbody>
</table>

The effective Cu concentration (EC<sub>50</sub>) is expressed in µM

**Fig. 6** EC<sub>50</sub> values calculated as the reduction of the photochemical yield (Y) in percentage of the control obtained after 3 h of copper exposure in three different treatments. Bars represent the average of the EC<sub>50</sub> values obtained in duplicate in the two experiments. Black and dashed bars correspond to the results obtained in the experiment 1 and experiment 2 respectively.

Discussion

The use of short-term physiological tests with natural water as the incubation medium and intact communities as key organisms (i.e. periphyton), has the great advantage of being simple methodologically and at the same time ecologically realistic. On the other hand, these tests introduce a high degree of complexity as the chemical behaviour of toxicants is affected by local water chemistry (Guasch et al. 2003). Some of these complexities were systematically analysed using cultures.
was that the cells were more stressed under P-limited conditions, resulting in the greater toxicity of metals under such conditions.

In our culture experiments, when comparing results from the toxicity tests performed with and without P in the media, a reduction in Cu toxicity of 1.6 times was observed when P was added to the media (in both, EXP 1 and 2). Since the experiments were done with P-repleted cultures, it was not expected that the algae would suffer P-limitation during exposure, thus, the results support the alternative hypothesis that P–Cu interaction in the media leads to a reduction in Cu bioavailability. It has been shown that phosphate precipitates with other metals and that this leads to decreased metal availability (Schulze and Brand 1978; Nalewajko and Paul 1985). However, this argument is difficult to apply to natural systems since a copper phosphate precipitate is thermodynamically unlikely to occur in natural waters, because of the presence of competing ligands.

Concerning the experiments performed with natural periphyton, since they were performed with site water, it was expected that differences in Cu sensitivity would integrate the effect of differences in nutritional conditions during growth as well as differences in P content in the water. In fact, the increase in Cu tolerance of periphyton grown under more eutrophic conditions (T1) compared to periphyton from more oligotrophic conditions (B1) (with a 3.6-fold increase of EC50 values) was equivalent to the change in Cu toxicity found in the P-limited cultures exposed to Cu without P in the media and the P-repleted cultures exposed to Cu with P in the media (around 2-3 times). Since the increase of Cu tolerance observed in N. perminuta cultures may only be attributed to the influence of P, it indicates that differences in Cu tolerance observed in field communities from different sites might be related with in situ differences in P availability.

Higher metal tolerance found in both, algal cultures and communities, grown under more eutrophic conditions may be attributed to polyphosphate bodies (PPB) production in algal cells under non-limiting conditions. It has been well described that when the concentration of P in the medium is high, the levels of cellular P are elevated with P in excess of immediate cell requirements being stored as PPB (Rhee 1972, 1973). Several studies have reported the role of PPB in metal detoxification (e.g. Jensen et al. 1982; Twiss and Nalewajko 1992). These authors stated that intracellular polyphosphate is important in sequestering metals in a detoxified form and protecting the cell from the toxic effects of the metals. Thus, detoxification by PPB could also explain the high tolerance found in natural communities developed under more eutrophic conditions and P-repleted cultures in the present study, but this aspect was not directly addressed.

In the channel experiments, the exogenous addition of P during the toxicity tests enhanced Cu tolerance of the communities from the more oligotrophic sites (B1 and B2). These observations might be related to lowered nutrient limitation by the P-addition during the tests compensating the effects of Cu. These results are in agreement with a previous study (Guasch et al. 2004) where copper toxicity was slightly reduced when nutrients were simultaneously added in nutrient-limited communities. They suggested that the addition of metals produces a strong nutrient limitation, and that the addition of the toxicant together with phosphorus may partially compensate the inhibitory effect on the physiological response of the community. Barranguet et al. (2002) found similar results.

Chemical factors other than phosphorus are known to modify the response of natural periphyton in fluvial ecosystems. The pH of the media has been considered an important factor influencing the toxicity of Cu on periphyton (Guasch et al. 2002) and on algal cultures (Stadorub et al. 1987). These authors showed an increase in Cu toxicity when pH was reduced. This can be explained by copper speciation, because the amount of Cu2+ (which is considered the most available form of Cu for algae) increases at lower pH (Guasch et al. 2002). In our study, water pH in the experimental channels, which was done using natural periphyton and site water, ranged between 7.96 and 8.16 units of pH. These pH differences could influence the toxicity of Cu on periphyton. Taking into account these arguments, a higher sensitivity would be expected in B2, which had the lowest water pH during the channel experiments. This expectation contrasts with the results found: B1 and T1 showed marked differences in EC50’s in spite of having similar water pH. We can therefore conclude that pH has not played a decisive role in explaining differences in the Cu sensitivity of natural periphyton in our experiments.

Differences in copper sensitivity between different periphyton communities have also been attributed to the protective role of biomass (Admiraal et al. 1999; Navarro et al. 2002; Guasch et al. 2003). Barranguet et al. (2002) suggested that the dependence between periphyton biomass and Cu toxicity was related to its unspecific mode of action. Cu toxicity was progressive by indiscriminately damaging the algae at or protruding progressively from the biofilm surface to deeper biofilm layers, thus, the magnitude of the effect depended on the initial biomass in the short term (Guasch et al. 2004). In our investigation, although the periphyton communities presented differences in algal biomass, a long duration of the toxicity tests (18 h) was specifically selected to allow the metal to attain the deeper layers of the biofilm, minimizing the possible influence of algal biomass on the toxicological response of periphyton. Differences in the Cu sensitivity...
of natural communities could also be attributed to different algal taxa. However, taking into account that metal sensitivity is species-specific (Hutchinson and Stokes 1975), the taxonomic resolution obtained in our study (done at algal group level) did not allow us to derive this information.

Several ecological implications may be suggested if the experimental results obtained are extrapolated to the ecosystem scale. Based on the results of this study, it is suggested that periphyton communities in fluvial systems draining forested areas will be more sensitive to copper since these sites will have low phosphate concentrations. In addition, oligotrophic sites will also have a low probability of receiving metal inputs due to the dominating land use. On the other hand, periphyton communities located in more human-impacted areas will suffer from eutrophication and potential metal inputs, as has been reported in previous studies (Guasch et al. 2009), and are expected to be more tolerant to Cu.

Acknowledgments This study was supported by two Spanish Ministry projects (REN2003-07702 and CGL2006-12785) and the EC project KEYBIOEFFECTS (MRTN-CT-2006-035695).

Open Access This article is distributed under the terms of the Creative Commons Attribution Noncommercial License which permits any noncommercial use, distribution, and reproduction in any medium, provided the original author(s) and source are credited.

References


Hutchinson TC, Stokes PM (1975) Heavy metal toxicity and algal bioassays. ASTM special technical publication 573:320–343


